# TREATMENT OF DISEASES WITH COMBINATIONS OF ALPHA 7 NICOTINIC ACETYLCHOLINE RECEPTOR AGONISTS AND OTHER COMPOUNDS

## CROSS REFERENCE TO RELATED APPLICATIONS

This application claims the benefit of US provisional application Serial No. 60/432527 filed on 11 December 2002, under 35 USC 119(e)(i), which is incorporated herein by reference in its entirety.

## FIELD OF INVENTION

The present invention relates to compositions and methods to treat diseases or condition with a Nicotinic acetylcholine receptors (nAChRs) full agonist relative to nicotine plus either an inhibitor of cholinesterase, and/or a beta secretase inhibitor, and/or a gamma secretase inhibitor collectively referred to as "inhibitors."

## BACKGROUND OF THE INVENTION

The α7 nAChR is one receptor system that has proved to be a difficult target for testing. Native α7 nAChR is not routinely able to be stably expressed in most mammalian cell lines (Cooper and Millar, *J. Neurochem.*, 1997, 68(5):2140-51). Another feature that makes functional assays of α7 nAChR challenging is that the receptor is rapidly (100 milliseconds) inactivated. This rapid inactivation greatly limits the functional assays that can be used to measure channel activity.

Recently, Eisele et al. has indicated that a chimeric receptor formed between the N-terminal ligand binding domain of the α7 nAChR (Eisele et al., *Nature*, 366(6454), p 479-83, 1993), and the pore forming C-terminal domain of the 5-$HT_3$ receptor expressed well in *Xenopus* oocytes while retaining nicotinic agonist sensitivity. Eisele et al. used the N-terminus of the avian (chick) form of the α7 nAChR receptor and the C-terminus of the mouse form of the 5-$HT_3$ gene. However, under physiological conditions the α7 nAChR is a calcium channel while the 5-$HT_3$R is a sodium and potassium channel. Indeed, Eisele et al. teaches that the chicken α7 nAChR/ mouse 5-$HT_3$R behaves quite differently than the native α7 nAChR with the pore element not conducting calcium but actually being blocked by calcium ions. WO 00/73431 A2 reports on assay conditions under which the 5-$HT_3$R can be made to

conduct calcium. This assay may be used to screen for agonist activity at this receptor.

Alzheimer's disease (AD) is a progressive degenerative disease of the brain primarily associated with aging. Clinical presentation of AD is characterized by loss of memory, cognition, reasoning, judgment, and orientation. As the disease progresses, motor, sensory, and linguistic abilities are also affected until there is global impairment of multiple cognitive functions. These cognitive losses occur gradually, but typically lead to severe impairment and eventual death in the range of four to twelve years.

Alzheimer's disease is characterized by two major pathologic observations in the brain: neurofibrillary tangles and beta amyloid (or neuritic) plaques, comprised predominantly of an aggregate of a peptide fragment know as A beta. Individuals with AD exhibit characteristic beta-amyloid deposits in the brain (beta amyloid plaques) and in cerebral blood vessels (beta amyloid angiopathy) as well as neurofibrillary tangles. Neurofibrillary tangles occur not only in Alzheimer's disease but also in other dementia-inducing disorders. On autopsy, large numbers of these lesions are generally found in areas of the human brain important for memory and cognition.

Beta-amyloid is a defining feature of AD, now believed to be a causative precursor or factor in the development of disease. Deposition of A beta in areas of the brain responsible for cognitive activities is a major factor in the development of AD. Beta-amyloid plaques are predominantly composed of amyloid beta peptide (A beta, also sometimes designated betaA4). A beta peptide is derived by proteolysis of the amyloid precursor protein (APP) and is comprised of 39-42 amino acids. Several proteases called secretases are involved in the processing of APP.

Cleavage of APP at the N-terminus of the A beta peptide by beta-secretase and at the C-terminus by one or more gamma-secretases constitutes the beta-amyloidogenic pathway, i.e. the pathway by which A beta is formed. Cleavage of APP by alpha-secretase produces alpha-sAPP, a secreted form of APP that does not result in beta-amyloid plaque formation. This alternate pathway precludes the formation of A beta peptide. A description of the proteolytic processing fragments of APP is found, for example, in U.S. Patent Nos. 5,441,870; 5,721,130; and 5,942,400.

An aspartyl protease has been identified as the enzyme responsible for processing of APP at the beta-secretase cleavage site. The beta-secretase enzyme has been disclosed using varied nomenclature, including BACE, Asp, and Memapsin. See, for example, Sinha et.al., 1999, *Nature* 402:537-554 (p501) and published PCT application WO00/17369.

Several lines of evidence indicate that progressive cerebral deposition of beta-amyloid peptide (A beta) plays a seminal role in the pathogenesis of AD and can precede cognitive symptoms by years or decades. See, for example, Selkoe, 1991, *Neuron* 6:487.

It has been proposed that A beta peptide accumulates as a result of APP processing by beta secretase and or gamma secretase thus inhibition of either enzymes' activity may be desirable for the treatment of AD. *In vivo* processing of APP at the beta-secretase cleavage site is thought to be a rate-limiting step in A beta production, and it, thus, may be a good therapeutic target for the treatment of AD. See for example, Sabbagh, M., et al., 1997, *Alz. Dis. Rev.* 3, 1-19.

Cognitive disorders, including Alzheimer's disease, are generally accompanied by symptoms of forgetfulness, confusion, memory loss and other symptoms resulting from aging, brain injury, or disease. The concomitant decrease in cognitive function during the aging process has been documented in various mammals, including humans. In particular, presenile and senile primary degenerative dementia appear to be common causes of mental deterioration among the elderly. The symptoms of cognitive disorder appear to be associated with decreased acetylcholine synthesis as well as impairment of the ACh receptive neurons. The activity of the enzyme choline acetyltransferase (ChAT), which catalyzes the synthesis of acetylcholine from choline and acetyl coenzyme A, can be severely reduced as reflected by the loss of cholinergic (acetylcholine releasing) nerve endings in the hippocampus. Conversely, alpha 7 nAChRs are generally intact. The cholinergic neurotransmission are thus recognized as critically important to memory function.

Presently, there are three general approaches to enhance cholinergic transmission in the central nervous system. The first approach is to enhance cholinergic neurons by excessive exposure to a form of choline. Such attempts have been mildly successful, but only in the early stages of Alzheimer's disease.

The second approach involves postsynaptic direct stimulation of alpha 7 nAChRs. The third approach involves the inhibition of acetylcholinesterase, the enzyme that metabolizes acetylcholine. Accordingly, new compositions and methods for treating diseases resulting from cholinergic hypofunction are desired.

## SUMMARY OF THE INVENTION

The present invention is useful for the treatment of, or preparation of a medicament for the treatment of, a wide variety of disease and disorders where the alpha 7 nAChR receptor is implicated, including any one or more of the following: cognitive and attention deficit symptoms of Alzheimer's, neurodegeneration associated with diseases such as Alzheimer's disease, pre-senile dementia (mild cognitive impairment), senile dementia, amyotrophic lateral sclerosis, traumatic brain injury, behavioral and cognitive problems in general and associated with brain tumors, AIDS dementia complex, dementia associated with Down's syndrome, dementia associated with Lewy Bodies, Huntington's disease, Parkinson's disease, age-related macular degeneration.

Diseases to be treated within the scope of the present invention, including Alzheimer's disease, are chronic neurodegenerative disorders. Acetylcholine-synthesizing neurons in the basal forebrain region and their cortical synaptic connections exhibit a well-characterized degeneration in Alzheimer's disease. The symptoms of this degeneration and can be treated with the drug combinations described herein.

Embodiments of the invention may include one or more or combination of the following. The present invention claims the method of treating the diseases discussed herein or preparing a medicament to so treat, using any compound that is a full agonist to an α7 Nicotinic Acetylcholine Receptor (nAChR) or α7 nAChR full agonists, described either herein or elsewhere to be administered with either: I) a cholinesterase inhibitor, II) a beta secretase inhibitor, III) a gamma secretase inhibitor or any combination of one, two, or three of the different inhibitors in combination with a α7 nAChR full agonists. The use of the term α7 nAChR full agonist is used interchangeably with α7 nAChR agonists when discussing the compounds of the present invention. Another aspect of the present invention includes α7 nAChR full agonists as described for example, but not by way of limitation, in any one or more of

the following patents and published applications: WO 01/60821A1, WO 01/36417A1, WO 02/100857A1, WO 03/042210A1, and WO 03/029252A1, all of which are incorporated herein by reference. In particular, by way of example and not limitation, some α7 nAChR full agonist are the compounds of Formula I as described herein.

The method or use of a compound of Formula I, where X is O, or X is S, and where the other variables of Formula I have any definition discussed herein.

The method or use of a compound of Formula I, where Azabicyclo is any one or more of I, II, III, IV, V, VI, or VII.

The method or use of a compound of Formula I, where W is any one or more of (A), (B), (C), (D), (E), (F), (G), or (H).

The present invention also includes pharmaceutical compositions containing the active compounds, and methods to treat the identified diseases.

The present invention is useful for the treatment of, or preparation of a medicament for the treatment of, a wide variety of disease and disorders where the alpha 7 nAChR is implicated, including any one or more of the following: cognitive and attention deficit symptoms of Alzheimer's, neurodegeneration associated with diseases such as Alzheimer's disease, pre-senile dementia (mild cognitive impairment), senile dementia, amyotrophic lateral sclerosis, traumatic brain injury, behavioral and cognitive problems in general and associated with brain tumors, AIDS dementia complex, dementia associated with Down's syndrome, dementia associated with Lewy Bodies, Huntington's disease, Parkinson's disease, age-related macular degeneration.

Another aspect of the present invention includes the method or use of a compound of Formula I, where Azabicyclo is any one or more of I, II, III, IV, V, VI, or VII. The method or use of a compound of Formula I, where $R_1$ is H, alkyl, cycloalkyl, haloalkyl, substituted phenyl, or substituted naphthyl; each $R_2$ is independently F, Cl, Br, I, alkyl, substituted alkyl, haloalkyl, cycloalkyl, aryl, or $R_2$ is absent; $R_{2\text{-}3}$ is H, F, Cl, Br, I, alkyl, haloalkyl, substituted alkyl, cycloalkyl, or aryl; each $R_3$ is H; and $R_4$ is H. The method or use of a compound of Formula I, where the variables of formula I have any definition discussed herein.

Another aspect of the present invention includes the method or use of a compound of Formula I, where W is any one or more of (A), (B), (C), (D), (E), (F),

(G), or (H). The method or use of a compound of Formula I, where W is any one or more of (A), (B), (C), (D), (E), (F), (G), or (H). The method or use of a compound of Formula I, where W is any one or more of (A), (B), (C), (D), (E), (F), (G), or (H), wherein the variables within each has any definition allowed. For example, and not by way of limitation, W includes any one or more of the following: 4-chlorobenz-1-yl; dibenzo[b,d]thiophene-2-yl; isoquinoline-3-yl; furo[2,3-c]pyridine-5-yl; 1,3-benzodioxole-5-yl; 2,3-dihydro-1,4-benzodioxine-6-yl; 1,3-benzoxazole-5-yl; thieno[2,3-c]pyridine-5-yl; thieno[3,2-c]pyridine-6-yl; [1]benzothieno[3,2-c]pyridine-3-yl; 1,3-benzothiazole-6-yl; thieno[3,4-c]pyridine-6-yl; 2,3-dihydro-1-benzofuran-5-yl; 1-benzofuran-5-yl; furo[3,2-c]pyridine-6-yl; [1]benzothieno[2,3-c]pyridine-3-yl; dibenzo[b,d]furan-2-yl; 1-benzofuran-6-yl; 2-naphthyl; 1H-indole-6-yl; pyrrolo[1,2-c]pyrimidine-3-yl; 1-benzothiophene-5-yl; 1-benzothiophene-5-yl; 1-benzothiophene-6-yl; pyrrolo[1,2-a]pyrazine-3-yl; 1H-indole-6-yl; pyrazino[1,2-a]indole-3-yl; 1,3-benzothiazole-6-yl; [1]benzofuro[2,3-c]pyridine-3-yl; [1]benzofuro[2,3-c]pyridine-3-yl; 2H-chromene-6-yl; indolizine-6-yl; and [1,3]dioxolo[4,5-c]pyridine-6-yl; any of which is optionally substituted as allowed in formula I. One of ordinary skill in the art will recognize how the variables are defined by comparing the named radicals with the different values for W. When W is (D), it is preferred that one of $R_{D-1}$ is the bond to C(X). Specific compounds within the scope of this invention include any one or more of the following as the free base or as a pharmaceutically acceptable salt thereof:

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-4-chlorobenzamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]dibenzo[b,d]thiophene-2-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]isoquinoline-3-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1,3-benzodioxole-5-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2-methylfuro[2,3-c]pyridine-5-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2,3-dihydro-1,4-benzodioxine-6-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-methylfuro[2,3-c]pyridine-5-carboxamide;

N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]isoquinoline-3-carboxamide;

N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-methylfuro[2,3-c]pyridine-5-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1,3-benzoxazole-5-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2-methyl-1,3-benzoxazole-5-carboxamide;

N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]thieno[2,3-c]pyridine-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]thieno[3,2-c]pyridine-6-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-ethylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-isopropylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]thieno[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]thieno[3,2-c]pyridine-6-carboxamide;
5-{[(2R)-7-azoniabicyclo[2.2.1]hept-2-ylamino]carbonyl}-3-ethylfuro[2,3-c]pyridin-6-ium dichloride;
5-{[(2R)-7-azoniabicyclo[2.2.1]hept-2-ylamino]carbonyl}-3-isopropylfuro[2,3-c]pyridin-6-ium dichloride;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-1-azabicyclo[2.2.2]oct-3-yl[1]benzothieno[3,2-c]pyridine-3-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1,3-benzothiazole-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-chlorofuro[2,3-c]pyridine-5-carboxamide;
N-1-azabicyclo[2.2.2]oct-3-ylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]thieno[3,4-c]pyridine-6-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]-3-methylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-3-methylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2,3-dihydro-1-benzofuran-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]thieno[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]furo[3,2-c]pyridine-6-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]thieno[3,2-c]pyridine-6-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]3-ethylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]3-isopropylfuro[2,3-c]pyridine-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-chlorofuro[2,3-c]pyridine-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]3-chlorofuro[2,3-c]pyridine-5-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]-4-chlorobenzamide;

N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]thieno[3,4-c]pyridine-6-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]dibenzo[b,d]thiophene-2-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl][1]benzothieno[2,3-c]pyridine-3-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl][1]benzothieno[2,3-c]pyridine-3-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-1-benzofuran-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]dibenzo[b,d]furan-2-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-bromofuro[2,3-c]pyridine-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-bromofuro[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1-benzofuran-6-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]-2-naphthamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]pyrrolo[1,2-c]pyrimidine-3-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]thieno[2,3-c]pyridine-5-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]thieno[3,2-c]pyridine-6-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-1H-indole-6-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]thieno[2,3-c]pyridine-5-carboxamide;
3-methyl-N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]-1-benzofuran-5-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]thieno[3,2-c]pyridine-6-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]pyrrolo[1,2-c]pyrimidine-3-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]-1,3-benzothiazole-6-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]pyrrolo[1,2-c]pyrimidine-3-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1-benzothiophene-5-carboxamide;

N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]pyrrolo[1,2-c]pyrimidine-3-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]pyrrolo[1,2-c]pyrimidine-3-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-3-bromofuro[2,3-c]pyridine-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-1,3-benzodioxole-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-bromo-1-benzofuran-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-bromo-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-bromothieno[2,3-c]pyridine-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-bromothieno[2,3-c]pyridine-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-1-benzothiophene-5-carboxamide;
N-[(3S)-1-azabicyclo[2.2.2]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-methyl-1-benzofuran-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-methyl-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2-methyl-1-benzofuran-6-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]-1-benzofuran-6-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]-1-benzofuran-6-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]-1-benzothiophene-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1-benzothiophene-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]pyrrolo[1,2-a]pyrazine-3-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-1-benzothiophene-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1-methyl-1H-indole-6-carboxamide;
N-[(3S)-1-azabicyclo[2.2.2]oct-3-yl]-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-isopropyl-1-benzofuran-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-isopropyl-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-ethynylfuro[2,3-c]pyridine-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1H-indazole-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2-methyl-1-benzofuran-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-2-methyl-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]pyrazino[1,2-a]indole-3-carboxamide;
3-bromo-N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]furo[2,3-c]pyridine-5-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]pyrrolo[1,2-a]pyrazine-3-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-7-methoxy-2-naphthamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]pyrrolo[1,2-a]pyrazine-3-carboxamide;
N-[(3R,5R)-1-azabicyclo[3.2.1]oct-3-yl]-1,3-benzothiazole-6-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl]-3-bromo-1-benzofuran-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl][1]benzofuro[2,3-c]pyridine-3-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl][1]benzofuro[2,3-c]pyridine-3-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-ethynyl-1-benzofuran-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-ethynyl-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2H-chromene-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-prop-1-ynyl-1-benzofuran-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2-phenyl-1,3-benzodioxole-5-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-6-bromopyrrolo[1,2-a]pyrazine-3-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-prop-1-ynylfuro[2,3-c]pyridine-5-carboxamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]pyrrolo[1,2-a]pyrazine-3-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]indolizine-6-carboxamide;
2-amino-N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-1,3-benzothiazole-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-6-ethynylpyrrolo[1,2-a]pyrazine-3-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-8-methoxy-2-naphthamide;
N-[(2S,3R)-2-methyl-1-azabicyclo[2.2.2]oct-3-yl]indolizine-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl][1,3]dioxolo[4,5-c]pyridine-6-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl][1,3]dioxolo[4,5-c]pyridine-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-cyano-1-benzofuran-5-carboxamide;
N-[(3R,4S)-1-azabicyclo[2.2.1]hept-3-yl][1,3]dioxolo[4,5-c]pyridine-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-ethyl-2,3-dihydro-1,4-benzodioxine-6-carboxamide;
N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-7-hydroxy-2-naphthamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-3-ethynylfuro[2,3-c]pyridine-5-carboxamide;
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-6-chloroisoquinoline-3-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-ethyl-2,3-dihydro-1,4-benzodioxine-6-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-ethyl-2,3-dihydro-1,4-benzodioxine-6-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-6-methylisoquinoline-3-carboxamide;

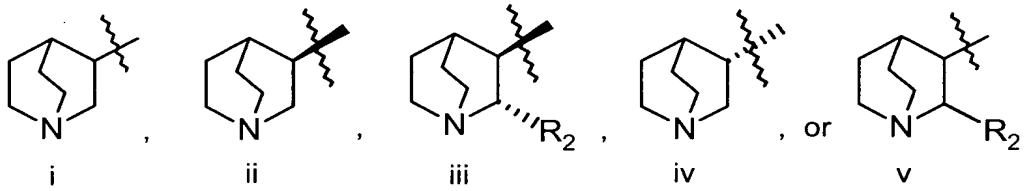
N-[(1S,2R,4R)-7-azabicyclo[2.2.1]hept-2-yl]-6-methylisoquinoline-3-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-3-cyanofuro[2,3-c]pyridine-5-carboxamide;

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-2-naphthamide; and

N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]dibenzo[b,d]furan-2-carboxamide.

The compounds of Formula I (Azabicyclo is I) have asymmetric centers on the quinuclidine ring. The compounds of the present invention include quinuclidines having 3*R* configuration, 2*S*, 3*R* configuration, or 3*S* configuration and also include racemic mixtures and compositions of varying degrees of streochemical purities. For example, and not by limitation, embodiments of the present invention include compounds of Formula I having the following stereospecificity and substitution:




i , ii , iii , iv , or v

wherein the Azabicyclo (i) is a racemic mixture;

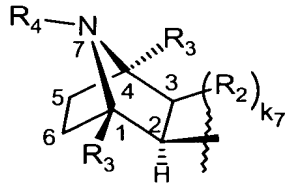
(ii) has the stereochemistry of 3*R* at C3;

(iii) has the 3*R*,2*S* stereochemistry at C3 and C2, respectively;

(iv) has the stereochemistry of 3*S* at C3; or

(v) is a racemic mixture; and for (iii) and (v), $R_2$ has any definition or specific value discussed herein.

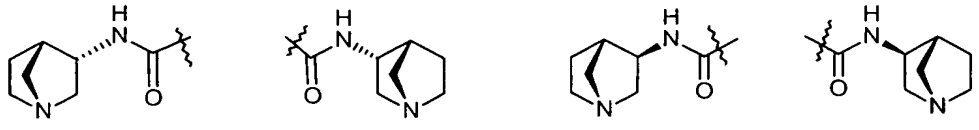
The compounds of Formula I (Azabicyclo is III) have asymmetric centers on the 7-azabicyclo[2.2.1]heptane ring which can exhibit a number of stereochemical configurations.



The terms *exo* and *endo* are stereochemical prefixes that describe the relative configuration of a substituent on a bridge (not a bridgehead) of a bicyclic system. If a substituent is oriented toward the larger of the other bridges, it is *endo*. If a substituent is oriented toward the smaller bridge it is *exo*. Depending on the substitution on the carbon atoms, the *endo* and *exo* orientations can give rise to different stereoisomers. For instance, when carbons 1 and 4 are substituted with hydrogen and carbon 2 is bonded to a nitrogen-containing species, the *endo* orientation gives rise to the possibility of a pair of enantiomers: either the 1*S*, 2*S*, 4*R* isomer or its enantiomer, the 1*R*, 2*R*, 4*S* isomer. Likewise, the *exo* orientation gives rise to the possibility of another pair of stereoisomers which are diastereomeric and C-2 epimeric with respect to the *endo* isomers: either the 1*R*, 2*S*, 4*S* isomer or its enantiomer, the 1*S*, 2*R*, 4*R* isomer. The compounds of this invention exist in the *exo* orientation. For example, when $R_2$ is absent (C3 is $-CH_2-$) and $R_3$ = H, the absolute stereochemistry is *exo*-(1*S*, 2*R*, 4*R*).

The compounds of the present invention have the *exo* orientation at the C-2 carbon and *S* configuration at the C-1 carbon and the *R* configuration at the C-2 and the C-4 carbons of the 7-azabicyclo[2.2.1]heptane ring. Unexpectedly, the inventive compounds exhibit much higher activity relative to compounds lacking the *exo* 2*R*, stereochemistry. For example, the ratio of activities for compounds having the *exo* 2*R* configuration to other stereochemical configurations may be greater than about 100:1. Although it is desirable that the stereochemical purity be as high as possible, absolute purity is not required. For example, pharmaceutical compositions can include one or more compounds, each having an *exo* 2*R* configuration, or mixtures of compounds having *exo* 2*R* and other configurations. In mixtures of compounds, those species possessing stereochemical configurations other than *exo* 2*R* act as diluents and tend to lower the activity of the pharmaceutical composition. Typically, pharmaceutical compositions including mixtures of compounds possess a larger percentage of species having the *exo* 2*R* configuration relative to other configurations.

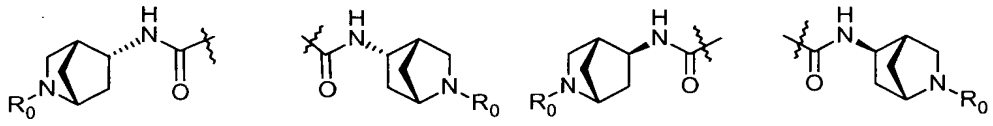
The compounds of Formula I (Azabicyclo is II) have asymmetric center(s) on the [2.2.1] azabicyclic ring at C3 and C4. The scope of this invention includes the separate stereoisomers of Formula I being *endo*-4*S*, *endo*-4*R*, *exo*-4*S*, *exo*-4*R*:



*endo*-4*S* *endo*-4*R* *exo*-4*S* *exo*-4*R*

The *endo* isomer is the isomer where the non-hydrogen substituent at C3 of the [2.2.1] azabicyclic compound is projected toward the larger of the two remaining bridges. The *exo* isomer is the isomer where the non-hydrogen substituent at C3 of the [2.2.1] azabicyclic compound is projected toward the smaller of the two remaining bridges. Thus, there can be four separate isomers: *exo*-4(*R*), *exo*-4(*S*), *endo*-4(*R*), and *endo*-4(*S*). Some embodiments of compounds of Formula I for when Azabicyclo is II include racemic mixtures where $R_2$ is absent ($k_2$ is 0) or is at C2 or C6; or Azabicyclo II has the *exo*-4(*S*) stereochemistry and $R_2$ has any definition discussed herein and is bonded at any carbon discussed herein.

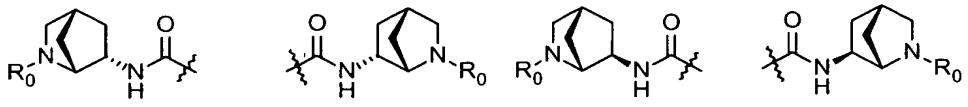
The compounds of Formula I (Azabicyclo III) have asymmetric center(s) on the [2.2.1] azabicyclic ring at C1, C4 and C5. The scope of this invention includes racemic mixtures and the separate stereoisomers of Formula I being (1*R*,4*R*,5*S*), (1*R*,4*R*,5*R*), (1*S*,4*S*,5*R*), (1*S*,4*S*,5*S*):



*endo*-1*R*,4*R*,5*R* *endo*-1*S*,4*S*,5*S* *exo*-1*R*,4*R*,5*S* *exo*-1*S*,4*S*,5*R*

The *endo* isomer is the isomer where the non-hydrogen substituent at C5 of the [2.2.1] azabicyclic compound is projected toward the larger of the two remaining bridges. The *exo* isomer is the isomer where the non-hydrogen substituent at C5 of the [2.2.1] azabicyclic compound is projected toward the smaller of the two remaining bridges. Thus, there can be four separate isomers: *exo*-(1*R*,4*R*,5*S*), *exo*-(1*S*,4*S*,5*R*), *endo*-(1*S*,4*S*,5*S*), *endo*-(1*R*,4*R*,5*R*). Another group of compounds of Formula I (Azabicyclo III) includes $R_{2\text{-}3}$ is absent, or is present and either at C3 or bonds to any carbon with sufficient valancy.

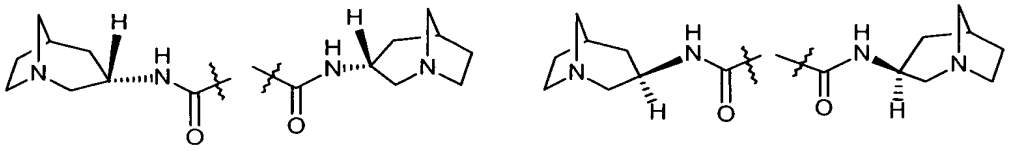
The compounds of Formula I (Azabicyclo IV) have asymmetric center(s) on the [2.2.1] azabicyclic ring at C1, C4 and C6. The scope of this invention includes racemic mixtures and the separate stereoisomers of Formula I being *exo*-(1*S*,4*R*,6*S*), *exo*-(1*R*,4*S*,6*R*), *endo*-(1*S*,4*R*,6*R*), and *endo*-(1*R*,4*S*,6*S*):



*endo*-1*R*,4*S*,6*S* *endo*-1*S*,4*R*,6*R* *exo*-1*R*,4*S*,6*R* *exo*-1*S*,4*R*,6*S*

The *endo* isomer is the isomer where the non-hydrogen substituent at C6 of the [2.2.1] azabicyclic compound is projected toward the larger of the two remaining bridges. The *exo* isomer is the isomer where the non-hydrogen substituent at C6 of the [2.2.1] azabicyclic compound is projected toward the smaller of the two remaining bridges. Thus, there can be four separate isomers: *exo*-(1*S*,4*R*,6*S*), *exo*-(1*R*,4*S*,6*R*), *endo*-(1*S*,4*R*,6*R*), and *endo*-(1*R*,4*S*,6*S*). Another group of compounds of Formula I (Azabicyclco IV) includes $R_{2\text{-}3}$ is H, or is other than H and bonded at C3 or is bonded to any carbon with sufficient valancy.

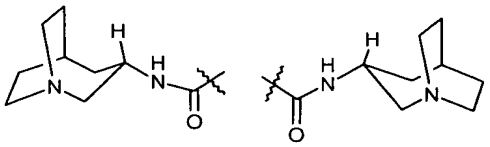
The compounds of Formula I have asymmetric center(s) on the [3.2.1] azabicyclic ring at C3 and C5. The scope of this invention includes the separate stereoisomers of Formula I being *endo*-3*S*, 5*R*, *endo*-3*R*, 5*S*, *exo*-3*R*, 5*R*, *exo*-3*S*, 5*S*:



*endo*-3*S*, 5*R* *endo*-3*R*, 5*S* *exo*-3*R*, 5*R* *exo*-3*S*, 5*S*

Another group of compounds of Formula I (Azabicyclo V) includes compounds where Azabicyclo V moiety has the stereochemistry of 3*R*, 5*R*, or is a racemic mixture and the moiety is either not substituted with $R_2$ (each is absent) or has one to two substituents being at either C2 and/or C4. When the moiety is substituted, the preferred substituents for substitution at C2 are alkyl, haloalkyl, substituted alkyl, cycloalkyl, or aryl; and for substitution at C4 are F, Cl, Br, I, alkyl, haloalkyl, substituted alkyl, cycloalkyl, or aryl.

The compounds of Formula I (Azabicyclo is VI) have asymmetric centers on the [3.2.2] azabicyclic ring with one center being at C3 when $R_2$ is absent. The scope of this invention includes racemic mixtures and the separate stereoisomers of Formula I being 3(*S*) and 3(*R*):



3(*S*) 3(*R*)

Another group of compounds of Formula I (Azabicyclo VI) includes compounds where Azabicyclo VI moiety is either not substituted with $R_2$ (each is absent) or has

one to two substituents with one being at either C2 or C4 or when two are present, one being at each C2 and C4. When the moiety is substituted, the preferred substituents for substitution at C2 are alkyl, haloalkyl, substituted alkyl, cycloalkyl, or aryl; and for substitution at C4 are F, Cl, Br, I, alkyl, haloalkyl, substituted alkyl, cycloalkyl, or aryl.

Stereoselective syntheses and/or subjecting the reaction product to appropriate purification steps produce substantially enantiomerically pure materials. Suitable stereoselective synthetic procedures for producing enantiomerically pure materials are well known in the art, as are procedures for purifying racemic mixtures into enantiomerically pure fractions.

The compounds of the present invention having the specified stereochemistry above have different levels of activity and that for a given set of values for the variable substitutuents one isomer may be preferred over the other isomers. Although it is desirable that the stereochemical purity be as high as possible, absolute purity is not required. It is preferred to carry out stereoselective syntheses and/or to subject the reaction product to appropriate purification steps so as to produce substantially enantiomerically pure materials. Suitable stereoselective synthetic procedures for producing enantiomerically pure materials are well known in the art, as are procedures for purifying racemic mixtures into enantiomerically pure fractions.

In another aspect, the present invention comprises a method of administering to a mammal an amount of at least one acetylcholinesterase inhibitor, beta secretase inhibitor, or gamma secretase inhibitor, collectively referred to as "an inhibitor," and an alpha 7 nAChR full agonist.

**Acetylcholinesterase Inhibitors**

When the inhibitor is an acetylcholinesterase inhibitor, the method would be used to treat diseases or conditions in a mammal, wherein the mammal experiences cholinergic hypofunction. As used herein, central and peripheral nervous system disorders involving cholinergic hypofunction include, but are not limited to, dementias, amnesias, cerebral insufficiencies, and psychiatric disturbances in the central nervous system and neuronal and smooth muscle dysfunction of the gut, skeletal muscle dysfunction for breathing, bladder, and secretory glands in the peripheral nervous system. The acetylcholinesterase inhibitor and alpha 7 nAChR full

agonist(s) can be administered together as a composition, or may be administered separately. They may be administered at the same or different times, but at some time point in the treatment both drugs should be in the patient's bloodstream at the same time.

The method would be used to treat diseases or conditions in a mammal, wherein the mammal experiences neurodegeneration leading to cholinergic hypofunction and concomitant central nervous system dysfunction. The central nervous system disorders involving cholinergic hypofunction include, but are not limited to, dementias, amnesias. The acetylcholinesterase inhibitor and alpha 7 nAChR full agonist(s) can be administered together as a composition, or may be administered separately.

The compositions of the invention can be administered using art-recognized techniques. Preferably, the inhibitor and the alpha 7 nAChR full agonist are administered orally, or parenterally. In general, however, the compositions of the invention can be administered using the same art-recognized techniques used for administration of acetylcholinesterase inhibitors and alpha 7 nAChR full agonists. Accordingly, techniques of administration need not be repeated here.

Acetylcholinesterase inhibitors including physostigmine, aricept, rivastigamine, galantamine, monoamine acridines and their derivatives (e.g., US Patent 4, 816,456), piperidinyl-alkanoyl heterocyclic compounds (e.g., EP 487 071), N-benzyl-piperidine derivatives (e.g., US Patent 5,106,856), 4-(1-benzylpiperidyl)-substituted fused quinoline derivatives (e.g., EP 481 429), cyclic amide derivatives (e.g., EP 468 187), and other typical acetylcholinesterase inhibitors such as carbonic acid derivatives (e.g., US Patent 5,602,176).

**Beta Secretase Inhibitors**

Various pharmaceutical agents have been proposed for the treatment of Alzheimer's disease but without any real success. Here we have determined that two classes of compounds may be especially effective in the treatment of Alzheimer's disease when combined with an alpha 7 nAChR full agonist. These are selective beta secretase inhibitors and selective gamma secretase inhibitors. Beta secretase inhibitors are more preferred and are described here in detail. By beta secretase inhibitors what is meant are compounds that are effective inhibitors of beta-secretase, that inhibit beta-secretase-mediated cleavage of APP, that are effective inhibitors of A

beta production, and/or are effective to reduce amyloid beta deposits or plaques. All beta-secretase mediated treatments suggested for the treatment and prevention of disease characterized by amyloid beta deposits or plaques, such as AD are included in the term beta-secretase inhibitors as used herein.

Illustrations of and non limiting examples of beta-secretase inhibitors are disclosed in the following references and by specific mention here are meant to be made part of this application, as if copied herein in whole, and intended to be incorporated herein by reference. These references and examples below are not intended to limit in any way the definition of a beta-secretase inhibitor discovered either before or after the filing of this application for patent.

Beta secretase inhibitors include the compounds disclosed in the following published patent applications and granted patents (incorporated herein by reference):

1. US 5,981,168, issued 9 November 1999, inventors P. B. Reiner and B.P. Connop. The compounds described in this publication, in particular the compounds disclosed on col. 4-8 and col.15-22. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

2. US 2002/0143177 A1, publication date 3 October 2002, inventors J. P. Beck, et al. The compounds described in this publication, in particular the compounds disclosed on pages 3–40, 46-63, and 68-107. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

3. US 2002/0128255 A1, published 12 September 2002, inventors J. P. Beck, et al. The compounds described in this publication, in particular the compounds disclosed on pages 3-38, and 43-265. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

4. US 2002/0115616 A1, published 22 August 2002, inventors J. G. Boyd and D.H Singleton. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

5. US 2002/0019403 A1, published 14 February 2002, inventors R. Hom, et al. The compounds described in this publication, in particular the compounds disclosed on pages 1-44, and 48-128. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

6. WO 00/77030 A1, published 21 December 2000, inventors J. Varghese, et al. The compounds described in this publication, in particular the compounds

disclosed on pages 4-6, 9-37 and 45-61. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

7. WO 01/70672 A2, published 27 September 2001, inventors R. Hom, et al. The compounds described in this publication, in particular the compounds described on pages 4-119, 137-178, and 189-234. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

8. WO 01/34639 A2, published 17 May 2001, inventors J. E. Audia, et al. The compounds described in this publication, in particular the compounds disclosed on pages 4-5, 7-35 and 52-56. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

9. WO 01/34571 A1, published 17 May 2001, inventors J. E. Audia, et al. The compounds described in this publication, in particular the compounds disclosed on pages 4-6, 8-43 and 60-66. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

10. WO 02/100856 A1, published 19 December 2002, inventors SR Pulley, et al. The compounds described in this publication, in particular the compounds disclosed on pages 4-25, 34-53, 79-108, 118-160. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

11. WO 02/100820 A1, published 19 December 2002, inventors M. Maillard and JA Tucker. The compounds described in this publication, in particular the compounds disclosed on pages 4-99, 122-199. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

12. WO 02/100818 A2, published 19 December 2002, inventors H.J. Schostarez and R.A. Chrusciel. The compounds described in this publication, in particular the compounds disclosed on pages 4-36, 46-52, 77-155, 164-205. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

13. WO 02/100399 A1, published 19 December 2002, inventors S. R. Pulley. The compounds described in this publication, in particular the compounds disclosed on pages 4-25, 35-53, 78-169. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

14. WO 02/98849 A2, published 12 December 2002, inventors J. Freskos, et al. The compounds described in this publication, in particular the compounds disclosed on pages 5-142, 164-182, 201-353. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

15. WO 02/94985 A2, published 28 November 2002, inventors J. E. Bruce, et al. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

16. WO 02/94768 A2, published 28 November 2002, inventors H. Schostarez, et al. The compounds described in this publication, in particular the compounds disclosed on pages 4-36, 44-107, 124-206, 223-287. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

17. WO 02/88101 A2, published 7 November 2002, inventors G. R. Bhisetti, et al. The compounds described in this publication, in particular the compounds disclosed on pages 4-7, 18-21, 32-88 and 98-200. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

18. WO 02/48150 A2, published 20 June 2002, inventors N. H. Greig, et al. The compounds described in this publication, in particular the compounds disclosed on pages 6-37, 48-50, 65-70 and 89-128. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

19. WO 02/02520 A2, published 10 January 2002, inventors J. P. Beck, et al. The compounds described in this publication, in particular the compounds disclosed on pages 8-98, 115-118, 122-158, and pages 166-284. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

20. WO 02/02518 A2, published 10 January 2002, inventors J. P. Beck, et al. The compounds described in this publication, in particular the compounds disclosed on page 8-99, 115-118, 122-158, and 166-284. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

21. WO 02/02512 A2, published 10 January 2002, inventors M. Maillaird, et al. The compounds described in this publication, in particular the compounds

disclosed on pages 8-96, 111-339, and 347-649. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

22. WO 02/02506 A2, published 10 January 2002, inventors L. Y. Fang and J. Varhese. The compounds described in this publication, in particular the compounds disclosed on pages 7-84, 100-103, 106-113, and 122-433. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

23. WO 02/02505 A2, published 10 January 2002, inventors L. Y. Fang, et al. The compounds described in this publication, in particular the compounds disclosed on pages 5-28, 29-61, 77-80, 83-92, and 100-135. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

24. WO 03/6453 A1, published 23 January 2003, inventors H.J. Schostarez and R.A. Chrusciel. The compounds described in this publication, in particular the compounds disclosed on pages 4-39, 47-55, 82-179. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

25. WO 03/6021 A1, published 23 January 2003, inventors H.J. Schostarez and R.A. Chrusciel. The compounds described in this publication, in particular the compounds disclosed on pages 4-38, 74-92, 102-130. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

26. WO 03/6013 A1, published 23 January 2003, inventors H.J. Schostarez and R.A. Chrusciel. The compounds described in this publication, in particular the compounds disclosed on pages 4-30, 38-45, 70-134, 143-170. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

27. WO 03/2122 A1, published 09 January 2003, inventorsJ. Varghese, et al. The compounds described in this publication, in particular the compounds disclosed on pages 4-24, 59-87. All compounds described or claimed in this publication are copied into this document and incorporated by reference herein.

**Gamma Secretase Inhibitors**

By gamma secretase inhibitors what is meant are compounds that are effective inhibitors of gamma-secretase, that inhibit gamma-secretase-mediated cleavage of APP, that are effective inhibitors of A beta production, and/or are effective to reduce amyloid beta deposits or plaques. All gamma-secretase mediated treatments suggested for the treatment and prevention of disease characterized by amyloid beta deposits or plaques, such as AD are included in the term gamma-secretase inhibitors as used herein.

In another aspect, the invention provides pharmaceutical compositions comprising a composition according to the invention and a pharmaceutically acceptable carrier or diluent and optionally other adjuvants. Acceptable carriers, diluents, and adjuvants are any of those commercially used in the art, in particular, those used in pharmaceutical compositions of acetyleholinesterase inhibitors and alpha 7 nAChR full agonists. Accordingly, such carriers, diluents, and adjuvants need not be repeated here.

In a combination therapy to treat the diseases or conditions discussed herein, the alpha 7 agonist and the inhibitor(s) can be administered simultaneously or at separate intervals. When administered simultaneously the alpha 7 agonist and the inhibitor(s) can be incorporated into a single pharmaceutical composition, e.g., a pharmaceutical combination therapy composition. Alternatively, two or more separate compositions, i.e., one containing alpha 7 agonist and the other(s) containing the inhibitor(s), can be administered simultaneously.

A pharmaceutical combination therapy composition can include therapeutically effective amounts of the alpha 7 agonist, noted herein, and therapeutically effective amount of the inhibitor(s). The combined administration of the alpha 7 agonist and the inhibitor(s) is expected to require less of the generally-prescribed dose for any of agents when used alone and or is expected to result in less frequent administration of either, both or all agents. These compositions may be formulated with common excipients, diluents or carriers, and compressed into tablets, or formulated elixirs or solutions for convenient oral administration or administered by intramuscular intravenous routes. The compounds can be administered rectally, topically, orally, sublingually, or parenterally and maybe formulated as sustained relief dosage forms and the like.

When separately administered, therapeutically effective amounts of compositions containing alpha 7 agonist and the inhibitor(s) are administered on a different schedule. One may be administered before the other as long as the time between the administrations falls within a therapeutically effective interval. A therapeutically effective interval is a period of time beginning when one of either (a) the alpha 7 agonist, or (b) one to three of the inhibitor(s) is(are) administered to a mammal and ending at the limit of the beneficial effect in the treatment of the disease or condition to be treated from the combination of (a) and (b). The methods of administration of the alpha 7 agonist and the inhibitor(s) may vary. Thus, any of the agents may be administered rectally, topically, orally, sublingually, or parenterally.

Further aspects and embodiments of the invention may become apparent to those skilled in the art from a review of the following detailed description, taken in conjunction with the examples and the appended claims. While the invention is susceptible of embodiments in various forms, described hereafter are specific embodiments of the invention with the understanding that the present disclosure is intended as illustrative, and is not intended to limit the invention to the specific embodiments described herein.

## DETAILED DESCRIPTION OF THE INVENTION

Surprisingly, we have found that α7 nAChR full agonists combined with either acetylcholinesterase inhibitors, beta secretase inhibitors and/or gamma secretase inhibitors can be used to treat any one or more of the following: cognitive and attention deficit symptoms of Alzheimer's, neurodegeneration associated with diseases such as Alzheimer's disease, pre-senile dementia (mild cognitive impairment), senile dementia, amyotrophic lateral sclerosis, traumatic brain injury, behavioral and cognitive problems in general and associated with brain tumors, AIDS dementia complex, dementia associated with Down's syndrome, dementia associated with Lewy Bodies, Huntington's disease, Parkinson's disease, age-related macular degeneration. Alpha 7 nAChR full agonists within the scope of the present invention include compounds of Formula I.

In another aspect, the present invention comprises a method of administering the alpha 7 agonist to a mammal with an effective amount of at least one of the following acetylcholinesterase inhibitor, beta secretase inhibitor, or gamma secretase

inhibitor, collectively referred to as "an inhibitor," and an alpha 7 nAChR full agonist. What is meant by acetylcholinesterase inhibitors, beta secretase inhibitors, and gamma secretase inhibitors is discussed herein.

In another aspect, the invention provides pharmaceutical compositions comprising a composition according to the invention and a pharmaceutically acceptable carrier or diluent and optionally other adjuvants. Acceptable carriers, diluents, and adjuvants are any of those commercially used in the art, in particular, those used in pharmaceutical compositions of acetyleholinesterase inhibitors and alpha 7 nAChR full agonists. Accordingly, such carriers, diluents, and adjuvants need not be repeated here.

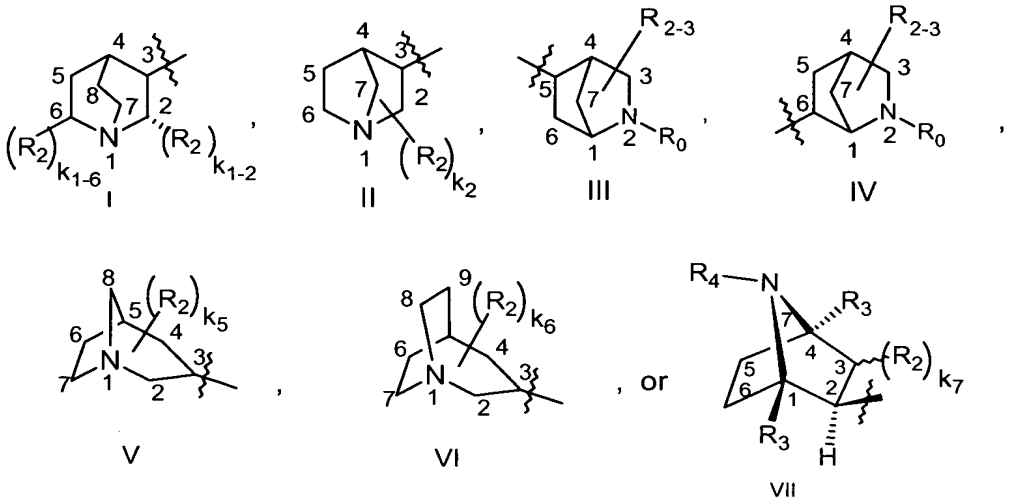
A pharmaceutical combination therapy composition can include therapeutically effective amounts of the compounds of Formula I, noted herein, and a therapeutically effective amount of the inhibitor. The combined administration of the compounds of Formula I and the inhibitor is expected to require less of the generally-prescribed dose for either agent when used alone and or is expected to result in less frequent administration of either, both or all agents. These compositions may be formulated with common excipients, diluents or carriers, and compressed into tablets, or formulated elixirs or solutions for convenient oral administration or administered by intramuscular intravenous routes. The compounds can be administered rectally, topically, orally, sublingually, or parenterally and maybe formulated as sustained relief dosage forms and the like.

The present invention claims any compound that is a full agonists relative to nicotine of an α7 Nicotinic Acetylcholine Receptors (nAChRs) or α7 nAChR full agonists, described either herein or elsewhere. Alpha 7 nAChR full agonists of the present invention include, but are not limited to compounds of Formula I as described herein. The present invention includes the administration of an alpha 7 nAChR full agonists in combination with a cholinesterase, and/or a beta secretase inhibitor, and/or a gamma secretase inhibitor, including a combination of all three inhibitors administered with the α7 nAChR full agonist. Non-limiting examples of α7 nAChR full agonists include compounds of Formula I:

Azabicyclo-N($R_1$)-C(=X)-W

Formula I

wherein Azabicyclo is



wherein X is O, or S;

$R_0$ is H, lower alkyl, substituted lower alkyl, or lower haloalkyl;

Each $R_1$ is H, alkyl, cycloalkyl, haloalkyl, substituted phenyl, or substituted naphthyl;

Each $R_2$ is independently F, Cl, Br, I, alkyl, substituted alkyl, haloalkyl, cycloalkyl, aryl, or $R_2$ is absent provided that $k_{1-2}$, $k_{1-6}$, $k_2$, $k_5$, $k_6$, or $k_7$ is 0;

$k_{1-2}$ is 0 or 1;

$k_{1-6}$ is 0 or 1, provided that the sum of $k_{1-2}$ and $k_{1-6}$ is one;

$k_2$ is 0 or 1;

$k_5$ is 0, 1, or 2;

$k_6$ is 0, 1, or 2;

$k_7$ is 0 or 1;

$R_{2-3}$ is H, F, Cl, Br, I, alkyl, haloalkyl, substituted alkyl, cycloalkyl, or aryl;

Each $R_3$ is independently H, alkyl, or substituted alkyl;

$R_4$ is H, alkyl, an amino protecting group, or an alkyl group having 1-3 substituents selected from F, Cl, Br, I, -OH, -CN, $-NH_2$, -NH(alkyl), or $-N(alkyl)_2$;

Lower alkyl is both straight- and branched-chain moieties having from 1-4 carbon atoms;

Lower haloalkyl is lower alkyl having 1 to (2n+1) substituent(s) independently selected from F, Cl, Br, or I where n is the maximum number of carbon atoms in the moiety;

Lower substituted alkyl is lower alkyl having 0-3 substituents independently selected from F, Cl, Br, or I and further having 1 substituent selected from $R_5$, $R_6$,

-CN, -$NO_2$, -$OR_8$, -$SR_8$, -$N(R_8)_2$, -$C(O)R_8$, -$C(O)OR_8$, -$C(S)R_8$, -$C(O)N(R_8)_2$, -$NR_8C(O)N(R_8)_2$, -$NR_8C(O)R_8$, -$S(O)R_8$, -$S(O)_2R_8$, -$OS(O)_2R_8$, -$S(O)_2N(R_8)_2$, -$NR_8S(O)_2R_8$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Alkyl is both straight- and branched-chain moieties having from 1-6 carbon atoms;

Haloalkyl is alkyl having 1 to (2n+1) substituent(s) independently selected from F, Cl, Br, or I where n is the maximum number of carbon atoms in the moiety;

Substituted alkyl is alkyl having 0-3 substituents independently selected from F, Cl, Br, or I and further having 1 substituent selected from $R_5$, $R_6$, -CN, -$NO_2$, -$OR_8$, -$SR_8$, -$N(R_8)_2$, -$C(O)R_8$, -$C(O)OR_8$, -$C(S)R_8$, -$C(O)N(R_8)_2$, -$NR_8C(O)N(R_8)_2$, -$NR_8C(O)R_8$, -$S(O)R_8$, -$S(O)_2R_8$, -$OS(O)_2R_8$, -$S(O)_2N(R_8)_2$, -$NR_8S(O)_2R_8$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Alkenyl is straight- and branched-chain moieties having from 2-6 carbon atoms and having at least one carbon-carbon double bond;

Haloalkenyl is alkenyl having 1 to (2n-1) substituent(s) independently selected from F, Cl, Br, or I where n is the maximum number of carbon atoms in the moiety;

Substituted alkenyl is alkenyl having 0-3 substituents independently selected from F, or Cl, and further having 1 substituent selected from $R_5$, $R_6$, -CN, -$NO_2$, -$OR_8$, -$SR_8$, -$N(R_8)_2$, -$C(O)R_8$, -$C(O)OR_8$, -$C(S)R_8$, -$C(O)N(R_8)_2$, -$NR_8C(O)N(R_8)_2$, -$NR_8C(O)R_8$, -$S(O)R_8$, -$S(O)_2R_8$, -$OS(O)_2R_8$, -$S(O)_2N(R_8)_2$, -$NR_8S(O)_2R_8$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Alkynyl is straight- and branched-chained moieties having from 2-6 carbon atoms and having at least one carbon-carbon triple bond;

Haloalkynyl is alkynyl having 1 to (2n-3) substituent(s) independently selected from F, Cl, Br, or I where n is the maximum number of carbon atoms in the moiety;

Substituted alkynyl is alkynyl having 0-3 substituents independently selected from F, or Cl, and further having 1 substituent selected from $R_5$, $R_6$, -CN, -$NO_2$, -$OR_8$, -$SR_8$, -$N(R_8)_2$, -$C(O)R_8$, -$C(O)OR_8$, -$C(S)R_8$, -$C(O)N(R_8)_2$, -$NR_8C(O)N(R_8)_2$,

$-NR_8C(O)R_8$, $-S(O)R_8$, $-S(O)_2R_8$, $-OS(O)_2R_8$, $-S(O)_2N(R_8)_2$, $-NR_8S(O)_2R_8$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Cycloalkyl is a cyclic alkyl moiety having from 3-6 carbon atoms;

Halocycloalkyl is cycloalkyl having 1-4 substituents independently selected from F, or Cl;

Substituted cycloalkyl is cycloalkyl having 0-3 substituents independently selected from F, or Cl, and further having 1 substituent selected from $R_5$, $R_6$, -CN, $-NO_2$, $-OR_8$, $-SR_8$, $-N(R_8)_2$, $-C(O)R_8$, $-C(O)OR_8$, $-C(S)R_8$, $-C(O)N(R_8)_2$, $-NR_8C(O)N(R_8)_2$, $-NR_8C(O)R_8$, $-S(O)R_8$, $-S(O)_2R_8$, $-OS(O)_2R_8$, $-S(O)_2N(R_8)_2$, $-NR_8S(O)_2R_8$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Heterocycloalkyl is a cyclic moiety having 4-7 atoms with 1-2 atoms within the ring being -S-, $-N(R_{10})-$, or -O-;

Haloheterocycloalkyl is heterocycloalkyl having 1-4 substituents independently selected from F, or Cl;

Substituted heterocycloalkyl is heterocycloalkylhaving 0-3 substituents independently selected from F, or Cl, and further having 1 substituent selected from $R_5$, $R_6$, -CN, $-NO_2$, $-OR_8$, $-SR_8$, $-N(R_8)_2$, $-C(O)R_8$, $-C(O)OR_8$, $-C(S)R_8$, $-C(O)N(R_8)_2$, $-NR_8C(O)N(R_8)_2$, $-NR_8C(O)R_8$, $-S(O)R_8$, $-S(O)_2R_8$, $-OS(O)_2R_8$, $-S(O)_2N(R_8)_2$, $-NR_8S(O)_2R_8$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Lactam heterocycloalkyl is a cyclic moiety having from 4-7 atoms with one atom being only nitrogen with the bond to the lactam heterocycloalkyl thru said atom being only nitrogen and having a =O on a carbon adjacent to said nitrogen, and having up to 1 additional ring atom being oxygen, sulfur, or nitrogen and further having 0-2 substituents selected from F, Cl, Br, I, or $R_7$ where valency allows;

Aryl is phenyl, substituted phenyl, naphthyl, or substituted naphthyl;

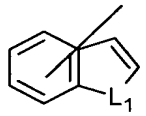
Substituted phenyl is a phenyl either having 1-4 substituents independently selected from F, Cl, Br, or I, or having 1 substituent selected from $R_{11}$ and 0-3 substituents independently selected from F, Cl, Br, or I;

Substituted naphthyl is a naphthalene moiety either having 1-4 substituents independently selected from F, Cl, Br, or I, or having 1 substituent selected from

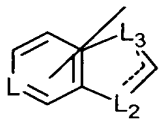
$R_{11}$ and 0-3 substituents independently selected from F, Cl, Br, or I, where the substitution can be independently on either only one ring or both rings of said naphthalene moiety;

Substituted phenoxy is a phenoxy either having 1-3 substituents independently selected from F, Cl, Br, or I, or having 1 substituent selected from $R_{11}$ and 0-2 substituents independently selected from F, Cl, Br, or I;

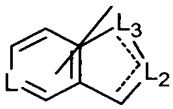
$R_5$ is 5-membered heteroaromatic mono-cyclic moieties containing within the ring 1-3 heteroatoms independently selected from the group consisting of -O-, =N-, -N($R_{10}$)-, and -S-, and having 0-1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I, or $R_5$ is 9-membered fused-ring moieties having a 6-membered ring fused to a 5-membered ring and having the formula




wherein $L_1$ is O, S, or $NR_{10}$,




wherein L is $CR_{12}$ or N, $L_2$ and $L_3$ are independently selected from $CR_{12}$, $C(R_{12})_2$, O, S, N, or $NR_{10}$, provided that both $L_2$ and $L_3$ are not simultaneously O, simultaneously S, or simultaneously O and S, or




wherein L is $CR_{12}$ or N, and $L_2$ and $L_3$ are independently selected from $CR_{12}$, O, S, N, or $NR_{10}$, and each 9-membered fused-ring moiety having 0-1 substituent selected from $R_9$ and further having 0-3 substituent(s) independently selected from F, Cl, Br, or I, wherein the $R_5$ moiety attaches to other substituents as defined in formula I at any position as valency allows;

$R_6$ is 6-membered heteroaromatic mono-cyclic moieties containing within the ring 1-3 heteroatoms selected from =N- and having 0-1 substituent selected from $R_9$ and 0-3 substituent(s) independently selected from F, Cl, Br, or I, or $R_6$ is 10-membered heteroaromatic bi-cyclic moieties containing within one or both rings 1-3

heteroatoms selected from =N-, including, but not limited to, quinolinyl or isoquinolinyl, each 10-membered fused-ring moiety having 0-1 substituent selected from $R_9$ and 0-3 substituent(s) independently selected from F, Cl, Br, or I, wherein the $R_6$ moiety attaches to other substituents as defined in formula I at any position as valency allows;

$R_7$ is alkyl, substituted alkyl, haloalkyl, $-OR_{11}$, -CN, $-NO_2$, $-N(R_8)_2$;

Each $R_8$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, alkyl substituted with 1 substituent selected from $R_{13}$, cycloalkyl substituted with 1 substituent selected from $R_{13}$, heterocycloalkyl substituted with 1 substituent selected from $R_{13}$, haloalkyl, halocycloalkyl, haloheterocycloalkyl, phenyl, or substituted phenyl;

$R_9$ is alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, haloheterocycloalkyl, $-OR_{14}$, $-SR_{14}$, $-N(R_{14})_2$, $-C(O)R_{14}$, $-C(O)N(R_{14})_2$, -CN, $-NR_{14}C(O)R_{14}$, $-S(O)_2N(R_{14})_2$, $-NR_{14}S(O)_2R_{14}$, $-NO_2$, alkyl substituted with 1-4 substituent(s) independently selected from F, Cl, Br, I, or $R_{13}$, cycloalkyl substituted with 1-4 substituent(s) independently selected from F, Cl, Br, I, or $R_{13}$, or heterocycloalkyl substituted with 1-4 substituent(s) independently selected from F, Cl, Br, I, or $R_{13}$;

$R_{10}$ is H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, phenyl, or phenyl having 1 substituent selected from $R_7$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

Each $R_{11}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, or haloheterocycloalkyl;

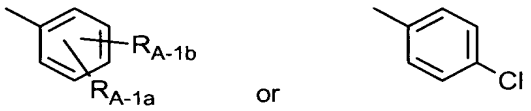
Each $R_{12}$ is independently H, F, Cl, Br, I, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, haloheterocycloalkyl, substituted alkyl, substituted cycloalkyl, substituted heterocycloalkyl, -CN, $-NO_2$, $-OR_{14}$, $-SR_{14}$, $-N(R_{14})_2$, $-C(O)R_{14}$, $-C(O)N(R_{14})_2$, $-NR_{14}C(O)R_{14}$, $-S(O)_2N(R_{14})_2$, $-NR_{14}S(O)_2RR_{14}$, or a bond directly or indirectly attached to the core molecule, provided that there is only one said bond to the core molecule within the 9-membered fused-ring moiety, further provided that where valency allows the fused-ring moiety has 0-1 substituent selected from alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, haloheterocycloalkyl, substituted alkyl, substituted cycloalkyl, substituted heterocycloalkyl, $-OR_{14}$, $-SR_{14}$, $-N(R_{14})_2$, $-C(O)R_{14}$, $-NO_2$, $-C(O)N(R_{14})_2$, -CN, $-NR_{14}C(O)R_{14}$, $-S(O)_2N(R_{14})_2$, or

$-NR_{14}S(O)_2R_{14}$, and further provided that the fused-ring moiety has 0-3 substituent(s) selected from F, Cl, Br, or I;

$R_{13}$ is $-OR_{14}$, $-SR_{14}$, $-N(R_{14})_2$, $-C(O)R_{14}$, $-C(O)N(R_{14})_2$, $-CN$, $-CF_3$, $-NR_{14}C(O)R_{14}$, $-S(O)_2N(R_{14})_2$, $-NR_{14}S(O)_2R_{14}$, or $-NO_2$;

Each $R_{14}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, or haloheterocycloalkyl;

wherein W is (A):




(A-1) (A-2)

wherein $R_{A-1a}$ is H, alkyl, alkenyl, alkynyl, cycloalkyl, heterocycloalkyl, haloalkyl, haloalkenyl, haloalkynyl, halocycloalkyl, haloheterocycloalkyl, substituted alkyl, substituted alkenyl, substituted alkynyl, substituted cycloalkyl, substituted heterocycloalkyl, aryl, $-R_5$, $R_6$, $-OR_{A-3}$, $-OR_{A-4}$, $-SR_{A-3}$, F, Cl, Br, I, $-N(R_{A-3})_2$, $-N(R_{A-5})_2$, $-C(O)R_{A-3}$, $-C(O)R_{A-5}$, $-CN$, $-C(O)N(R_{A-3})_2$, $-C(O)N(R_{A-6})_2$, $-NR_{A-3}C(O)R_{A-3}$, $-S(O)R_{A-3}$, $-OS(O)_2R_{A-3}$, $-NR_{A-3}S(O)_2R_{A-3}$, $-NO_2$, and $-N(H)C(O)N(H)R_{A-3}$;

$R_{A-1b}$ is $-O-R_{A-3}$, $-S-R_{A-3}$, $-S(O)-R_{A-3}$, $-C(O)-R_{A-7}$, and alkyl substituted on the $\omega$ carbon with $R_{A-7}$ where said $\omega$ carbon is determined by counting the longest carbon chain of the alkyl moiety with the C-1 carbon being the carbon attached to the phenyl ring attached to the core molecule and the $\omega$ carbon being the carbon furthest from said C-1 carbon;

Each $R_{A-3}$ is independently selected from H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, halo-heterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or substituted phenyl;

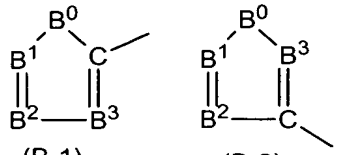
$R_{A-4}$ is selected from cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, or substituted heterocycloalkyl;

Each $R_{A-5}$ is independently selected from cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or substituted phenyl;

Each $R_{A-6}$ is independently selected from alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, halo-heterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or substituted phenyl;

$R_{A-7}$ is selected from aryl, $R_5$, or $R_6$;

wherein W is (B):




wherein $B^0$ is -O-, -S-, or -N($R_{B-0}$)-;

$B^1$ and $B^2$ are independently selected from =N-, or =C($R_{B-1}$)-;

$B^3$ is =N-, or =CH-, provided that when both $B^1$ and $B^2$ are =C($R_{B-1}$)- and $B^3$ is =CH-, only one =C($R_{B-1}$)- can be =CH-, and further provided that when $B^0$ is -O-, $B^2$ is =C($R_{B-1}$)- and $B^3$ is =C(H)-, $B^1$ cannot be =N-,

$R_{B-0}$ is H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, haloheterocycloalkyl, substituted alkyl, limited substituted alkyl, substituted cycloalkyl, substituted heterocycloalkyl, or aryl, and provided that when B is (B-2) and $B^3$ is =N- and $B^0$ is N($R_{B-0}$), $R_{B-0}$ cannot be phenyl or substituted phenyl;

$R_{B-1}$ is H, alkyl, alkenyl, alkynyl, cycloalkyl, heterocycloalkyl, haloalkyl, haloalkenyl, haloalkynyl, halocycloalkyl, haloheterocycloalkyl, substituted alkyl, substituted alkenyl, substituted alkynyl, substituted cycloalkyl, substituted heterocycloalkyl, limited substituted alkyl, limited substituted alkenyl, limited substituted alkynyl, aryl, -$OR_{B-2}$, -$OR_{B-3}$, -$SR_{B-2}$, -$SR_{B-3}$, F, Cl, Br, I, -$N(R_{B-2})_2$, -$N(R_{B-3})_2$, -$C(O)R_{B-2}$, -$C(O)R_{B-3}$, -$C(O)N(R_{B-2})_2$, -$C(O)N(R_{B-3})_2$, -CN, -$NR_{B-2}C(O)R_{B-4}$, -$S(O)_2N(R_{B-2})_2$, -$OS(O)_2R_{B-4}$, -$S(O)_2R_{B-2}$, -$S(O)_2R_{B-3}$, -$NR_{B-2}S(O)_2R_{B-2}$, -N(H)C(O)N(H)$R_{B-2}$, -$NO_2$, $R_5$, and $R_6$;

Each $R_{B-2}$ is independently H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or substituted phenyl;

Each $R_{B-3}$ is independently H, alkyl, haloalkyl, limited substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl;

$R_{B-4}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, or haloheterocycloalkyl;

wherein W is (C):

(C) is a six-membered heterocyclic ring system having 1-2 nitrogen atoms or a 10-membered bicyclic-six-six-fused-ring system having up to two nitrogen atoms within either or both rings, provided that no nitrogen is at a bridge of the bicyclic-six-six-fused-ring system, and further having 1-2 substitutents independently selected from $R_{C-1}$;

Each $R_{C-1}$ is independently H, F, Cl, Br, I, alkyl, haloalkyl, substituted alkyl, alkenyl, haloalkenyl, substituted alkenyl, alkynyl, haloalkynyl, substituted alkynyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, lactam heterocycloalkyl, phenyl, substituted phenyl, $-NO_2$, -CN, $-OR_{C-2}$, $-SR_{C-2}$, $-SOR_{C-2}$, $-SO_2R_{C-2}$, $-NR_{C-2}C(O)R_{C-3}$, $-NR_{C-2}C(O)R_{C-2}$, $-NR_{C-2}C(O)R_{C-4}$, $-N(R_{C-2})_2$, $-C(O)R_{C-2}$, $-C(O)_2R_{C-2}$, $-C(O)N(R_{C-2})_2$, -SCN, $-NR_{C-2}C(O)R_{C-2}$, $-S(O)N(R_{C-2})_2$, $-S(O)_2N(R_{C-2})_2$, $-NR_{C-2}S(O)_2R_{C-2}$, $R_5$, or $R_6$;

Each $R_{C-2}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, alkyl substituted with 1 substituent selected from $R_{C-5}$, cycloalkyl substituted with 1 substituent selected from $R_{C-5}$, heterocycloalkyl substituted with 1 substituent selected from $R_{C-5}$, haloalkyl, halocycloalkyl, haloheterocycloalkyl, phenyl, or substituted phenyl;

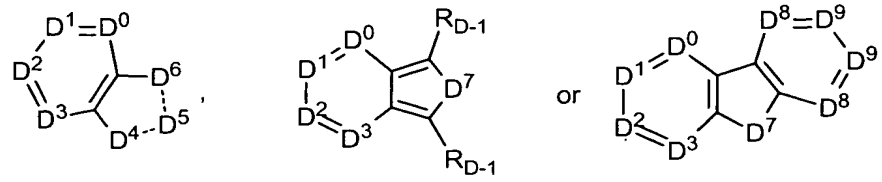
Each $R_{C-3}$ is independently H, alkyl, or substituted alkyl;

$R_{C-4}$ is H, alkyl, an amino protecting group, or an alkyl group having 1-3 substituents selected from F, Cl, Br, I, -OH, -CN, $-NH_2$, -NH(alkyl), or $-N(alkyl)_2$;

$R_{C-5}$ is -CN, $-CF_3$, $-NO_2$, $-OR_{C-6}$, $-SR_{C-6}$, $-N(R_{C-6})_2$, $-C(O)R_{C-6}$, $-SOR_{C-6}$, $-SO_2RR_{C-6}$, $-C(O)N(R_{C-6})_2$, $-NR_{C-6}C(O)R_{C-6}$, $-S(O)_2N(R_{C-6})_2$, or $-NR_{C-6}S(O)_2R_{C-6}$;

Each $R_{C-6}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, or haloheterocycloalkyl;

wherein W is (D):



provided that the bond between the -C(=X)- group and the W group may be attached at any available carbon atom within the D group as provided in $R_{D-1}$, $R_{D-3}$, and $R_{D-4}$;

$D^0$, $D^1$, $D^2$, and $D^3$ are N or $C(R_{D-1})$ provided that up to one of $D^0$, $D^1$, $D^2$, or $D^3$ is N and the others are $C(R_{D-1})$, further provided that when the core molecule is attached at $D^2$ and $D^0$ or $D^1$ is N, $D^3$ is C(H), and further provided that there is only one attachment to the core molecule;

$D^4$---$D^5$---$D^6$ is selected from $N(R_{D-2})$-$C(R_{D-3})$=$C(R_{D-3})$, N=$C(R_{D-3})$-$C(R_{D-4})_2$, $C(R_{D-3})$=$C(R_{D-3})$-$N(R_{D-2})$, $C(R_{D-3})_2$-$N(R_{D-2})$-$C(R_{D-3})_2$, $C(R_{D-4})_2$-$C(R_{D-3})$=N, $N(R_{D-2})$-$C(R_{D-3})_2$-$C(R_{D-3})_2$, $C(R_{D-3})_2$-$C(R_{D-3})_2$-$N(R_{D-2})$, O-$C(R_{D-3})$=$C(R_{D-3})$, O-$C(R_{D-3})_2$-$C(R_{D-3})_2$, $C(R_{D-3})_2$-O-$C(R_{D-3})_2$, $C(R_{D-3})$=$C(R_{D-3})$-O, $C(R_{D-3})_2$-$C(R_{D-3})_2$-O, S-$C(R_{D-3})$=$C(R_{D-3})$, S-$C(R_{D-3})_2$-$C(R_{D-3})_2$, $C(R_{D-3})_2$-S-$C(R_{D-3})_2$, $C(R_{D-3})$=$C(R_{D-3})$-S, or $C(R_{D-3})_2$-$C(R_{D-3})_2$-S;

provided that when C(X) is attached to W at $D^2$ and $D^6$ is O, $N(R_{D-2})$, or S, $D^4$---$D^5$ is not CH=CH;

and further provided that when C(X) is attached to W at $D^2$ and $D^4$ is O, $N(R_{D-2})$, or S, $D^5$---$D^6$ is not CH=CH;

Each $R_{D-1}$ is independently H, F, Br, I, Cl, -CN, $-CF_3$, $-OR_{D-5}$, $-SR_{D-5}$, $-N(R_{D-5})_2$, or a bond to -C(X)- provided that only one of $R_{D-1}$, $R_{D-3}$, and $R_{D-4}$ is said bond;

Each $R_{D-2}$ is independently H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, or $R_6$;

Each $R_{D-3}$ is independently H, F, Br, Cl, I, alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, heterocycloalkyl, substituted heterocycloalkyl, lactam heterocycloalkyl, -CN, $-NO_2$, $-OR_{D-10}$, $-C(O)N(R_{D-11})_2$, $-NR_{D-10}COR_{D-12}$, $-N(R_{D-10})_2$, $-SR_{D-10}$, $-S(O)_2R_{D-10}$, $-C(O)R_{D-12}$, $-CO_2R_{D-10}$, aryl, $R_5$, $R_6$, a bond to -C(X)- provided that only one of $R_{D-1}$, $R_{D-3}$, and $R_{D-4}$ is said bond;

Each $R_{D-4}$ is independently H, F, Br, Cl, I, alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, heterocycloalkyl, substituted heterocycloalkyl, lactam heterocycloalkyl, -CN, -$NO_2$, -$OR_{D-10}$, -$C(O)N(R_{D-11})_2$, -$NR_{D-10}COR_{D-12}$, -$N(R_{D-11})_2$, -$SR_{D-10}$, -$CO_2R_{D-10}$, aryl, $R_5$, $R_6$, a bond to -C(X)- provided that only one of $R_{D-1}$, $R_{D-3}$, and $R_{D-4}$ is said bond;

Each $R_{D-5}$ is independently H, $C_{1-3}$ alkyl, or $C_{2-4}$ alkenyl;

$D^7$ is O, S, or $N(R_{D-2})$;

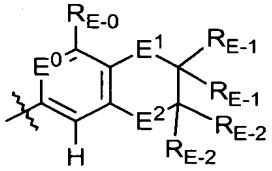
$D^8$ and $D^9$ are $C(R_{D-1})$, provided that when the molecule is attached to the phenyl moiety at $D^9$, $D^8$ is CH;

Each $R_{D-10}$ is H, alkyl, cycloalkyl, haloalkyl, substituted phenyl, or substituted naphthyl;

Each $R_{D-11}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, alkyl substituted with 1 substituent selected from $R_{13}$, cycloalkyl substituted with 1 substituent selected from $R_{13}$, heterocycloalkyl substituted with 1 substituent selected from $R_{13}$, haloalkyl, halocycloalkyl, haloheterocycloalkyl, phenyl, or substituted phenyl;

$R_{D-12}$ is H, alkyl, substituted alkyl, cycloalkyl, haloalkyl, heterocycloalkyl, substituted heterocycloalkyl, substituted phenyl, or substituted naphthyl;

wherein W is (E):




$E^0$ is CH or N;

$R_{E-0}$ is H, F, Cl, Br, I, alkyl, alkenyl, alkynyl, cycloalkyl, heterocycloalkyl, haloalkyl, haloalkenyl, haloalkynyl, halocycloalkyl, haloheterocycloalkyl, substituted alkyl, substituted alkenyl, substituted alkynyl, substituted cycloalkyl, substituted heterocycloalkyl, aryl, $R_5$, $R_6$, -$OR_{E-3}$, -$OR_{E-4}$, -$SR_{E-3}$, -$SR_{E-5}$, -$N(R_{E-3})_2$, -$NR_{E-3}R_{E-6}$, -$N(R_{E-6})_2$, -$C(O)R_{E-3}$, -CN, -$C(O)N(R_{E-3})_2$, -$NR_{E-3}C(O)R_{E-3}$, -$S(O)R_{E-3}$, -$S(O)R_{E-5}$, -$OS(O)_2R_{E-3}$, -$NR_{E-3}S(O)_2R_{E-3}$, -$NO_2$, or -$N(H)C(O)N(H)R_{E-3}$;

$E^1$ is O, $CR_{E-1-1}$, or $C(R_{E-1-1})_2$, provided that when $E^1$ is $CR_{E-1-1}$, one $R_{E-1}$ is a bond to $CR_{E-1-1}$, and further provided that at least one of $E^1$ or $E^2$ is O;

Each $R_{E-1-1}$ is independently H, F, Br, Cl, CN, alkyl, haloalkyl, substituted alkyl, alkynyl, cycloalkyl, $-OR_E$, or $-N(R_E)_2$, provided that at least one $R_{E-1-1}$ is H when $E^1$ is $C(R_{E-1-1})_2$;

Each $R_{E-1}$ is independently H, alkyl, substituted alkyl, haloalkyl, cycloalkyl, heterocycloalkyl, or a bond to $E^1$ provided that $E^1$ is $CR_{E-1-1}$;

$E^2$ is O, $CR_{E-2-2}$, or $C(R_{E-2-2})_2$, provided that when $E^2$ is $CR_{E-2-2}$, one $R_{E-2}$ is a bond to $CR_{E-2-2}$, and further provided that at least one of $E^1$ or $E^2$ is O;

Each $R_{E-2-2}$ is independently H, F, Br, Cl, CN, alkyl, haloalkyl, substituted alkyl, alkynyl, cycloalkyl, $-OR_E$, or $-N(R_E)_2$, provided that at least one $R_{E-2-2}$ is H when $E^2$ is $C(R_{E-2-2})_2$;

Each $R_{E-2}$ is independently H, alkyl, substituted alkyl, haloalkyl, cycloalkyl, heterocycloalkyl, or a bond to $E^2$ provided that $E^2$ is $CR_{E-2-2}$;

Each $R_E$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, or haloheterocycloalkyl;

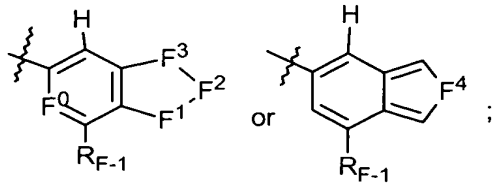
Each $R_{E-3}$ is independently H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I or substituted phenyl;

$R_{E-4}$ is H, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or substituted phenyl;

Each $R_{E-5}$ is independently H, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, or $R_6$;

Each $R_{E-6}$ is independently alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, $R_5$, $R_6$, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

wherein W is (F):



(F-1) (F-2)

$F^0$ is C(H) wherein $F^1$---$F^2$---$F^3$ is selected from O-C($R_{F-2}$)=N, O-C($R_{F-3}$)($R_{F-2}$)-N($R_{F-4}$), O-C($R_{F-3}$)($R_{F-2}$)-S, O-N=C($R_{F-3}$), O-C($R_{F-2}$)($R_{F-5}$)-O, O-C($R_{F-2}$)($R_{F-3}$)-O, S-C($R_{F-2}$)=N, S-C($R_{F-3}$)($R_{F-2}$)-N($R_{F-4}$), S-N=C($R_{F-3}$), N=C($R_{F-2}$)-O, N=C($R_{F-2}$)-S, N=C($R_{F-2}$)-N($R_{F-4}$), N($R_{F-4}$)-N=C($R_{F-3}$), N($R_{F-4}$)-C($R_{F-3}$)($R_{F-2}$)-O, N($R_{F-4}$)-C($R_{F-3}$)($R_{F-2}$)-S, N($R_{F-4}$)-C($R_{F-3}$)($R_{F-2}$)-N($R_{F-4}$), C($R_{F-3}$)$_2$-O-N($R_{F-4}$), C($R_{F-3}$)$_2$-N($R_{F-4}$)-O, C($R_{F-3}$)$_2$-N($R_{F-4}$)-S, C($R_{F-3}$)=N-O, C($R_{F-3}$)=N-S, C($R_{F-3}$)=N-N($R_{F-4}$), C($R_{F-3}$)($R_{F-6}$)-C($R_{F-2}$)($R_{F-6}$)-C($R_{F-3}$)($R_{F-6}$), or C($R_{F-3}$)$_2$-C($R_{F-2}$)($R_{F-3}$)-C($R_{F-3}$)$_2$;

$F^0$ is N wherein $F^1$---$F^2$---$F^3$ is selected from O-C($R_{F-2}$)=N, O-C($R_{F-3}$)($R_{F-2}$)-N($R_{F-4}$), O-C($R_{F-3}$)($R_{F-2}$)-S, O-N=C($R_{F-3}$) O-C($R_{F-2}$)($R_{F-3}$)-O, S-C($R_{F-2}$)=N, S-C($R_{F-3}$)($R_{F-2}$)-N($R_{F-4}$), S-N=C($R_{F-3}$), N=C($R_{F-2}$)-O, N=C($R_{F-2}$)-S, N=C($R_{F-2}$)-N($R_{F-4}$), N($R_{F-4}$)-N=C($R_{F-3}$), N($R_{F-4}$)-C($R_{F-3}$)($R_{F-2}$)-O, N($R_{F-4}$)-C($R_{F-3}$)($R_{F-2}$)-S, N($R_{F-4}$)-C($R_{F-3}$)($R_{F-2}$)-N($R_{F-4}$), C($R_{F-3}$)$_2$-O-N($R_{F-4}$), C($R_{F-3}$)$_2$-N($R_{F-4}$)-O, C($R_{F-3}$)$_2$-N($R_{F-4}$)-S, C($R_{F-3}$)=N-O, C($R_{F-3}$)=N-S, C($R_{F-3}$)=N-N($R_{F-4}$), C($R_{F-3}$)=C($R_{F-2}$)-C($R_{F-3}$)$_2$, or C($R_{F-3}$)$_2$-C($R_{F-2}$)($R_{F-3}$)-C($R_{F-3}$)$_2$;

$F^4$ is N($R_{F-7}$), O, or S;

$R_{F-1}$ is H, F, Cl, Br, I, -CN, -$CF_3$, -O$R_{F-8}$, -S$R_{F-8}$, or -N($R_{F-8}$)$_2$;

$R_{F-2}$ is H, F, alkyl, haloalkyl, substituted alkyl, lactam heterocycloalkyl, phenoxy, substituted phenoxy, $R_5$, $R_6$, -N($R_{F-4}$)-aryl, -N($R_{F-4}$)-substituted phenyl, -N($R_{F-4}$)-substituted naphthyl, -O-substituted phenyl, -O-substituted naphthyl, -S-substituted phenyl, -S-substituted naphthyl, or alkyl substituted on the ω carbon with $R_{F-9}$ where said ω carbon is determined by counting the longest carbon chain of the alkyl moiety with the C-1 carbon being the carbon attached to W and the ω carbon being the carbon furthest, e.g., separated by the greatest number of carbon atoms in the chain, from said C-1 carbon;

$R_{F-3}$ is H, F, Br, Cl, I, alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, heterocycloalkyl, substituted heterocycloalkyl, lactam heterocycloalkyl, -CN, -$NO_2$, -O$R_{F-8}$,

-C(O)N($R_{F-8}$)$_2$, -NH$R_{F-8}$, -N$R_{F-8}$C(O)$R_{F-8}$, -N($R_{F-8}$)$_2$, -S$R_{F-8}$, -C(O)$R_{F-8}$, -$CO_2R_{F-8}$, aryl, $R_5$, or $R_6$;

$R_{F-4}$ is H, or alkyl;

Each $R_{F-5}$ is independently F, Br, Cl, I, alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, -CN, -$CF_3$, -O$R_{F-8}$, -C(O)$NH_2$, -NH$R_{F-8}$, -S$R_{F-8}$, -$CO_2R_{F-8}$, aryl, phenoxy, substituted phenoxy, heteroaryl, -N($R_{F-4}$)-aryl, or -O-substituted aryl;

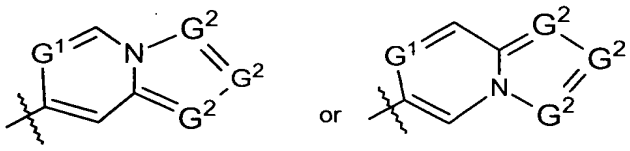
One of $R_{F-6}$ is H, alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, -CN, F, Br, Cl, I, -O$R_{F-8}$, -C(O)$NH_2$, -NH$R_{F-8}$, -S$R_{F-8}$, -$CO_2R_{F-8}$, aryl, $R_5$, or $R_6$, and each of the other two $R_{F-6}$ is independently selected from alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, -CN, F, Br, Cl, I, -O$R_{F-8}$, -C(O)$NH_2$, -NH$R_{F-8}$, -S$R_{F-8}$, -$CO_2R_{F-8}$, aryl, $R_5$, or $R_6$;

$R_{F-7}$ is H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, phenyl, or phenyl having 1 substituent selected from $R_9$ and further having 0-3 substituents independently selected from F, Cl, Br, or I;

$R_{F-8}$ is H, alkyl, substituted alkyl, cycloalkyl, haloalkyl, heterocycloalkyl, substituted heterocycloalkyl, substituted phenyl, or substituted naphthyl;

$R_{F-9}$ is aryl, $R_5$, or $R_6$;

wherein W is (G):




$G^1$ is N or CH;

Each $G^2$ is N or C($R_{G-1}$), provided that no more than one $G^2$ is N;

Each $R_{G-1}$ is independently H, alkyl, substituted alkyl, haloalkyl, alkenyl, substituted alkenyl, haloalkenyl, alkynyl, substituted alkynyl, haloalkynyl, -CN, -$NO_2$, F, Br, Cl, I, -C(O)N($R_{G-3}$)$_2$, -N($R_{G-3}$)$_2$, -S$R_{G-6}$, -S(O)$_2R_{G-6}$, -O$R_{G-6}$, -C(O)$R_{G-6}$, -$CO_2R_{G-6}$, aryl, $R_5$, $R_6$, or two $R_{G-1}$ on adjacent carbon atoms may combine for W to be a 6-5-6 fused-tricyclic-heteroaromatic-ring system optionally substituted on the newly formed ring where valency allows with 1-2 substitutents independently selected from F, Cl, Br, I, and $R_{G-2}$;

$R_{G-2}$ is alkyl, alkenyl, alkynyl, cycloalkyl, heterocycloalkyl, haloalkyl, haloalkenyl, haloalkynyl, halocycloalkyl, haloheterocycloalkyl, $-OR_{G-8}$, $-SR_{G-8}$, $-S(O)_2R_{G-8}$, $-S(O)R_{G-8}$, $-OS(O)_2R_{G-8}$, $-N(R_{G-8})_2$, $-C(O)R_{G-8}$, $-C(S)R_{G-8}$, $-C(O)OR_{G-8}$, $-CN$, $-C(O)N(R_{G-8})_2$, $-NR_{G-8}C(O)R_{G-8}$, $-S(O)_2N(R_{G-8})_2$, $-NR_{G-8}S(O)_2R_{G-8}$, $-NO_2$, $-N(R_{G-8})C(O)N(R_{G-8})_2$, substituted alkyl, substituted alkenyl, substituted alkynyl, substituted cycloalkyl, substituted heterocycloalkyl, lactam heterocycloalkyl, phenyl, phenyl having 0-4 substituents independently selected from F, Cl, Br, I and $R_{G-7}$, naphthyl, or naphthyl having 0-4 substituents independently selected from F, Cl, Br, I, or $R_{G-7}$;

provided that when$G^2$ adjacent to the bridge N is $C(R_{G-1})$ and the other $G^2$ are CH, that $R_{G-1}$ is other than H, F, Cl, I, alkyl, substituted alkyl or alkynyl;

Each $R_{G-3}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, alkyl substituted with 1 substituent selected from $R_{G-4}$, cycloalkyl substituted with 1 substituent selected from $R_{G-4}$, heterocycloalkyl substituted with 1 substituent selected from $R_{G-4}$, haloalkyl, halocycloalkyl, haloheterocycloalkyl, phenyl, or substituted phenyl;

$R_{G-4}$ is $-OR_{G-5}$, $-SR_{G-5}$, $-N(R_{G-5})_2$, $-C(O)R_{G-5}$, $-SOR_{G-5}$, $-SO_2R_{G-5}$, $-C(O)N(R_{G-5})_2$, $-CN$, $-CF_3$, $-NR_{G-5}C(O)R_{G-5}$, $-S(O)_2N(R_{G-5})_2$, $-NR_{G-5}S(O)_2R_{G-5}$, or $-NO_2$;

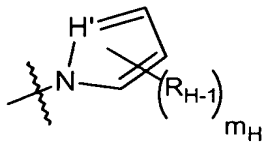
Each $R_{G-5}$ is independently H, alkyl, cycloalkyl, heterocycloalkyl, haloalkyl, halocycloalkyl, or haloheterocycloalkyl;

$R_{G-6}$ is H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, phenyl, or phenyl having 0-4 substituents independently selected from F, Cl, Br, I, and $R_{G-7}$;

$R_{G-7}$ is alkyl, substituted alkyl, haloalkyl, $-OR_{G-5}$, $-CN$, $-NO_2$, $-N(R_{G-3})_2$;

Each $R_{G-8}$ is independently H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, phenyl, or phenyl substituted with 0-4 independently selected from F, Cl, Br, I, or $R_{G-7}$;

wherein W is (H)



H' is N or CH;

Each $R_{H-1}$ is independently F, Cl, Br, I, -CN, -$NO_2$, alkyl, haloalkyl, substituted alkyl, alkenyl, haloalkenyl, substituted alkenyl, alkynyl, haloalkynyl, substituted alkynyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, lactam heterocyclcoalkyl, aryl, $R_5$, $R_6$, -$OR_8$, -$SR_8$, -$SOR_8$, -$SO_2R_8$, -SCN, -$S(O)N(R_8)_2$, -$S(O)_2N(R_8)_2$, -$C(O)R_8$, -$C(O)_2R_8$, -$C(O)N(R_8)_2$, $C(R_8)$=N-$OR_8$, -$NC(O)R_5$, -$NC(O)R_{H-3}$, -$NC(O)R_6$, -$N(R_8)_2$, -$NR_8C(O)R_8$, -$NR_8S(O)_2R_8$, or two $R_{H-1}$ on adjacent carbon atoms may fuse to form a 6-membered ring to give a 5-6 fused, bicyclic moiety where the 6-membered ring is optionally substituted with 1-3 substitutents selected from $R_{H-2}$;

$m_H$ is 0, 1, or 2;

$R_{H-2}$ is alkyl, alkenyl, alkynyl, cycloalkyl, heterocycloalkyl, haloalkyl, haloalkenyl, haloalkynyl, halocycloalkyl, haloheterocycloalkyl, -$OR_{H-3}$, -$SR_{H-3}$, -$S(O)_2R_{H-3}$, -$S(O)R_{H-3}$, -$OS(O)_2R_{H-3}$, -$N(R_{H-3})_2$, -$C(O)R_{H-3}$, -$C(S)R_{H-3}$, -$C(O)OR_{H-3}$, -CN, -$C(O)N(R_{H-3})_2$, -$NR_{H-3}C(O)R_{H-3}$, -$S(O)_2N(R_{H-3})_2$, -$NR_{H-3}S(O)_2R_{H-3}$, -$NO_2$, -$N(R_{H-3})C(O)N(R_{H-3})_2$, substituted alkyl, substituted alkenyl, substituted alkynyl, substituted cycloalkyl, substituted heterocycloalkyl, lactam heterocycloalkyl, phenyl, phenyl having 0-4 substituents independently selected from F, Cl, Br, I and $R_7$, naphthyl, naphthyl having 0-4 substituents independently selected from F, Cl, Br, I, or $R_7$, or two $R_{H-2}$ on adjacent carbon atoms may combine to form a three-ring-fused-5-6-6 system optionally substituted with up to 3 substituents independently selected from Br, Cl, F, I, -CN, -$NO_2$, -$CF_3$, -$N(R_{H-3})_2$, -$N(R_{H-3})C(O)R_{H-3}$, alkyl, alkenyl, and alkynyl;

Each $R_{H-3}$ is independently H, alkyl, haloalkyl, substituted alkyl, cycloalkyl, halocycloalkyl, substituted cycloalkyl, heterocycloalkyl, haloheterocycloalkyl, substituted heterocycloalkyl, phenyl, or phenyl substituted with 0-4 independently selected from F, Cl, Br, I, or $R_7$;

or pharmaceutical composition, pharmaceutically acceptable salt, racemic mixture, or pure enantiomer thereof.

The present invention is useful in the treatment of, or preparation of medicament(s) for the treatment of, a wide variety of disease and disorders where the alpha 7 nAChR is implicated, including cognitive and attention deficit symptoms of Alzheimer's, neurodegeneration associated with diseases such as Alzheimer's disease, pre-senile dementia (mild cognitive impairment), senile dementia, amyotrophic lateral sclerosis, traumatic brain injury, behavioral and cognitive problems in general and associated with brain tumors, AIDS dementia complex, dementia associated with Down's syndrome, dementia associated with Lewy Bodies, Huntington's disease, Parkinson's disease, age-related macular degeneration.

Abbreviations which are well known to one of ordinary skill in the art may be used (e.g., "Ph" for phenyl, "Me" for methyl, "Et" for ethyl, "h" or "hr" for hour or hours, "min" for minute or minutes, and "rt" for room temperature).

All temperatures are in degrees Centigrade.

Room temperature is within the range of 15-25 degrees Celsius.

AChR refers to acetylcholine receptor.

nAChR refers to nicotinic acetylcholine receptor.

Pre-senile dementia is also known as mild cognitive impairment.

$5HT_3R$ refers to the serotonin-type 3 receptor.

α-btx refers to α-bungarotoxin.

FLIPR refers to a device marketed by Molecular Devices, Inc. designed to precisely measure cellular fluorescence in a high throughput whole-cell assay. (Schroeder et. al., *J. Biomolecular Screening*, 1(2), p 75-80, 1996).

TLC refers to thin-layer chromatography.

HPLC refers to high pressure liquid chromatography.

MeOH refers to methanol.

EtOH refers to ethanol.

IPA refers to isopropyl alcohol.

THF refers to tetrahydrofuran.

DMSO refers to dimethylsulfoxide.

DMF refers to *N*,*N*-dimethylformamide.

EtOAc refers to ethyl acetate.

TMS refers to tetramethylsilane.

TEA refers to triethylamine.

DIEA refers to *N,N*-diisopropylethylamine.

MLA refers to methyllycaconitine.

Ether refers to diethyl ether.

HATU refers to O-(7-azabenzotriazol-1-yl)-N,N,N', N'-tetramethyluronium hexafluorophosphate.

CDI refers to carbonyl diimidazole.

NMO refers to N-methylmorpholine-N-oxide.

TPAP refers to tetrapropylammonium perruthenate.

$Na_2SO_4$ refers to sodium sulfate.

$K_2CO_3$ refers to potassium carbonate.

$MgSO_4$ refers to magnesium sulfate.

When $Na_2SO_4$, $K_2CO_3$, or $MgSO_4$ is used as a drying agent, it is anhydrous.

As used herein, "acetylcholinesterase inhibitor," or "beta secretase inhibitor" include their respective pharmaceutically acceptable salts, such as hydrochlorides, tartrates, and the like.

Halogen is F, Cl, Br, or I.

The carbon atom content of various hydrocarbon-containing moieties is indicated by a prefix designating the minimum and maximum number of carbon atoms in the moiety, i.e., the prefix $C_{i-j}$ indicates a moiety of the integer 'i" to the integer "j" carbon atoms, inclusive. Thus, for example, $C_{1-6}$ alkyl refers to alkyl of one to six carbon atoms.

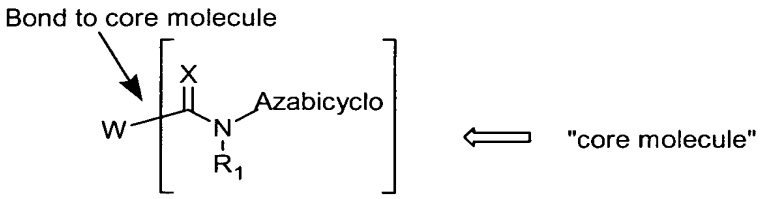
Non-inclusive examples of heteroaryl compounds that fall within the definition of $R_5$ and $R_6$ include, but are not limited to, thienyl, benzothienyl, pyridyl, thiazolyl, quinolyl, pyrazinyl, pyrimidyl, imidazolyl, furanyl, benzofuranyl, benzothiazolyl, isothiazolyl, benzisothiazolyl, benzisoxazolyl, benzimidazolyl, indolyl, benzoxazolyl, pyrazolyl, triazolyl, tetrazolyl, isoxazolyl, oxazolyl, pyrrolyl, isoquinolinyl, cinnolinyl, indazolyl, indolizinyl, phthalazinyl, pydridazinyl, triazinyl, isoindolyl, purinyl, oxadiazolyl, furazanyl, benzofurazanyl, benzothiophenyl, benzothiazolyl, quinazolinyl, quinoxalinyl, naphthridinyl, and furopyridinyl.

Non-inclusive examples of heterocycloalkyl include, but are not limited to, tetrahydrofurano, tetrahydropyrano, morpholino, pyrrolidino, piperidino, piperazine, azetidino, azetidinono, oxindolo, dihydroimidazolo, and pyrrolidinono

Some of the amines described herein require the use of an amine-protecting group to ensure functionalization of the desired nitrogen. One of ordinary skill in the art would appreciate where, within the synthetic scheme to use said protecting group. Amino protecting group includes, but is not limited to, carbobenzyloxy (CBz), *tert* butoxy carbonyl (BOC) and the like. Examples of other suitable amino protecting groups are known to person skilled in the art and can be found in "Protective Groups in Organic synthesis," 3rd Edition, authored by Theodora Greene and Peter Wuts.

Alkyl substituted on an ω carbon with $R_{A-7}$ is determined by counting the longest carbon chain of the alkyl moiety with the C-1 carbon being the carbon attached to the W moiety and the ω carbon being the carbon furthest, e.g., separated by the greatest number of carbon atoms in the chain, from said C-1 carbon. Therefore, when determining the ω carbon, the C-1 carbon will be the carbon attached, as valency allows, to the W moiety and the ω carbon will be the carbon furthest from said C-1 carbon.

The core molecule is Azabicyclo-N($R_1$)-C(=X)-:




Mammal denotes human and other mammals.

Brine refers to an aqueous saturated sodium chloride solution.

Equ means molar equivalents.

IR refers to infrared spectroscopy.

Lv refers to leaving groups within a molecule, including Cl, OH, or mixed anhydride.

NMR refers to nuclear (proton) magnetic resonance spectroscopy, chemical shifts are reported in ppm (δ) downfield from TMS.

MS refers to mass spectrometry expressed as m/e or mass/charge unit. HRMS refers to high resolution mass spectrometry expressed as m/e or mass/charge unit. $[M+H]^+$ refers to an ion composed of the parent plus a proton. $[M-H]^-$ refers to an ion composed of the parent minus a proton. $[M+Na]^+$ refers to an ion composed of the parent plus a sodium ion. $[M+K]^+$ refers to an ion composed of the parent plus a

potassium ion. EI refers to electron impact. ESI refers to electrospray ionization. CI refers to chemical ionization. FAB refers to fast atom bombardment.

Compounds of the present invention may be in the form of pharmaceutically acceptable salts. The term "pharmaceutically acceptable salts" refers to salts prepared from pharmaceutically acceptable non-toxic bases including inorganic bases and organic bases, and salts prepared from inorganic acids, and organic acids. Salts derived from inorganic bases include aluminum, ammonium, calcium, ferric, ferrous, lithium, magnesium, potassium, sodium, zinc, and the like. Salts derived from pharmaceutically acceptable organic non-toxic bases include salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines, such as arginine, betaine, caffeine, choline, N, N-dibenzylethylenediamine, diethylamine, 2-diethylaminoethanol, 2-dimethylamino-ethanol, ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, hydrabamine, isopropylamine, lysine, methylglucamine, morpholine, piperazine, piperidine, polyamine resins, procaine, purines, theobromine, triethylamine, trimethylamine, tripropylamine, and the like. Salts derived from inorganic acids include salts of hydrochloric acid, hydrobromic acid, hydroiodic acid, sulfuric acid, phosphoric acid, phosphorous acid and the like. Salts derived from pharmaceutically acceptable organic non-toxic acids include salts of $C_{1-6}$ alkyl carboxylic acids, di-carboxylic acids, and tri-carboxylic acids such as acetic acid, propionic acid, fumaric acid, succinic acid, tartaric acid, maleic acid, adipic acid, and citric acid, and aryl and alkyl sulfonic acids such as toluene sulfonic acids and the like.

By the term "effective amount" of a compound as provided herein is meant a nontoxic but sufficient amount of the compound(s) to provide the desired therapeutic effect. As pointed out below, the exact amount required will vary from subject to subject, depending on the species, age, and general condition of the subject, the severity of the disease that is being treated, the particular compound(s) used, the mode of administration, and the like. Thus, it is not possible to specify an exact "effective amount." However, an appropriate effective amount may be determined by one of ordinary skill in the art using only routine experimentation.

In addition to the compound(s) of Formula I, the compositions use may also comprise one or more non-toxic, pharmaceutically acceptable carrier materials or

excipients. A generally recognized compendium of such methods and ingredients is Remington's Pharmaceutical Sciences by E.W. Martin (Mark Publ. Co., 15th Ed., 1975). The term "carrier" material or "excipient" herein means any substance, not itself a therapeutic agent, used as a carrier and/or diluent and/or adjuvant, or vehicle for delivery of a therapeutic agent to a subject or added to a pharmaceutical composition to improve its handling or storage properties or to permit or facilitate formation of a dose unit of the composition into a discrete article such as a capsule or tablet suitable for oral administration. Excipients can include, by way of illustration and not limitation, diluents, disintegrants, binding agents, adhesives, wetting agents, polymers, lubricants, glidants, substances added to mask or counteract a disagreeable taste or odor, flavors, dyes, fragrances, and substances added to improve appearance of the composition. Acceptable excipients include lactose, sucrose, starch powder, cellulose esters of alkanoic acids, cellulose alkyl esters, talc, stearic acid, magnesium stearate, magnesium oxide, sodium and calcium salts of phosphoric and sulfuric acids, gelatin, acacia gum, sodium alginate, polyvinyl-pyrrolidone, and/or polyvinyl alcohol, and then tableted or encapsulated for convenient administration. Such capsules or tablets may contain a controlled-release formulation as may be provided in a dispersion of active compound in hydroxypropyl-methyl cellulose, or other methods known to those skilled in the art. For oral administration, the pharmaceutical composition may be in the form of, for example, a tablet, capsule, suspension or liquid. If desired, other active ingredients may be included in the composition.

In addition to the oral dosing, noted above, the compositions of the present invention may be administered by any suitable route, e.g., parenterally, bucal, intravaginal, and rectal, in the form of a pharmaceutical composition adapted to such a route, and in a dose effective for the treatment intended. Such routes of administration are well known to those skilled in the art. The compositions may, for example, be administered parenterally, e.g., intravascularly, intraperitoneally, subcutaneously, or intramuscularly. For parenteral administration, saline solution, dextrose solution, or water may be used as a suitable carrier. Formulations for parenteral administration may be in the form of aqueous or non-aqueous isotonic sterile injection solutions or suspensions. These solutions and suspensions may be prepared from sterile powders or granules having one or more of the carriers or diluents mentioned for use in the formulations for oral administration. The

compounds may be dissolved in water, polyethylene glycol, propylene glycol, EtOH, corn oil, cottonseed oil, peanut oil, sesame oil, benzyl alcohol, sodium chloride, and/or various buffers. Other adjuvants and modes of administration are well and widely known in the pharmaceutical art.

The serotonin type 3 receptor ($5HT_3R$) is a member of a superfamily of ligand-gated ion channels, which includes the muscle and neuronal nAChR, the glycine receptor, and the γ-aminobutyric acid type A receptor. Like the other members of this receptor superfamily, the $5HT_3R$ exhibits a large degree of sequence homology with α7 nAChR but functionally the two ligand-gated ion channels are very different. For example, α7 nAChR is rapidly inactivated, is highly permeable to calcium and is activated by acetylcholine and nicotine. On the other hand, $5HT_3R$ is inactivated slowly, is relatively impermeable to calcium and is activated by serotonin. These experiments suggest that the α7 nAChR and $5HT_3R$ proteins have some degree of homology, but function very differently. Indeed the pharmacology of the channels is very different. For example, Ondansetron, a highly selective $5HT_3R$ antagonist, has little activity at the α7 nAChR. The converse is also true. For example, GTS-21, a highly selective α7 nAChR full agonist, has little activity at the $5HT_3R$.

α7 nAChR is a ligand-gated $Ca^{++}$ channel formed by a homopentamer of α7 subunits. Previous studies have established that α-bungarotoxin (α-btx) binds selectively to this homopetameric, α7 nAChR subtype, and that α7 nAChR has a high affinity binding site for both α-btx and methyllycaconitine (MLA). α7 nAChR is expressed at high levels in the hippocampus, ventral tegmental area and ascending cholinergic projections from nucleus basilis to thalamocortical areas. α7 nAChR full agonists increase neurotransmitter release, and increase cognition, arousal, attention, learning and memory.

Data from human and animal pharmacological studies establish that nicotinic cholinergic neuronal pathways control many important aspects of cognitive function including attention, learning and memory (Levin, E.D., *Psychopharmacology*, 108:417-31, 1992; Levin, E.D. and Simon B.B., *Psychopharmacology*, 138:217-30, 1998). For example, it is well known that nicotine increases cognition and attention in humans. ABT-418, a compound that activates α4β2 and α7 nAChR, improves cognition and attention in clinical trials of Alzheimer's disease and attention-deficit disorders (Potter, A. et. al., *Psychopharmacology (Berl).*, 142(4):334-42, Mar. 1999;

Wilens, T. E. et. al., *Am. J. Psychiatry*, 156(12):1931-7, Dec. 1999). It is also clear that nicotine and selective but weak α7 nAChR full agonists increase cognition and attention in rodents and non-human primates.

Selective α7 nAChR full agonists may be found using a functional assay on FLIPR (see WO 00/73431 A2). FLIPR is designed to read the fluorescent signal from each well of a 96 or 384 well plate as fast as twice a second for up to 30 minutes. This assay may be used to accurately measure the functional pharmacology of α7 nAChR and $5HT_3R$. To conduct such an assay, one uses cell lines that expressed functional forms of the α7 nAChR using the α7/5-$HT_3$ channel as the drug target and cell lines that expressed functional $5HT_3R$. In both cases, the ligand-gated ion channel was expressed in SH-EP1 cells. Both ion channels can produce robust signal in the FLIPR assay.

As discussed, the compounds of the present invention are α7 nAChR full agonists. Therefore, as another aspect of the present invention, the compounds of the present invention may be used to treat a variety of diseases including cognitive and attention deficit symptoms of Alzheimer's, neurodegeneration associated with diseases such as Alzheimer's disease, pre-senile dementia (also known as mild cognitive impairment), and senile dementia.

Alzheimer's disease has many aspects, including cognitive and attention deficits. Currently, these deficits are treated with cholinesterase inhibitors. These inhibitors slow the break down of acetylcholine, and thereby provide a general nonspecific increase in the activity of the cholinergic nervous system. Since the drugs are nonspecific, they have a wide variety of side effects. Thus, there is a need for a drug that stimulates a portion of the cholinergic pathways and thereby provides improvement in the cognitive and attention deficits associated with Alzheimer's disease without the side effects created by nonspecific stimulation of the cholinergic pathways.

Neurodegeneration is a common problem associated with diseases such as Alzheimer's disease. While the current drugs treat some of the symptoms of this disease, they do not control the underlying pathology of the disease. Accordingly, it would be desirable to provide a drug that can slow the progress of Alzheimer's disease.

Pre-senile dementia (mild cognitive impairment) concerns memory impairment rather than attention deficit problems and otherwise unimpaired cognitive functioning. Mild cognitive impairment is distinguished from senile dementia in that mild cognitive impairment involves a more persistent and troublesome problem of memory loss for the age of the patient. There currently is no medication specifically identified for treatment of mild cognitive impairment, due somewhat to the newness of identifying the disease. Therefore, there is a need for a drug to treat the memory problems associated with mild cognitive impairment.

Senile dementia is not a single disease state. However, the conditions classified under this name frequently include cognitive and attention deficits. Generally, these deficits are not treated. Accordingly, there is a need for a drug that provides improvement in the cognitive and attention deficits associated with senile dementia.

As discussed, the compounds of the present invention are α7 nAChR full agonists. Therefore, yet other diseases to be treated with compounds of the present invention include treating the cognitive and attention deficits as well as the neurodegeneration associated with any one or more or combination of the following: amyotrophic lateral sclerosis, traumatic brain injury, behavioral and cognitive problems associated with brain tumors, AIDS dementia complex, dementia associated with Down's syndrome, dementia associated with Lewy Bodies, Huntington's disease, Parkinson's disease, age-related macular degeneration.

Amyotrophic lateral sclerosis, also known as Lou Gehrig's disease, belongs to a class of disorders known as motor neuron diseases wherein specific nerve cells in the brain and spinal cord gradually degenerate to negatively affect the control of voluntary movement. Currently, there is no cure for amyotrophic lateral sclerosis although patients may receive treatment from some of their symptoms and although Riluzole has been shown to prolong the survival of patients. Therefore, there is a need for a pharmaceutical agent to treat this disease.

Traumatic brain injury occurs when the brain is damaged from a sudden physical assault on the head. Symptoms of the traumatic brain injury include confusion and other cognitive problems. Therefore, there is a need to address the symptoms of confusion and other cognitive problems.

Brain tumors are abnormal growths of tissue found inside of the skull. Symptoms of brain tumors include behavioral and cognitive problems. Surgery, radiation, and chemotherapy are used to treat the tumor, but other agents are necessary to address associated symptoms. Therefore, there is a need to address the symptoms of behavioral and cognitive problems.

Acquired immune deficiency syndrome (AIDS) results from an infection with the human immunodeficiency virus (HIV). This virus attacks selected cells and impairs the proper function of the immune, nervous, and other systems. HIV infection can cause other problems such as, but not limited to, difficulties in thinking, otherwise known as AIDS dementia complex. Therefore, there is a need to drugs to relieve the confusion and mental decline of persons with AIDS.

Persons with Down's syndrome have in all or at least some of their cells an extra, critical portion of the number 21 chromosome. Adults who have Down's syndrome are known to be at risk for Alzheimer-type dementia. Currently, there is no proven treatment for Down's syndrome. Therefore, there is a need to address the dementia associated with Down's syndrome.

Dementia with Lewy Bodies is a neurodegenerative disorder involving abnormal structures known as Lewy bodies found in certain areas of the brain. Symptoms of dementia with Lewy bodies include, but are not limited to, fluctuating cognitive impairment with episodic delirium. Currently, treatment concerns addressing the parkinsonian and psychiatric symptoms. However, medicine to control tremors or loss of muscle movement may actually accentuate the underlying disease of dementia with Lewy bodies. Therefore, there is a need of a pharmaceutical agent to treat dementia with Lewy bodies.

Genetically programmed degeneration of neurons in certain areas of the brain cause Huntington's disease. Early symptoms of Huntington's disease include mood swings, or trouble learning new things or remembering a fact. Most drugs used to treat the symptoms of Huntington's disease have side effects such as fatigue, restlessness, or hyperexcitability. Currently, there is no treatment to stop or reverse the progression of Huntington's disease. Therefore, there is a need of a pharmaceutical agent to address the symptoms with fewer side effects.

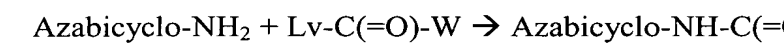
Parkinson's disease is a neurological disorder characterized by tremor, hypokinesia, and muscular rigidity. Currently, there is no treatment to stop the

progression of the disease. Therefore, there is a need of a pharmaceutical agent to address Parkinson's.

The key step in the preparation of this class of compounds is the coupling of the Azabicyclo moiety with the requisite acid chloride (Lv = Cl), mixed anhydride (e.g., Lv = diphenyl phosphoryl, bis(2-oxo-3-oxazolidinyl)phosphinyl, or acyloxy of the general formula of O-C(O)-$R_{Lv}$, where $R_{Lv}$ includes phenyl or t-butyl), or carboxylic acid (Lv =OH) in the presence of an activating reagent. Suitable activating reagents are well known in the art, for examples see Kiso, Y., Yajima, H. "Peptides" pp. 39-91, San Diego, CA, Academic Press, (1995), and include, but are not limited to, agents such as carbodiimides, phosphonium and uronium salts (such as HATU).

Compounds of Formula I can be prepared as shown in Scheme 1. The key step in the preparation of this class of compounds is the coupling of an azabicyclic moiety with the requisite acid chloride (Lv = Cl), mixed anhydride (e.g., Lv = diphenyl phosphoryl, bis(2-oxo-3-oxazolidinyl)phosphinyl, or acyloxy of the general formula of O-C(O)-$R_{Lv}$, where $R_{Lv}$ includes phenyl or t-butyl), or carboxylic acid (Lv =OH) in the presence of an activating reagent. Suitable activating reagents are well known in the art, for examples see Kiso, Y., Yajima, H. "Peptides" pp. 39-91, San Diego, CA, Academic Press, (1995), and include, but are not limited to, agents such as carbodiimides, phosphonium and uronium salts (such as HATU).

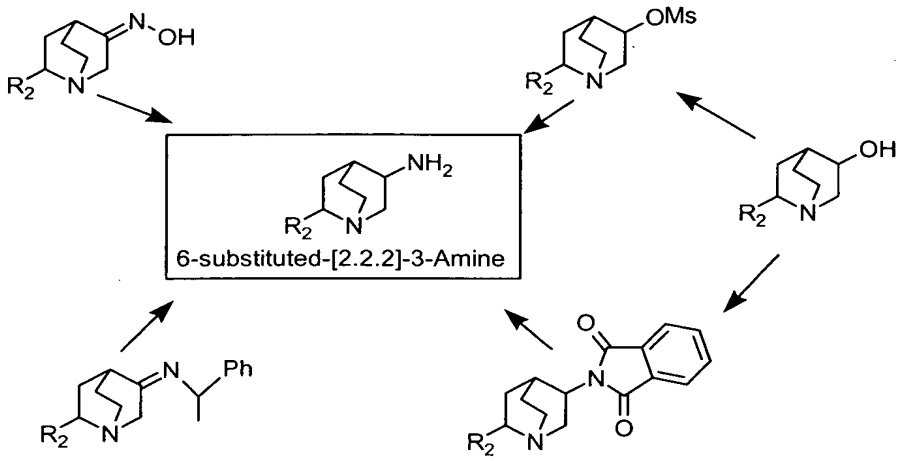
Scheme 1




Generally, the carboxylic acid is activated with a uronium salt, preferably HATU (see *J. Am. Chem. Soc.*, 4397 (1993)), in the presence of the Azabicyclico moiety and a base such as DIEA in DMF to afford the desired amides. Alternatively, the carboxylic acid is converted to the acyl azide by using DPPA; the appropriate amine precursor is added to a solution of the appropriate anhydride or azide to give the desired final compounds. In some cases, the ester (Lv being OMe or OEt) may be reacted directly with the amine precursor in refluxing methanol or ethanol to give the compounds of Formula I.

Certain 6-substituted-[2.2.2]-3-amines (Azabicyclo I) are known in the art. The preparation of compounds where $R_2$ is present is described in *Acta Pol. Pharm.* 179-85 (1981). Alternatively, the 6-substituted-[2.2.2]-3-amine can be prepared by

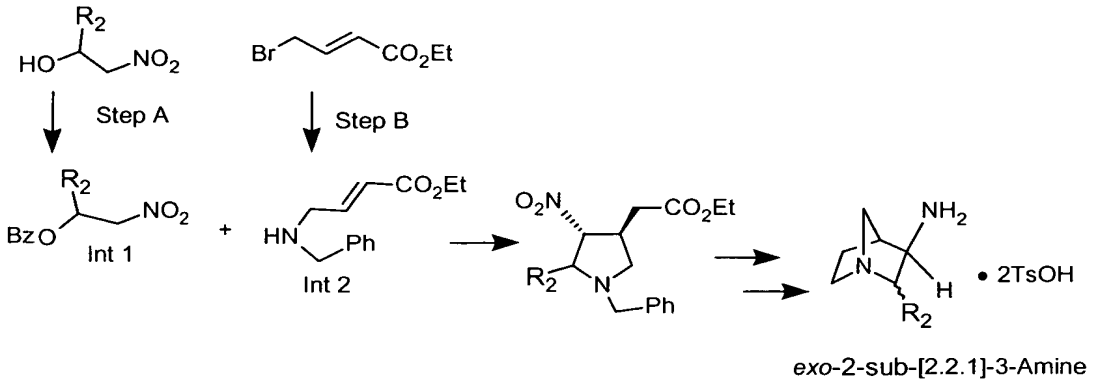
reduction of an oxime or an imine of the corresponding 6-substituted-3-quinuclidinone by methods known to one of ordinary skill in the art (see *J. Labelled Compds. Radiopharm.*, 53-60 (1995), *J. Med. Chem.* 988-995, (1998), *Synth. Commun.* 1895-1911 (1992), *Synth. Commun.* 2009-2015 (1996)). Alternatively, the 6-substituted-[2.2.2]-3-amine can be prepared from a 6-substituted-3-hydroxyquinuclidine by Mitsunobu reaction followed by deprotection as described in *Synth. Commun.* 1895-1911 (1995). Alternatively, the 6-substituted-[2.2.2]-3-amine can be prepared by conversion of a 6-substituted-3-hydroxyquinuclidine into the corresponding mesylate or tosylate, followed by displacement with sodium azide and reduction as described in *J. Med. Chem.* 587-593 (1975).



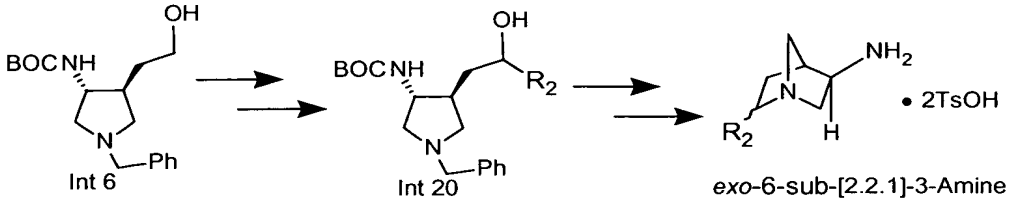

The oximes can be prepared by treatment of the 3-quinuclidinones with hydroxylamine hydrochloride in the presence of base. The imines can be prepared by treatment of the 3-quinuclidinones with a primary amine under dehydrating conditions. The 3-hydroxyquinuclidines can be prepared by reduction of the 3-quinuclidinones. The 6-substituted-3-quinuclidinones can be prepared by known procedures (see *J. Gen. Chem. Russia* 3791-3795, (1963), *J. Chem. Soc. Perkin Trans. I* 409-420 (1991), *J. Org. Chem.* 3982-3996 (2000)).

One of ordinary skill in the art will recognize that the methods described for the reaction of the unsubstituted 3-amino-1-azabicyclo[2.2.1]heptane ($R_2$=absent) are equally applicable to substituted compounds ($R_2$ is present). For where Azabicyclo is II, compounds where $R_2$ is present can be prepared from appropriately substituted nitro alcohols using procedures described in *Tetrahedron* (1997), 53, p. 11121 as shown below. Methods to synthesize nitro alcohols are well known in the art (see *J. Am. Chem. Soc.* (1947), 69, p 2608). The scheme below is a modification of the

synthesis of *exo*-3-amino-1-azabicyclo[2.2.1]heptane as the bis(hydro para-toluenesulfonate) salt, described in detail herein, to show how to obtain these amine precursors. The desired salt can be made using standard procedures.

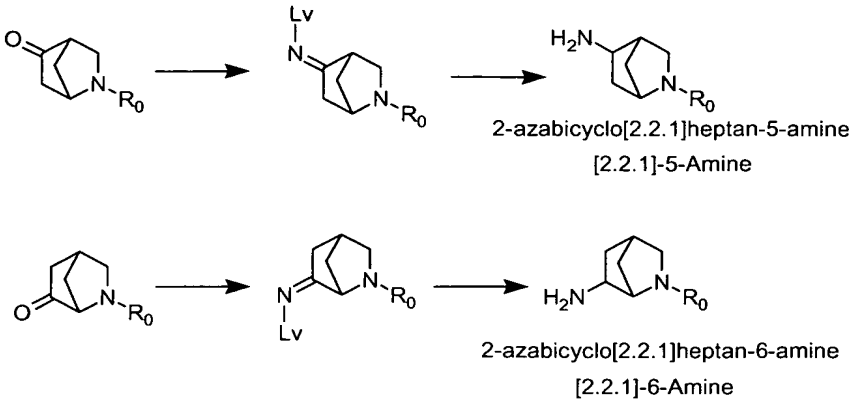



Compounds for Azabicyclo II where $R_2$ is present can also be prepared by modification of intermediates described in the synthesis of *exo*-3-amino-1-azabicyclo[2.2.1]heptane as the bis(hydro para-toluenesulfonate) salt, described in detail herein. For example, Int 6 can be oxidized to the aldehyde and treated with an organometallic reagent to provide Int 20 using procedures described in *Tetrahedron*, (1999), 55, p 13899. Int 20 can be converted into the amine using methods described for the synthesis of *exo*-3-amino-1-azabicyclo[2.2.1]heptane as the bis(hydro para-toluenesulfonate) salt. Once the amine is obtained, the desired salt can be made using standard procedures.




The schemes used are for making *exo*-3-amino-1-azabicyclo[2.2.1]heptane. However, the modifications discussed are applicable to make the *endo* isomer also.

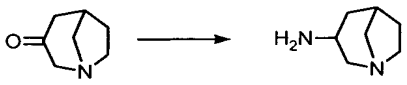
There are several methods by which the amine precursor for Azabicyclo III and Azabicyclo IV can be obtained:




where Lv can be $-CH_2Ph$, -CH(Me)Ph, -OH, -OMe, or $-OCH_2Ph$.

The respective amine precursors for Azabicyclo III and Azabicyclo IV can be prepared by reduction of an oxime or an imine of the corresponding *N*-2-azabicyclo[2.2.1]-heptanone by methods known to one skilled in the art (see *J. Labelled Compds. Radiopharm.*, 53-60 (1995), *J. Med. Chem.* 988-995, (1998), *Synth. Commun.* 1895-1911 (1992), *Synth. Commun.* 2009-2015 (1996)). The oximes can be prepared by treatment of the *N*-2-azabicyclo[2.2.1]heptanones with hydroxylamine hydrochloride in the presence of a base. The imines can be prepared by treatment of the *N*-2-azabicyclo[2.2.1]-heptanones with a primary amine under dehydrating conditions. The *N*-2-azabicyclo[2.2.1]heptanones can be prepared by known procedures (see *Tet. Lett.* 1419-1422 (1999), *J. Med. Chem.* 2184-2191 (1992), *J. Med. Chem.* 706-720 (2000), *J. Org. Chem.*, 4602-4616 (1995)).

The *exo*- and *endo*-1-azabicyclo[3.2.1]octan-3-amines are prepared from 1-azabicyclic[3.2.1]octan-3-one (Thill, B. P., Aaron, H. S., *J. Org. Chem.*, 4376-4380 (1968)) according to the general procedure as discussed in Lewin, A.H., et al., *J. Med. Chem.*, 988-995 (1998).




One of ordinary skill in the art will also recognize that the methods described for the reaction of the unsubstituted 1-azabicyclo[3.2.1]octan-3-amine or 1-azabicyclo[3.2.2]nonan-3-amine ($R_2$=absent) are equally applicable to substituted compounds ($R_2$ is present). The $R_2$ substituent may be introduced as known to one skilled in the art through standard alkylation chemistry. Exposure of 1-azabicyclo[3.2.1]octan-3-one or 1-azabicyclo[3.2.2]nonan-3-one to a hindered base such as LDA (lithium diisopropylamide) in a solvent such as THF or ether between

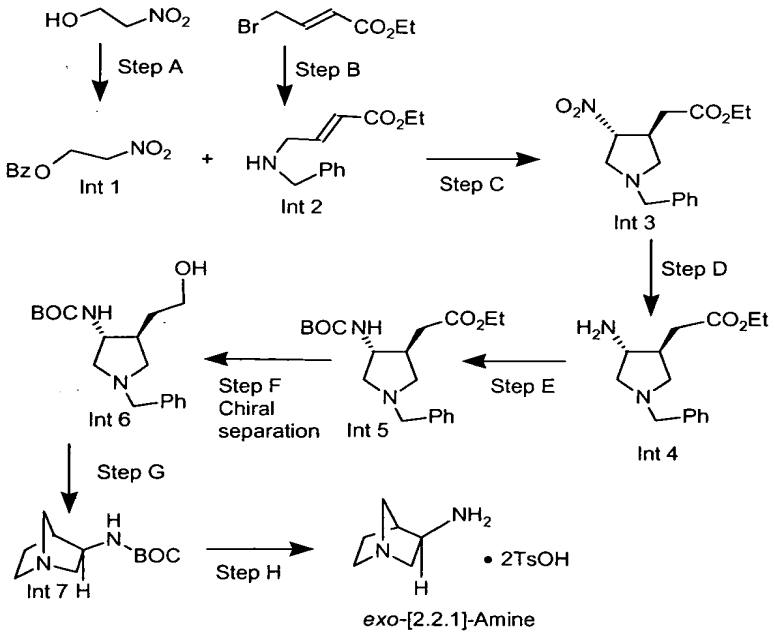
0°C to -78°C followed by the addition of an alkylating agent ($R_2Lv$, where Lv = Cl, Br, I, OTs, etc.) will, after being allowed to warm to about 0°C to rt followed by an aqueous workup, provide the desired compound as a mixture of isomers. Chromatographic resolution (flash, HPLC, or chiral HPLC) will provided the desired purified alkylated ketones. From there, formation of the oxime and subsequent reduction will provide the desired *endo* or *exo* isomers.

## AMINES

Preparation of *N*-(2*S*,3*R*)-2-methyl-1-azabicyclo[2.2.2]octan-3-amine dihydrochloride (2*S*-methyl-2.2.2-Amine): See, e.g., US 20020042428 A1.

**Preparation of the 1-azabicyclo-2.2.1 Amines:**

Synthesis of *exo*-3-amino-1-azabicyclo[2.2.1]heptane as the bis(hydro para-toluenesulfonate) salt (*exo*-[2.2.1]-Amine):




Step A. Preparation of 2-(benzoyloxy)-1-nitroethane (Int 1).

Benzoyl chloride (14.9 mL, 128 mmol) is added to a stirred solution of nitroethanol (9.2 mL, 128 mmol) in dry benzene (120 mL). The solution is refluxed for 24 hr and then concentrated *in vacuo*. The crude product is purified by flash chromatography on silica gel. Elution with hexanes-EtOAc (80:20) affords Int 1 as a white solid (68% yield): $^1$H NMR ($CDCl_3$) δ 8.0, 7.6, 7.4, 4.9, 4.8.

Step B. Preparation of ethyl *E*-4-(benzylamino)-2-butenoate (Int 2).

Ethyl E-4-bromo-2-butenoate (10 mL, 56 mmol, tech grade) is added to a stirred solution of benzylamine (16 mL, 146 mmol) in $CH_2Cl_2$ (200 mL) at rt. The reaction mixture stirs for 15 min, and is diluted with ether (1 L). The mixture is washed with saturated aqueous $NaHCO_3$ solution (3x) and water, dried ($Na_2SO_4$), filtered and concentrated *in vacuo*. The residue is purified by flash chromatography on silica gel. Elution with hexanes-EtOAc (70:30) affords Int 2 as a clear oil (62% yield): $^1$H NMR ($CDCl_3$) δ 7.4-7.2, 7.0, 6.0, 4.2, 3.8, 3.4, 2.1-1.8, 1.3.

Step C. Preparation of *trans*-4-nitro-1-(phenylmethyl)-3-pyrrolidineacetic acid ethyl ester (Int 3).

A solution of Int 1 (6.81 g, 34.9 mmol) and Int 2 (7.65 g, 34.9 mmol) in EtOH (70 mL) stirs at rt for 15 h and is then concentrated *in vacuo*. The residue is diluted with ether (100 mL) and saturated aqueous $NaHCO_3$ solution (100 mL). The organic layer is separated and dried ($Na_2SO_4$), filtered and concentrated *in vacuo*. The crude product is purified by flash chromatography on silica gel. Elution with hexanes-EtOAc (85:15) affords Int 3 as a clear oil (76% yield): $^1$H NMR ($CDCl_3$) δ 7.4-7.3, 4.8-4.7, 4.1, 3.8-3.6, 3.3-3.0, 2.7-2.6, 2.4-2.3, 1.2.

Step D. Preparation of *trans*-4-amino-1-(phenylmethyl)-3-pyrrolidineacetic acid ethyl ester (Int 4).

A mixture of Int 3 (3.28 g, 11.2 mmol) and RaNi (1.5 g) in EtOH (100 mL) is placed in a Parr bottle and hydrogenated for 4 h under an atmosphere of hydrogen (46 psi) at rt. The mixture is filtered through a pad of Celite, and the solvent is removed *in vacuo* to afford Int 4 as a clear oil (100% yield): $^1$H NMR (300 MHz, $CDCl_3$) δ 7.3-7.2, 4.1, 3.6, 3.2, 3.0-2.9, 2.8, 2.8-2.6, 2.6-2.4, 2.30-2.2, 1.2.

Step E. Preparation of *trans*-4-(1,1-dimethylethoxycarbonylamido)-1-(phenylmethyl)-3-pyrrolidineacetic acid ethyl ester (Int 5).

Di-*tert*-butyldicarbonate (3.67 g, 16.8 mmol) is added to a stirred solution of Int 4 (2.94 g, 11.2 mmol) in $CH_2Cl_2$ (30 mL) cooled in an ice bath. The reaction is allowed to warm to rt and stirred overnight. The mixture is concentrated *in vacuo*.

The crude product is purified by flash chromatography on silica gel. Elution with hexanes-EtOAc (80:20) affords Int 5 as a white solid (77% yield): $^1$H NMR (300 MHz, $CDCl_3$) δ 7.4-7.2, 5.1-4.9, 4.1, 4.0-3.8, 3.6, 3.2-3.0, 2.8-2.6, 2.5-2.4, 2.3-2.1, 1.4, 1.3.

Step F. Preparation of *trans* (*tert*-butoxycarbonylamino)-4-(2-hydroxyethyl)-1-(N-phenylmethyl) pyrrolidine (Int 6).

$LiAlH_4$ powder (627 mg, 16.5 mmol) is added in small portions to a stirred solution of Int 5 (3.0 g, 8.3 mmol) in anhydrous THF (125 mL) in a -5°C bath. The mixture is stirred for 20 min in a –5°C bath, then quenched by the sequential addition of water (0.6 mL), 15% (w/v) aqueous NaOH (0.6 mL) and water (1.8 mL). Excess anhydrous $K_2CO_3$ is added, and the mixture is stirred for 1 h, then filtered. The filtrate is concentrated *in vacuo*. The residue is purified by flash chromatography on silica gel. Elution with EtOAc affords Int 6 as a white solid (94% yield): $^1$H NMR ($CDCl_3$) δ 7.4-7.3, 5.3-5.2, 4.1-4.0, 3.9-3.7, 3.3-3.2, 2.8-2.7, 2.3-2.1, 1.7, 1.5.

Int 6 is a racemic mixture that can be resolved via chromatography using a Diacel chiral pack AD column. From the two enantiomers thus obtained, the (+)-enantiomer, $[\alpha]^{25}_D$ +35 (c 1.0, MeOH), gives rise to the corresponding enantiomerically pure *exo*-4-*S* final compounds, whereas the (-)-enantiomer, $[\alpha]^{25}_D$ -34 (c 0.98, MeOH), gives rise to enantiomerically pure *exo*-4-*R* final compounds. The methods described herein use the (+)-enantiomer of Int 6 to obtain the enantiomerically pure *exo*-4-*S* final compounds. However, the methods used are equally applicable to the (-)-enantiomer of Int 6, making non-critical changes to the methods provided herein to obtain the enantiomerically pure *exo*-4-*R* final compounds.

Step G. Preparation of *exo* 3-(*tert*-butoxycarbonylamino)-1-azabicyclo[2.2.1]heptane (Int 7).

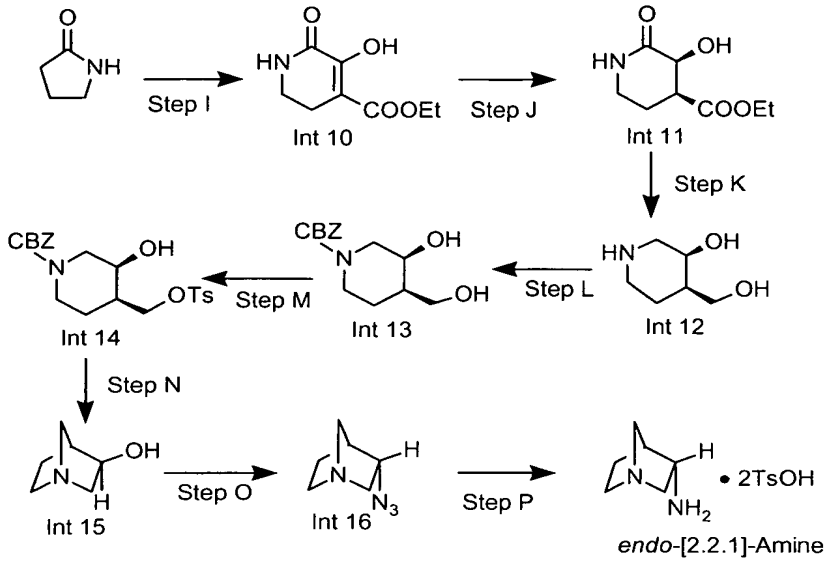
TEA (8.0 g, 78.9 mml) is added to a stirred solution of Int 6 (2.5 g, 7.8 mmol) in $CH_2Cl_2$ (50 mL), and the reaction is cooled in an ice-water bath. $CH_3SO_2Cl$ (5.5 g, 47.8 mmol) is then added dropwise, and the mixture is stirred for 10 min in an ice-water bath. The resulting yellow mixture is diluted with saturated aqueous $NaHCO_3$

solution, extracted with $CH_2Cl_2$ several times until no product remains in the aqueous layer by TLC. The organic layers are combined, washed with brine, dried ($Na_2SO_4$) and concentrated *in vacuo*. The residue is dissolved in EtOH (85 mL) and is heated to reflux for 16 h. The reaction mixture is allowed to cool to rt, transferred to a Parr bottle and treated with 10% Pd/C catalyst (1.25 g). The bottle is placed under an atmosphere of hydrogen (53 psi) for 16 h. The mixture is filtered through Celite, and fresh catalyst (10% Pd/C, 1.25 g) is added. Hydrogenolysis continues overnight. The process is repeated three more times until the hydrogenolysis is complete. The final mixture is filtered through Celite and concentrated *in vacuo*. The residue is purified by flash chromatography on silica gel. Elution with $CHCl_3$-MeOH-$NH_4OH$ (90:9.5:0.5) affords Int 7 as a white solid (46% yield): $^1$H NMR ($CDCl_3$) δ 5.6-5.5, 3.8-3.7, 3.3-3.2, 2.8-2.7, 2.0-1.8, 1.7-1.5, 1.5.

Step H. Preparation of *exo*-3-amino-1-azabicyclo[2.2.1]heptane bis(hydro-*para*-toluenesulfonate).

*Para*-toluenesulfonic acid monohydrate (1.46 g, 7.68 mmol) is added to a stirred solution of Int 7 (770 mg, 3.63 mmol) in EtOH (50 mL). The reaction mixture is heated to reflux for 10 h, followed by cooling to rt. The precipitate is collected by vacuum filtration and washed with cold EtOH to give *exo*-[2.2.1]-Amine as a white solid (84% yield): $^1$H NMR ($CD_3OD$) δ 7.7, 7.3, 3.9-3.7, 3.7-3.3, 3.2, 2.4, 2.3-2.2, 1.9-1.8.

Synthesis of *endo*-3-amino-1-azabicyclo[2.2.1]heptane as the bis(hydro para-toluenesulfonate) salt (*endo*-[2.2.1]-Amine):




Step I. Preparation of ethyl 5-hydroxy-6-oxo-1,2,3,6-tetrahydropyridine-4-carboxylate (Int 10).

Absolute EtOH (92.0 mL, 1.58 mol) is added to a mechanically stirred suspension of potassium ethoxide (33.2 g, 395 mmol) in dry toluene (0.470 L). When the mixture is homogeneous, 2-pyrrolidinone (33.6 g, 395 mmol) is added, and then a solution of diethyl oxalate (53.1 mL, 390 mmol) in toluene (98 mL) is added via an addition funnel. After complete addition, toluene (118 mL) and EtOH (78 mL) are added sequentially. The mixture is heated to reflux for 18 h. The mixture is cooled to rt and aqueous HCl (150 mL of a 6.0 M solution) is added. The mixture is mechanically stirred for 15 min. The aqueous layer is extracted with $CH_2Cl_2$, and the combined organic layers are dried ($MgSO_4$), filtered and concentrated *in vacuo* to a yellow residue. The residue is recrystallized from EtOAc to afford Int 10 as a yellow solid (38% yield): $^1$H NMR ($CDCl_3$) δ 11.4, 7.4, 4.3, 3.4, 2.6, 1.3.

Step J. Preparation of ethyl *cis*-3-hydroxy-2-oxopiperidine-4-carboxylate (Int 11).

A mixture of Int 10 (15 g, 81 mmol) and 5% rhodium on carbon (2.0 g) in glacial acetic acid is placed under an atmosphere of hydrogen (52 psi). The mixture is shaken for 72 h. The mixture is filtered through Celite, and the filtrate is concentrated *in vacuo* to afford Int 11 as a white solid (98% yield): $^1$H NMR ($CDCl_3$) δ 6.3, 4.2, 4.0-3.8, 3.4, 3.3-3.2, 2.2, 1.3.

Step K. Preparation of *cis*- 4-(hydroxymethyl)piperidin-3-ol (Int 12).

Int 11 (3.7 g, 19.9 mmol) as a solid is added in small portions to a stirred solution of $LiAlH_4$ in THF (80 mL of a 1.0 M solution) in an ice-water bath. The mixture is warmed to rt, and then the reaction is heated to reflux for 48 h. The mixture is cooled in an ice-water bath before water (3.0 mL, 170 mmol) is added dropwise, followed by the sequential addition of NaOH (3.0 mL of a 15% (w/v) solution) and water (9.0 mL, 500 mmol). Excess $K_2CO_3$ is added, and the mixture is stirred vigorously for 15 min. The mixture is filtered, and the filtrate is concentrated *in vacuo* to afford Int 12 as a yellow powder (70% yield): $^1$H NMR (DMSO-$d_6$) δ 4.3, 4.1, 3.7, 3.5-3.2, 2.9-2.7, 2.5-2.3, 1.5, 1.3.

Step L. Preparation of benzyl *cis*-3-hydroxy-4-(hydroxymethyl)piperidine-1-carboxylate (Int 13).

*N*-(benzyloxy carbonyloxy)succinimide (3.04 g, 12.2 mmol) is added to a stirred solution of Int 12 (1.6 g, 12.2 mmol) in saturated aqueous $NaHCO_3$ (15 mL) at rt. The mixture is stirred at rt for 18 h. The organic and aqueous layers are separated. The aqueous layer is extracted with ether (3X). The combined organic layers are dried ($K_2CO_3$), filtered and concentrated *in vacuo* to afford Int 13 as a yellow oil (99% yield): $^1$H NMR ($CDCl_3$) δ 7.4-7.3, 5.2, 4.3, 4.1, 3.8-3.7, 3.0-2.8, 2.1, 1.9-1.7, 1.4..

Step M. Preparation of benzyl *cis*-3-hydroxy-4-[(4-methylphenyl)sulfonyl oxymethyl]piperidine-1-carboxylate (Int 14).

*Para*-toluenesulfonyl chloride (1.0 g, 5.3 mmol) is added to a stirred solution of Int 13 (3.6 g, 5.3 mmol) in pyridine (10 mL) in a -15°C bath. The mixture is stirred for 4 h, followed by addition of HCl (4.5 mL of a 6.0 M solution). $CH_2Cl_2$ (5 mL) is added. The organic and aqueous layers are separated. The aqueous layer is extracted with $CH_2Cl_2$. The combined organic layers are washed with brine, dried ($MgSO_4$), filtered and concentrated *in vacuo* to afford Int 14 as a colorless oil (78% yield): $^1$H NMR ($CDCl_3$) δ 7.8, 7.4-7.2, 5.1, 4.3-4.2, 4.1, 3.9-3.8, 2.9-2.7, 2.4, 1.9, 1.6-1.3.

Step N. Preparation of *exo*-1-azabicyclo[2.2.1]heptan-3-ol (Int 15).

A mixture of Int 14 (3.6 g, 8.6 mmol) and 10% Pd/C catalyst (500 mg) in EtOH (50 mL) is placed under an atmosphere of hydrogen. The mixture is shaken for

16 h. The mixture is filtered through Celite. Solid $NaHCO_3$ (1.1 g, 13 mmol) is added to the filtrate, and the mixture is heated in an oil bath at 50°C for 5 h. The solvent is removed *in vacuo*. The residue is dissolved in saturated aqueous $K_2CO_3$ solution. Continuous extraction of the aqueous layer using a liquid-liquid extraction apparatus (18 h), followed by drying the organic layer over anhydrous $K_2CO_3$ and removal of the solvent *in vacuo* affords Int 15 as a white solid (91% yield): $^1$H NMR δ 3.8, 3.0-2.8, 2.6-2.5, 2.4-2.3, 1.7, 1.1.

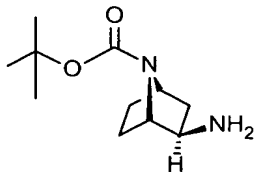
Step O. Preparation of *endo*-3-azido-1-azabicyclo[2.2.1]heptane (Int 16).

To a mixture of Int 15 (1.0 g, 8.9 mmol) and triphenyl phosphine (3.0 g, 11.5 mmol) in toluene-THF (50 mL, 3:2) in an ice-water bath are added sequentially a solution of hydrazoic acid in toluene (15 mL of ca. 2 M solution) and a solution of diethyl azadicarboxylate (1.8 mL, 11.5 mmol) in toluene (20 mL). The mixture is allowed to warm to rt and stir for 18 h. The mixture is extracted with aqueous 1.0M HCl solution. The aqueous layer is extracted with EtOAc, and the combined organic layers are discarded. The pH of the aqueous layer is adjusted to 9 with 50% aqueous NaOH solution. The aqueous layer is extracted with $CH_2Cl_2$ (3X), and the combined organic layers are washed with brine, dried ($Na_2SO_4$), filtered and concentrated *in vacuo*. The crude product is purified by flash chromatography on silica gel. Elution with $CHCl_3$-MeOH-$NH_4OH$ (92:7:1) affords Int 16 as a colorless oil (41% yield): $^1$H NMR ($CDCl_3$) δ 4.1, 3.2, 2.8, 2.7-2.5, 2.2, 1.9, 1.5.

Step P. Preparation of *endo*-3-amino-1-azabicyclo[2.2.1]heptane bis(hydro-*para*-toluenesulfonate).

A mixture of Int 16 (250 mg, 1.8 mmol) and 10% Pd/C catalyst (12 mg) in EtOH (10 mL) is placed under an atmosphere of hydrogen (15 psi). The mixture is stirred for 1 h at rt. The mixture is filtered through Celite, and the filtrate is concentrated *in vacuo*. The residue is dissolved in EtOH (10 mL) and *para*-toluenesulfonic acid monohydrate (690 mg, 3.7 mmol) is added. The mixture is stirred for 30 min, and the precipitate is filtered. The precipitate is washed sequentially with cold EtOH and ether. The precipitate is dried *in vacuo* to afford *endo*-[2.2.1]-Amine as a white solid (85% yield): $^1$H NMR ($CD_3OD$) δ 7.7, 7.3, 4.2, 3.9, 3.6-3.4, 3.3-3.2, 2.4, 2.3, 2.1.

**Preparation of *exo-tert*-butyl (1*S*, 2*R*, 4*R*)-(+)-2-amino-7-azabicyclo[2.2.1]heptane-7-carboxylate (7-aza-[2.2.1]-Amine):**




7-aza-[2.2.1]-Amine

Preparation of methyl-3-bromo-propiolate:

Methyl propiolate (52 ml, 0.583 mole) is combined with recrystallized *N*-bromo-succinimide (120 g, 0.674 mole) in 1,700 ml acetone under nitrogen. The solution is treated with silver nitrate (9.9 g, 0.0583 mole) neat in a single lot and the reaction is stirred 6 h at RT. The acetone is removed under reduced pressure (25°C, bath temperature) to provide a gray slurry. The slurry is washed with 2 x 200 ml hexane, the gray solid is removed by filtration, and the filtrate is concentrated *in vacuo* to provide 95 g of a pale yellow oily residue. The crude material was distilled via short path under reduced pressure (65°C, about 25 mm Hg) into a dry ice/acetone cooled receiver to give 83.7 g (88%) of methyl-3-bromo-propiolate as a pale yellow oil. Anal. calc'd for $C_4H_3BrO_2$: C, 29.48; H, 1.86. Found: C, 29.09; H, 1.97.

Preparation of 7-*tert*-butyl 2-methyl 3-bromo-7-azabicyclo[2.2.1]hepta-2,5-diene-2,7-dicarboxylate.

Methyl-3-bromo-propiolate (83.7 g, 0.513 mole) is added to *N*-*t*-butyloxy-pyrrole (430 ml, 2.57 mole) under nitrogen. The dark mixture is warmed in a 90 °C bath for 30 h, is cooled, and the bulk of the excess *N*-*t*-butyloxy-pyrrole is removed *in vacuo* using a dry ice/acetone condenser. The dark oily residue is chromatographed over 1 kg silica gel (230-400 mesh) eluting with 0-15% EtOAc/hexane. The appropriate fractions are combined and concentrated to afford 97 g (57%) of 7-*tert*-butyl 2-methyl 3-bromo-7-azabicyclo[2.2.1]hepta-2,5-diene-2,7-dicarboxylate as a dark yellow oil. HRMS (FAB) calc'd for $C_{13}H_{16}BrNO_4$+H: 330.0341, found 330.0335 $(M+H)^+$.

Preparation of (+/–) *Endo*-7-tert-butyl 2-methyl 7-azabicyclo[2.2.1]heptane-2,7-dicarboxylate.

7-*tert*-Butyl 2-methyl 3-bromo-7-azabicyclo[2.2.1]hepta-2,5-diene-2,7-dicarboxylate (97 g, 0.294 mole) is added to10% Pd/C (6.8g) in 900 ml absolute EtOH in a PARR bottle. The suspension is diluted with a solution of $NaHCO_3$ (25 g, 0.301 mole) in 250 ml water and the mixture is hydrogenated at 50 PSI for 2.5 h. The catalyst is removed by filtration, is washed with fresh EtOH, and the filtrate is concentrated *in vacuo* to give a residue. The residue is partitioned between 1 x 200 ml saturated $NaHCO_3$ and $CH_2Cl_2$ (4 x 100 ml). The combined organic layer is dried (1:1 $K_2CO_3$/ $MgSO_4$) and concentrated *in vacuo* to afford 72.8 g (98%) of (+/–) *endo*-7-*tert*-butyl 2-methyl 7-azabicyclo[2.2.1]heptane-2,7-dicarboxylate. MS (EI) for $C_{14}H_{22}O_4$, *m/z*: 255 $(M)^+$.

Preparation of (+/–) *exo*-7-(*tert*-butoxycarbonyl)-7-azabicyclo[2.2.1]heptane-2-carboxylic acid.

(+/–)*Endo*-7-*tert*-butyl 2-methyl 7-azabicyclo[2.2.1]heptane-2,7-dicarboxylate (72.8 g, 0.285 mole) is dissolved in 1000 ml dry MeOH in a dried flask under nitrogen. The solution is treated with solid NaOMe (38.5 g, 0.713 mole) neat, in a single lot and the reaction is warmed to reflux for 4h. The mixture is cooled to 0°C, is treated with 400 ml water, and the reaction is stirred 1h as it warms to RT. The mixture is concentrated *in vacuo* to about 400 ml and the pH of the aqueous residue is adjusted to 4.5 with 12N HCl. The precipitate is collected and dried. The tan, slightly tacky solid is washed with 2 x 100 ml 60% ether in hexane and is dried to provide 47 g (68%) of *exo*-7-(*tert*-butoxycarbonyl)-7-azabicyclo[2.2.1]heptane-2-carboxylic acid as an off-white powder. HRMS (FAB) calc'd for $C_{12}H_{19}NO_4$+H: 242.1392, found 242.1390 $(M+H)^+$.

Preparation of (+/–) *exo-tert*-butyl 2-{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate.

(+/–)*Exo*-7-(*tert*-butoxycarbonyl)-7-azabicyclo[2.2.1]heptane-2-carboxylic acid (32.5 g, 0.135 mole) is combined with TEA (24.4 ml, 0.175 mole) in 560 ml dry toluene in a dry flask under nitrogen. The solution is treated drop-wise with diphenylphosphoryl azide (37.7 ml, 0.175 mole), and is allowed to stir for 20 min at RT. The mixture is treated with benzyl alcohol (18.1 ml, 0.175 mole), and the

reaction is stirred overnight at 50°C. The mixture is cooled, is extracted successively with 2 x 250 ml 5% citric acid, 2 x 200 ml water, 2 x 200 ml saturated sodium bicarbonate, and 2 x 100 ml saturated NaCl. The organic layer is dried ($MgSO_4$) and concentrated *in vacuo* to an amber oil. The crude material was chromatographed over 800 g silica gel (230-400 mesh), eluting with 15-50% EtOAc/hexane. The appropriate fractions are combined and concentrated to give 44 g (94%) of (+/-) *exo-tert*-butyl 2-{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate as a pale oil. $^1$H NMR ($CDCl_3$) δ 1.29-1.60, 1.44, 1.62-2.01, 3.76-3.88, 4.10, 4.24, 5.10, 7.36 ppm.

Preparation of *exo-tert*-butyl (1*S*, 2*R*, 4*R*)-(+)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate and *exo-tert*-butyl (1*R*, 2*S*, 4*S*)-(–)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate.

The isolated (+/–) *exo-tert*-butyl 2-{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate is resolved via preparative chiral HPLC (50x500 mm Chiralcel OJ column, 30 deg. C, 70 mL/min. 10/90 (v/v) isopropanol/heptane). The resolution affords 10.5 g of *exo-tert*-butyl (1*S*, 2*R*, 4*R*)-(+)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate and 15.5 g of *exo-tert*-butyl-(1*R*, 2*S*, 4*S*)(-)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate.

The 2*R* enantiomer is triturated with 12 ml ether followed by 12 ml hexane (to remove lingering diastereo and enantiomeric impurities) and is dried to afford 9.5 g (43%) of purified *exo-tert*-butyl (1*S*, 2*R*, 4*R*)-(+)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate with 99% enantiomeric excess. MS (EI) for $C_{19}H_{26}N_2O_4$, *m/z*: 346 $(M)^+$. $[\alpha]^{25}_D$ = 22, (*c* 0.42, chloroform).

The 2*S* enantiomer is triturated with 20 ml ether followed by 20 ml hexane to give 14 g (64%) of purified *exo-tert*-butyl (1*R*, 2*S*, 4*S*)-(–)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate with 99% enantiomeric excess. MS (EI) for $C_{19}H_{26}N_2O_4$, *m/z*: 346 $(M)^+$. $[\alpha]^{25}_D$ = -23, (*c* 0.39, chloroform).

Preparation of *exo-tert*-butyl-(1*S*, 2*R*, 4*R*)-(+)-2-amino-7-azabicyclo[2.2.1]heptane-7-carboxylate (7-aza-[2.2.1]-Amine).

*Exo-tert*-butyl (1*S*, 2*R*, 4*R*)-(+)-2{[(benzyloxy)carbonyl]amino}-7-azabicyclo[2.2.1]heptane-7-carboxylate (9.5 g, 27.4 mmol) is combined with 950 mg

10% Pd/C in 75 ml absolute EtOH in a 500 ml Parr bottle. The reaction mixture is hydrogenated at 50 PSI for 3h, the catalyst is removed by filtration, and the filter cake was washed with MeOH. The filtrate is concentrated *in vacuo* to give 6.4 g of a residue. The crude material is chromatographed over 200 g silica gel (230-400 mesh) eluting with 7% $CH_3OH/CHCl_3$ containing 1% conc. $NH_4OH$. The appropriate fractions are combined and concentrated to give 5.61 g (96%) of *exo-tert*-butyl-(1*S*, 2*R*, 4*R*)-(+)-2-amino-7-azabicyclo[2.2.1]heptane-7-carboxylate as a pale oil. MS (EI) for $C_{11}H_{20}N_2O_2$, *m/z*: 212 (M)+. $[\alpha]^{25}D$ = 9, (*c* 0.67, chloroform).

**Preparation of 1-azabicyclo[3.2.1]octan-3-amine:**

**Preparation of the 3*R*,5*R*-[3.2.1]-Amine:**

**(3*S*)-1-[(*S*)-1-Phenethyl]-5-oxo-3-pyrrolidine-carboxylic acid:**

According to the literature procedure (Nielsen *et al.* J. Med. Chem **1990**, 70-77), a mixture of itaconic acid (123.17 g, 946.7 mmol) and (*S*)-(-)-α-methyl benzylamine (122.0 mL, 946.4 mmol) were heated (neat) in a 160°C oil bath for 4 h. Upon cooling, MeOH (~200 mL) was added and the resulting solid collected by filtration. The solid was treated with EtOH (~700 mL) and warmed using a steam bath until ~450 mL solvent remained. After cooling to rt, the solid was collected and dried to afford 83.2 g as a white crystalline solid: $[\alpha]^{25}_D$ = -80 (*c* 0.97, DMSO). MS (EI) *m/z* 233 ($M^+$).

The lack of a resonance 3.59 indicates a single diastereomer. The other diastereomer can be retrieved from the initial MeOH triturant. Attempts to crystallize this material generally led to small quantities of (3*RS*)-1-[(*S*)-1-phenethyl]-5-oxo-3-pyrrolidine-carboxylic acid.

**(3*S*)-1-[(*S*)-1-Phenethyl]-3-(hydroxymethyl)pyrrolidine:**

A suspension (3*S*)-1-[(*S*)-1-phenethyl]-5-oxo-3-pyrrolidine-carboxylic acid (82.30 g, 352.8 mmol) in $Et_2O$ (200 mL) was added in small portions to a slurry of $LiAlH_4$ (17.41 g, 458.6 mmol) in $Et_2O$ (700 mL). The mixture began to reflux during the addition. The addition funnel containing the suspension was rinsed with $Et_2O$ (2 x 50 mL), and the mixture was heated in a 50 °C oil bath for an additional 2 h and first allowed to cool to rt and then further cooled using an ice bath. The mixture was

carefully treated with $H_2O$ (62 mL). The resulting precipitate was filtered, rinsed with $Et_2O$, and discarded. The filtrate was concentrated to a yellow oil. When EtOAc was added to the oil, a solid began to form. Hexane was then added and removed by filtration and dried to afford 43.3 g as a white solid. $[\alpha]^{25}_D = -71$ (*c* 0.94, $CHCl_3$). MS (EI) *m/z* 205 ($M^+$).

**(3*R*)-1-[(*S*)-1-Phenethyl]-3-(cyanomethyl)pyrrolidine:**

A solution of (3*S*)-1-[(*S*)-1-phenethyl]-3-(hydroxymethyl)pyrrolidine (42.75 g, 208.23 mmol) in chloroform (350 mL) was heated to reflux under $N_2$. The solution was treated with a solution of thionyl chloride (41.8 mL, 573 mmol) in chloroform (40 mL) dropwise over 45 min. The mixture stirred for an additional 30 min, was cooled and concentrated. The residue was diluted with $H_2O$ (~200 mL), 1 N NaOH was added until a pH ~ 8 (pH paper). A small portion (~50 mL) of sat. $NaHCO_3$ was added and the basic mixture was extracted with EtOAc (3 x 400 mL), washed with brine, dried ($MgSO_4$), filtered and concentrated to give 46.51 g of a red-orange oil for (3*S*)-1-[(*S*)-1-phenethyl]-3-(chloromethyl)pyrrolidine: $R_f$: 0.50 (EtOAc-hexane 1:1); MS (ESI+) *m/z* 224.2 ($MH^+$). The chloride (46.35 g, 208.0 mmol) was transferred to a flask, dimethyl sulfoxide (200 mL) was added, and the solution was treated with NaCN (17.84 g, 363.9 mmol). The mixture was heated under $N_2$ in a 100°C oil bath overnight and was cooled. The brown mixture was poured into $H_2O$ (300 mL) and extracted with EtOAc (1000 mL in portions). The combined organic layer was washed with $H_2O$ (6 x ~50 mL), brine (~100 mL), dried ($MgSO_4$), filtered and concentrated to give 40.61 g as an orange-red oil: $R_f$: 0.40 (EtOAc-$PhCH_3$ 1:1). MS (ESI+) for *m/z* 215.2 ($M+H^+$).

**(3*R*)-Methyl 1-[(*S*)-1-phenylethly]pyrrolidine-3-acetate:**

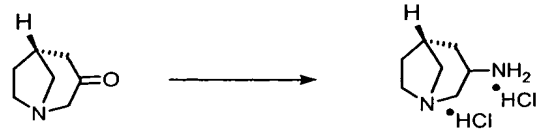
Acetyl chloride (270 mL, 3.8 mol) was carefully added to a flask containing chilled (0°C) methanol (1100 mL). After the addition was complete, the acidic solution stirred for 45 min (0 °C) and then (3*R*)-1-[(*S*)-1-phenethyl]-3-(cyanomethyl)pyrrolidine (40.50 g, 189.0 mmol) in methanol (200 mL) was added. The ice bath was removed and the mixture stirred for 100 h at rt. The resulting suspension was concentrated. Water (~600 mL) was added, the mixture stirred for 45 min and then the pH was adjusted (made basic) through the addition of ~700 mL sat.

aq. $NaHCO_3$. The mixture was extracted with EtOAc (3 x 300 mL). The combined organics were washed with brine, dried ($MgSO_4$), filtered through celite and concentrated to give 36.86 g as an orange-red oil. MS (ESI+) *m/z* 248.2 ($M+H^+$).

**(5*R*)-1-Azabicyclo[3.2.1]octan-3-one hydrochloride:**

A solution of (3*R*)-methyl 1-[(*S*)-1-phenylethly]pyrrolidine-3-acetate (25.72g, 104.0 mmol) in THF (265 mL) was cooled under $N_2$ in a $CO_2$/acetone bath. Next, $ICH_2Cl$ (22.7 mL, 312.0 mmol) was added, and the mixture stirred for 30 min. A solution of 2.0M lithium diisopropylamide (heptane/THF/ethylbenzene, 156 mL, 312 mmol) was added slowly over 30 min. The internal temperature reached a maximum of –40°C during this addition. After 1 h, sat. $NH_4Cl$ (100 mL) was added and the mixture was allowed to warm to rt. The organic layer was separated, dried ($MgSO_4$), filtered and concentrated. The resulting red-brown foam was chromatographed (300 g $SiO_2$, $CHCl_3$-MeOH-$NH_4OH$ (89:10:1) followed by $CHCl_3$-MeOH (3:1). The product fractions were pooled and concentrated to afford (5*R*)-3-oxo-1-[(1*S*)-1-phenylethyl]-1-azoniabicyclo[3.2.1]octane chloride (10.12g) as a tan foam (MS (ESI+) *m/z* 230.1 ($M+H^+$). This foam (10.1 g, 38 mmol) was taken up in MeOH (500 mL), 10% Pd(C) (3.0 g) added and the mixture was hydrogenated (45 psi) overnight. The mixture was filtered and re-subjected to the reduction conditions (9.1 g, 10% Pd/C, 50 psi). After 5 h, TLC indicated the consumption of the (5*R*)-3-oxo-1-[(1*S*)-1-phenylethyl]-1-azoniabicyclo[3.2.1]octane chloride. The mixture was filtered, concentrated and triturated (minimal *i*PrOH) to give 3.73 g in two crops, as an off-white solid: $[\alpha]^{25}_D$ = 33 (*c* 0.97, DMSO). MS (EI) *m/z* 125 ($M^+$).

**(3*R*,5*R*)-1-azabicyclo[3.2.1]octan-3-amine dihydrochloride:**

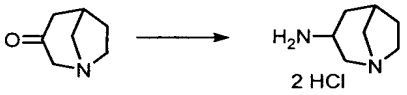



(3*R*, 5*R*)-[3.2.1]-Amine

To a flask containing (5*R*)-1-azabicyclo[3.2.1]octan-3-one hydrochloride (3.64 g, 22.6 mmol), hydroxylamine hydrochloride (2.04 g, 29.4 mmol), and ethanol (130 mL) was added sodium acetate trihydrate (9.23 g, 67.8 mmol). The mixture stirred for 3 h and was filtered and concentrated. The resulting white solid was taken up in *n*-

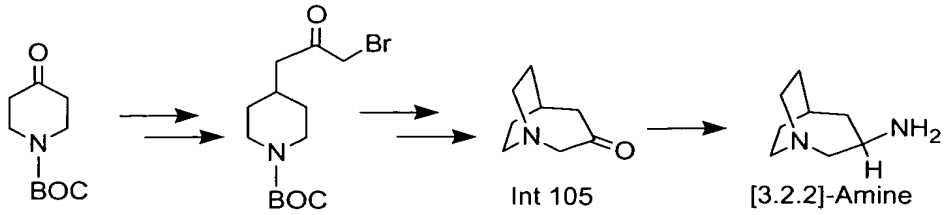
propanol (100 mL) and sodium (~13.6 g, 618 mmol) was added over 20-25 portions. The reaction spontaneously began to reflux, and the reaction was heated in an oil bath (100°C). The addition was complete in ~20 min and the mixture had solidified after ~40 min. The oil bath was removed and *n*-propanol (2 x 25 mL) was added dissolving the remaining sodium metal. The mixture was carefully quenched through the dropwise addition of $H_2O$ (100 mL). Saturated aq. NaCl (20 mL) was added, and the layers were separated. The organic layer was dried ($MgSO_4$), filtered, treated with freshly prepared MeOH/HCl, and concentrated. The resulting solid was triturated with 30 mL EtOH, filtered and dried *in vaccuo* to afford 3.51 g as a white solid: $[\alpha]^{25}_D$ = -3 (*c* 0.94, DMSO). MS (FAB) *m/z* 127 ($MH^+$).

**Preparation of *endo*-1-azabicyclo[3.2.1]octan-3-amine dihydrochloride (*endo*-[3.2.1]-Amine):**




A mixture of 1-azabicyclo[3.2.1]octan-3-one hydrochloride (2.80 g, 17.3 mmol), ethanol (25 mL), and hydroxylamine hydrochloride (1.56 g, 22.4 mmol) is treated with sodium acetate trihydrate (7.07 g, 51.2 mmol). The mixture is stirred for 3 h and evaporated *in vacuo*. The residue is diluted with $CH_2Cl_2$, treated with charcoal, filtered and evaporated. The resulting oxime (3.1 mmol) is treated with acetic acid (30 mL) and hydrogenated at 50 psi over $PtO_2$ (50 mg) for 12 h. The mixture is then filtered and evaporated. The residue is taken up in a minimal amount of water (6 mL) and the pH is adjusted to >12 using solid NaOH. The mixture is then extracted with ethyl acetate (4 X 25 mL), dried ($MgSO_4$), filtered, treated with ethereal HCl, and evaporated to give the give *endo*-[3.2.1]-Amine.

**Preparation of the 3.2.2 Amines:**



*tert*-Butyl 4-(2-oxopropylidene)piperidine-1-carboxylate (Int 101):

Sodium hydride (60% oil dispersion, 2.01 g, 50.2 mmol) is washed with pentane (3X) and suspended in dry THF (40 mL). The solution is cooled to 0°C before diethyl (2-oxopropyl)phosphonate (9.75 g, 50.2 mmol) is added dropwise. After complete addition, the solution is warmed to rt and stirred for 30 min. *tert*-Butyl 4-oxo-1-piperidinecarboxylate (5.0g, 25.1 mmol) is added in portions over 10 min, followed by stirring at rt for 2 h. A saturated aqueous solution of ammonium chloride is added, followed by dilution with ether. The organic layer is extracted with water. The organic layer is dried ($MgSO_4$), filtered and concentrated to a yellow oil. The crude product is purified by flash chromatography on silica gel. Elution with hexanes-ether (60:40) gave 4.5 g (75%)of Int 101 as a white solid: $^1$H NMR ($CDCl_3$) δ 6.2, 3.5, 3.4, 2.9, 2.3, 2.2, 1.5.

Preparation of *tert*-butyl 4-(2-oxopropyl)piperidine-1-carboxylate (Int 102):

A mixture of Int 101 (4.5 g, 19 mmol) and 10% palladium on activated carbon (450mg) in EtOH (150 mL) is placed in a Parr bottle and hydrogenated for 5 h at 50 psi. The mixture is filtered through Celite, and the filtrate is concentrated *in vacuo* to afford 4.3 g (94%) of Int 102 as a clear oil: $^1$H NMR ($CDCl_3$) δ 4.1, 2.8, 2.4, 2.2, 2.0, 1.7, 1.5, 1.1.

*tert*-Butyl 4-(3-bromo-2-oxopropyl)piperidine-1-carboxylate (Int 103):

To a stirred solution lithium hexamethyldisilylamide in THF (20. 0 mL, 1.0 M) in a –78 °C bath is added chlorotrimethylsilane (11.0 mL, 86.4 mmol) dropwise. The mixture is stirred at –78 °C for 20 min, followed by addition of Int 102 (3.21 g, 13.3 mmol) in a solution of THF (50 mL) dropwise. After complete addition, the mixture is stirred at –78 °C for 30 min. The mixture is warmed to 0°C in an ice-water bath and phenyltrimethylammonium tribromide (5.25 g, 14.0 mmol) is added. The mixture is stirred in an ice-bath for 30 min, followed by the addition of water and ether. The aqueous layer is washed with ether, and the combined organic layers are washed with saturated aqueous sodium thiosulfate solution. The organic layer is dried ($MgSO_4$), filtered and concentrated *in vacuo* to afford a yellow oil. The crude product is purified by flash chromatography on silica gel. Elution with hexanes-ether (60:40) gave 2.2 g

(52%) of Int 103 as a lt. yellow oil: $^1$H NMR ($CDCl_3$) δ 4.2-4.1, 3.9, 2.8, 2.7, 2.6, 2.1-2.0, 1.7, 1.5, 1.2-1.1.2.

1-Bromo-3-piperidin-4-ylacetone trifluoroacetate (Int 104):

To a stirred solution of Int 103 (2.2 g, 6.9 mmol) in $CH_2Cl_2$ (30 mL) in an ice-water bath is added trifluoroacetic acid (10 mL, 130 mmol). The mixture is stirred at 0°C for 30 min. The volatiles are removed *in vacuo* to afford 2.0 g (87%) of Int 104 as a yellow residue: MS (ESI) for $C_8H_{15}BrNO$ [M+H] *m/e* 220.

1-Azabicyclo[3.2.2]nonan-3-one (Int 105):

To a stirred solution of DIEA (13 mL) in acetoniltrile (680 mL) at reflux temperature is added a solution of Int 104 (2.0 g, 6.0 mmol) in acetonitrile (125 mL) over a 4 h period via syringe pump. The mixture is kept at reflux temperature overnight. The mixture is concentrated *in vacuo* and the remaining residue is partitioned between a saturated aqueous potassium carbonate solution and $CHCl_3$-MeOH (90:10). The aqueous layer is extracted with $CHCl_3$-MeOH (90:10), and the combined organic layers are dried ($MgSO_4$), filtered and concentrated *in vacuo* to a brown oil. The crude product is purified by flash chromatography on silica gel. Elution with $CHCl_3$-MeOH-$NH_4OH$ (95:4.5:0.5) gives 600 mg (72%) of Int 105 as a clear solid: $^1$H NMR ($CDCl_3$) δ 3.7, 3.3-3.2, 3.1-3.0, 2.7, 2.3, 2.0-1.8.

1-Azabicyclo[3.2.2]nonan-3-amine bis(4-methylbenzenesulfonate) ([3.2.2]-Amine):

To a stirred mixture of Int 105 (330 mg, 2.4 mmol) and sodium acetate•trihydrate (670 mg, 4.8 mmol) in EtOH (6.0 mL) is added hydroxylamine•hydrochloride (200 mg, 2.8 mmol). The mixture is stirred at rt for 10 h. The mixture is filtered and the filtrate is concentrated *in vacuo* to a yellow solid. To a solution of the solid (350 mg, 2.3 mmol) in *n*-propanol (30 mL) at reflux temperature is added sodium metal (2.0 g, 87 mmol) in small portions over 30 min. Heating at reflux is continued for 2 h. The solution is cooled to rt and brine is added. The mixture is extracted with *n*-propanol, and the combined organic layers are concentrated *in vacuo*. The residue is taken up in $CHCl_3$ and the remaining solids are filtered. The filtrate is dried ($MgSO_4$), filtered and concentrated *in vacuo* to a clear

solid. To a stirred solution of the solid (320 mg, 2.3 mmol) in EtOH (4 mL) is added *p*-toluenesulfonic acid monohydrate (875 mg, 4.6 mmol). The solution is warmed in a water bath to 45°C for 30 min, followed by concentration of the solvent to afford 710 mg (62%) of [3.2.2]-Amine as a white solid: $^1$H NMR ($CD_3OD$) δ 7.7, 7.3, 4.1-3.9, 3.6-3.4, 2.6-2.5, 2.4, 2.2-2.1, 2.1-2.0, 1.9.

Resolution of stereoisomers:

The amine can be coupled to form the appropriate amides or thioamides as a racemic mixture. The racemic mixture can then be resolved by chromatography using chiral columns or chiral HPLC, techniques widely known in the art, to provide the requisite resolved enantiomers 3(*R*) and 3(*S*) of said amides.

Coupling procedures using the Azabicyclo moieties discussed herein with various W moieties discussed herein to prepare compounds of formula I are discussed in the following, all of which are incorporated herein by reference: US 6,492,386; US 6,500,840; US 6,562,816; US 2003/0045540A1; US 2003/0055043A1; US 2003/0069296A1; US 2003/0073707A1; US 2003/015089A1; US 2003/0130305A1; US 2003/0153595A1; WO 03/037896; WO 03/40147; WO 03/070728; WO 03/070731; WO 03/070732. Although the compounds made therein may be for one specific Azabicyclo moiety, the procedures discussed, or slight non-critical changes thereof, can be used to make the compounds of formula I.

The intermediates providing the W of formula I either are commercially available or prepared using known procedures, making non-critical changes.

Compounds of Formula I where W is (D) are made using the coupling procedures discussed herein and in the literature, making non-critical changes to obtain the desired compounds. The following intermediates to provide W as (D) of formula I are for exemplification only and are not intended to limit the scope of the present invention. Other intermediates within the scope of the present invention can be obtained using known procedures or by making slight modifications to known procedures.

**Intermediate D1: furo[2,3-c]pyridine-5-carboxylic acid**

There are many routes for obtaining the carboxylic acid including the preparation of the acid discussed herein and also from hydrolyzing the ester, the preparation of which is discussed in US 6,265,580. n-Butyl furo[2,3-c]pyridine-5-carboxylate is hydrolyzed to the corresponding carboxylate salt on treatment with sodium or potassium hydroxide in aqueous methanol or acetonitrile-methanol mixtures. Acidification to pH 2.5-3.5 generates the carboxylic acid, which is isolated as a solid. The free base can also be prepared directly from n-butyl furo[2,3-c]pyridine-5-carboxylate by direct condensation using at least 1.5 molar equivalents of (R)-3-aminoquinuclidine and heating in ethanol or n-butyl alcohol.

2-Chloro-3-pyridinol (20.0 g, 0.154 mole), $NaHCO_3$ (19.5g, 0.232 mole, 1.5 equ), and 150 mL of water are placed in a flask. The flask is placed in an oil bath at $90^0$C, and after 5 min, 37% aqueous formaldehyde (40.5 mL, 0.541 mole, 3.5 equ) is added in six unequal doses in the following order: 12 mL, 3 x 8 mL, then 2.2 mL all at 90-min intervals and then the final 2.3 mL after the reaction stirs for 15 h at 90°C. The reaction is stirred at 90°C for another 4 h and then cooled by placing the flask in an ice bath. The pH of the reaction is then adjusted to 1 using 6N HCl. The reaction is stirred for 1.5 h in an ice bath allowing an undesired solid to form. The undesired solid is removed by filtration, and the filtrate is extracted seven times with EtOAc. The combined organic extracts are concentrated *in vacuo*, toluene is added to the flask and removed *in vacuo* to azeotrope water, and then $CH_2Cl_2$ is added and removed *in vacuo* to obtain 2-chloro-6-(hydroxymethyl)-3-pyridinol (I-1-D) as a pale yellow solid (81% yield) sufficiently pure for subsequent reaction. MS (EI) for $C_6H_6ClNO_2$, *m/z*: 159 $(M)^+$.

I-1-D (11.6 g, 72.7 mmol) and $NaHCO_3$ (18.3 g, 218 mmol) are added to 200 mL $H_2O$. The mixture is stirred until homogeneous, the flask is placed in an ice bath, iodine (19.4 g, 76.3 mmol) is added, and the reaction is stirred over the weekend at rt. The pH of the mixture is adjusted to 3 with 2N $NaHSO_4$, and the mixture is extracted with 4 x 50 mL EtOAc. The combined organic layer is dried ($MgSO_4$), is filtered, and the filtrate is concentrated *in vacuo* to a yellow solid. The crude solid is washed with EtOAc to provide 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-2-D) as an off-white solid (62% yield), and the filtrate is concentrated to a small volume and is chromatographed over 250 g silica gel (230-400 mesh) eluting with 2.5:4.5:4:0.1

EtOAc/$CH_2Cl_2$/hexane/acetic acid to afford additional pure I-2-D (12% yield). MS (EI) for $C_6H_5ClINO_2$, *m/z*: 285$(M)^+$.

I-2-D (13.9 g, 48.6 mmol) is combined with trimethylsilylacetylene (9.6 mL, 68 mmol), bis(triphenylphosphine) palladium dichloride (1.02 g, 1.46 mmol) and cuprous iodide (139 mg, 0.73 mmol) in 80 mL $CHCl_3$/40 mL THF under $N_2$. TEA (21 mL, 151 mmol) is added, and the reaction is stirred 3 h at rt and is diluted with 200 mL $CHCl_3$. The mixture is washed with 2 x 150 mL 5% HCl and the combined aqueous layers are extracted with 2 x 50 mL $CHCl_3$. The combined organic layer is washed with 100 mL 50% saturated NaCl, is dried ($MgSO_4$), and concentrated *in vacuo* to an amber oil. The crude material is chromatographed over 350 g silica gel (230-400 mesh), eluting with 35% EtOAc/hexane to afford 2-chloro-6-(hydroxymethyl)-4-[(trimethylsilyl)ethynyl]-3-pyridinol (I-3-D) as a golden solid (92% yield). MS (EI) for $C_{11}H_{14}ClNO_2Si$, *m/z*: 255$(M)^+$.

I-3-D (7.9 g, 31.2 mmol) and cuprous iodide (297 mg, 1.6 mmol) in 60 mL EtOH/60 mL TEA are added to a flask. The reaction is placed in an oil bath at 70°C for 3.5h, is cooled to rt, and concentrated *in vacuo*. The residue is partitioned between 100 mL 5% HCl and $CH_2Cl_2$ (4 x 50 mL). The combined organic layer is dried ($MgSO_4$), filtered, and concentrated *in vacuo* to give 6.5 g of a crude amber solid. The crude material is chromatographed over 300 g silica gel (230-400 mesh) eluting with 30-40% EtOAc/hexane. Two sets of fractions with two different desired compounds are identified by TLC/UV. The two compounds eluted separately. The early-eluting pool of fractions is combined and concentrated to afford [7-chloro-2-(trimethylsilyl)furo[2,3-c]pyridin-5-yl]methanol (I-5-D) as a white solid (46% yield). The later-eluting pool of fractions is combined and concentrated to provide (7-chlorofuro[2,3-c]pyridin-5-yl)methanol (I-4-D) as a white solid (27% yield). MS (EI) for $C_8H_6ClNO_2$, *m/z*: 183 $(M)^+$ for I-4-D. HRMS (FAB) calculated for $C_{11}H_{14}ClNO_2Si$ *m/z*: 255.0482, found 255.0481 for I-5-D.

I-5-D (1.05 g, 4.1 mmol) and 10% Pd/C catalyst (1.05 g) are placed in 20 mL absolute EtOH. Cyclohexene (4 mL, 40.1 mmol) is added, and the reaction is refluxed for 2.5h, and then filtered through celite. The filter cake is washed with 1:1 EtOH/$CH_2Cl_2$, and the filtrate is concentrated to a pale yellow solid. The residue is partitioned between 40 mL saturated $NaHCO_3$ and extracted with $CH_2Cl_2$ (4 x 20 mL). The combined organic layer is dried ($MgSO_4$), filtered, and then concentrated *in*

*vacuo* to a pale oil (1.04 g). The pale oil is chromatographed over 50 g silica gel (230-400 mesh) eluting with 50-70% EtOAc/hexane to afford 5-hydroxymethyl-2-trimethylsilyl-furo[2,3-c]pyridine (I-14-D) as a white solid (90% yield). MS (EI) for $C_{11}H_{15}NO_2Si$, *m/z*: 221$(M)^+$.

I-14-D (770 mg, 3.48 mmol) is dissolved in 10 mL MeOH. 2N NaOH (3 mL, 6 mmol) is added, and the reaction is stirred for 1.5 h at rt. The solution is concentrated *in vacuo* to a residue. Water (20 mL) is added to the residue and extracted with 4 x 10 mL $CH_2Cl_2$. The combined organic layer is dried ($K_2CO_3$), filtered, and concentrated *in vacuo* to afford furo[2,3-c]pyridin-5-yl methanol (I-16-D) as a white solid (90% yield). Analysis calculated for $C_8H_7NO_2$: C, 64.42; H, 4.73; N, 9.39. Found: C, 64.60; H, 4.56; N, 9.44.

Alternatively, I-3-D is used to obtain I-16-D with fewer steps: I-3-D (44.6 g, 174.4 mmol) is combined with cuprous iodide (1.66 g, 8.72 mmol) and diisopropylamine (44 ml, 300 mmol) in 300 ml methanol under nitrogen. The reaction is warmed to 45-50°C for 6 h, is cooled to rt and treated with 100 ml saturated $NaHCO_3$ followed by 100 ml 2N NaOH. The dark mixture is stirred overnight, filtered through celite, the volatiles removed *in vacuo* and the residue is partitioned between 1 x 500 ml water and 4 x 200 ml $CH_2Cl_2$ (some filtrations is required to effect good separation). The combined organic layer is dried ($MgSO_4$) and concentrated *in vacuo* to afford I-4-D (25.25g, 79%) as a pale orange solid. Anal. Calcd for $C_8H_6ClNO_2$: C,52.34; H,3.29; N,7.63. Found: C,52.27; H,3.23; N,7.57.

I-4-D (32.0 g, 174 mmol) is combined with zinc powder (34.2 g, 523 mmol) in absolute EtOH (900 mL), using an overhead stirrer. The mixture is heated to 70°C, HCl (87.2 mL, 1.05 mol) is added slowly drop-wise, and the mixture is heated to reflux for 1 h. The mixture is cooled slightly, filtered to remove the metallic zinc and concentrated to near-dryness. The yellow oil is diluted with $H_2O$ (150 mL) and EtOAc (950 mL) and is treated slowly drop-wise with 20% $Na_2CO_3$ (310 mL) as the mixture is warmed to reflux. The vigorously stirred (using overhead stirrer) mixture is refluxed for 1 h, cooled slightly and the organics removed via cannula under reduced pressure. Additional EtOAc (600 mL) is added, the mixture is heated to reflux for 1 h, cooled slightly and the organics removed as above. More EtOAc (600 mL) is added, the mixture is stirred at rt overnight then heated to reflux for 1 h, cooled slightly and most of the organics removed as above. The remaining mixture is filtered

through celite, rinsed with EtOAc until no additional product elutes, and the layers separated. The aqueous layer is further extracted with EtOAc (2 X 400 mL). The combined organics are dried ($MgSO_4$) and concentrated to a dark yellow solid (23.6 g). The crude material is chromatographed over 900 g slurry-packed silica gel, eluting with 60% EtOAc / hexane (3 L), 70% EtOAc / hexane (2 L), and finally 100% EtOAc. The appropriate fractions are combined and concentrated *in vacuo* to afford I-16-D (19.5 g, 75%) as a white solid. Anal. Calcd for $C_8H_7NO_2$: C,64.42; H,4.73; N,9.39; Found: C,64.60; H,4.56; N,9.44.

Oxalyl chloride (685μL, 7.8 mmol) is dissolved in 30 mL $CH_2Cl_2$ in a dry flask under $N_2$. The flask is placed in a dry-ice/acetone bath, DMSO (1.11 mL, 15.6 mmol) in 5 mL $CH_2Cl_2$ is added drop-wise, and the mixture is stirred for 20 min. I-16-D (1.0 g, 6.7 mmol) in 10 mL $CH_2Cl_2$ is added, and the reaction is stirred 30 min at -78°C. TEA (4.7 mL, 33.5 mmol) is added, the reaction is allowed to warm to rt, is stirred 1h, and washed with 25 mL saturated $NaHCO_3$. The organic layer is dried ($K_2CO_3$), filtered, and concentrated *in vacuo* to an orange solid. The crude material is chromatographed over 50 g silica gel (230-400 mesh) eluting with 33% EtOAc/ hexane to provide furo[2,3-c]pyridine-5-carbaldehyde (I-17-D) as a white solid (86% yield). MS (EI) for $C_8H_5NO_2$, *m/z*: 147 $(M)^+$.

I-17-D (850 mg, 5.8 mmol) is dissolved in 10 mL DMSO. $KH_2PO_4$ (221 mg, 1.6 mmol) in 3 mL $H_2O$ is added and then $NaClO_2$ (920 mg, 8.2 mmol) in 7 mL $H_2O$ is added, and the reaction is stirred 3 h at rt. The reaction is diluted with 25 mL water, the pH is adjusted to 10 with 2N NaOH, and the mixture is extracted with 3 x 20 mL ether. The combined ether layer is discarded. The pH of the aqueous layer is adjusted to 3.5 with 10% aqueous HCl and is extracted with 13 x 10 mL 10% $MeOH/CH_2Cl_2$. The $MeOH/CH_2Cl_2$ organic layer is dried ($Na_2SO_4$), filtered, and concentrated *in vacuo* to a pale oil. The residual DMSO is removed under a stream of $N_2$ at rt to provide a white paste. The paste is dissolved in MeOH and concentrated to dryness. The white solid is washed with ether and dried to afford crude furo[2,3-c]pyridine-5-carboxylic acid (I-18-D) (94% yield). MS (ESI) for $C_8H_5NO_3$, 162.8 $(M-H)^-$.

**Intermediate D2: Furo[3,2-c]pyridine-6-carboxylic acid**

3-Bromofuran (8.99 mL, 100.0 mmol) is dissolved in DMF (8.5 mL), cooled to 0°C, treated dropwise with $POCl_3$ (9.79 mL, 105.0 mmol), stirred for 1 h at RT and

then heated to 80°C for 2 h. The mixture is cooled to RT, poured over ice (1 kg) and neutralized to pH 9 with solid $K_2CO_3$. The mixture is stirred for 1 h, extracted with $Et_2O$ (3 X 500 mL), dried ($K_2CO_3$) and concentrated to a dark brown oil. The crude material is chromatographed over 600 g slurry-packed silica gel, eluting with 6% EtOAc/hexane (4L), 8% EtOAc/hexane (2L), 10% EtOAc/hexane (1L), and finally 20% EtOAc/hexane. The appropriate fractions are combined and concentrated *in vacuo* to afford 14.22 g (81%) of 3-bromo-2-furaldehyde as a yellow oil. MS (EI) *m/z*: 174 ($M^+$).

3-Bromo-2-furaldehyde (14.22 g, 81.3 mmol) is combined with ethylene glycol (6.55 mL, 117.4 mmol) and *para*-toluene sulfonic acid monohydrate (772 mg, 4.06 mmol) in benzene (200 mL) and heated to reflux with a Dean-Stark trap for 5 h. Additional ethylene glycol (1.64 mL, 29.41 mmol) and benzene (150 mL) are added and the solution is heated for an additional 2 h. The mixture is cooled to RT, treated with saturated $NaHCO_3$ and stirred for 0.5 h. The layers are separated and the organics are dried ($Na_2SO_4$) and concentrated to a brown oil (18.8 g). The crude material is chromatographed over 700 g slurry-packed silica gel, eluting with 15% EtOAc/hexane. The appropriate fractions are combined and concentrated *in vacuo* to afford 16.45 g (92%) of 2-(3-bromo-2-furyl)-1,3-dioxolane as a yellow-orange oil. MS (EI) *m/z*: 218 ($M^+$).

2-(3-Bromo-2-furyl)-1,3-dioxolane (438 mg, 2.0 mmol) is dissolved in $Et_2O$ (5 mL) in a dry flask under nitrogen, cooled to -78°C, treated dropwise with *tert*-butyllithium (2.59 mL, 4.4 mmol) and stirred for 1 h. DMF (178 µL, 2.3 mmol) in $Et_2O$ (2 mL) is added dropwise, the mixture stirred for 4 h at -78°C, then treated with oxalic acid dihydrate (504 mg, 4.0 mmol) followed by water (2 mL). The cooling bath is removed and the mixture allowed to warm to RT over 1 h. The mixture is diluted with water (20 mL) and EtOAc (20 mL), the layers are separated and the aqueous layer extracted with EtOAc (1 X 20 mL). The organics are dried ($Na_2SO_4$) and concentrated to a yellow oil. The crude material is chromatographed over 12 g slurry-packed silica gel, eluting with 15% EtOAc/hexane. The appropriate fractions are combined and concentrated *in vacuo* to afford 228 mg (68%) of 2-(1,3-dioxolan-2-yl)-3-furaldehyde as a pale yellow oil. MS (EI) *m/z*: 168 ($M^+$).

2-(1,3-Dioxolan-2-yl)-3-furaldehyde (2.91 g, 17.31 mmol) is combined with formic acid (17 mL, 451 mmol) and water (4.25 mL) and stirred at RT for 18 h. The

mixture is slowly transferred into a solution of $NaHCO_3$ (45 g, 541 mmol) in water (600 mL), then strirred for 0.5 h. EtOAc (200 mL) is added, the layers separated and the aqueous layer extracted with EtOAc (2 X 200 mL). The combined organics are dried ($Na_2SO_4$) and concentrated to a yellow oil (3.28 g). The crude material is chromatographed over 90 g slurry-packed silica gel, eluting with 20% EtOAc/hexane. The appropriate fractions are combined and concentrated to afford 2.45 g of furan-2,3-dicarbaldehyde slightly contaminated with ethylene glycol diformate as a yellow oil. $^1H$ NMR ($CDCl_3$): δ 7.00 (d, $J$ = 2 Hz, 1 H), 7.67 (d, $J$ = 2 Hz, 1 H), 10.07 (s, 1 H), 10.49 (s, 1 H) ppm.

Methyl (acetylamino)(dimethoxyphosphoryl)acetate (2.34 g, 9.8 mmol) is dissolved in $CHCl_3$ (40 mL), treated with DBU (1.46 mL, 9.8 mmol), stirred for 5 min then added dropwise to a 0°C solution of furan-2,3-dicarbaldehyde (1.65 g, 8.9 mmol) in $CHCl_3$ (80 mL). The mixture is stirred for 2.5 h as the cooling bath expires then 5.5 h at RT and finally 24 h at 50°C. The mixture is concentrated *in vacuo* to a yellow oily-solid (6.66 g). The crude material is chromatographed over a standard 100g slurry-packed silica gel, eluting with 65% EtOAc/hexane. The appropriate fractions are combined and concentrated *in vacuo* to afford 1.30 g (82%) of methyl furo[3,2-c]pyridine-6-carboxylate as a yellow solid. MS (EI) *m/z*: 177 ($M^+$).

Methyl furo[3,2-c]pyridine-6-carboxylate (1.55 g, 8.74 mmol) is dissolved in MeOH (30 mL) and $H_2O$ (15 mL), treated with 3 N NaOH (6.4 mL) and stirred at RT for 7 h. The mixture is concentrated to dryness, dissolved in $H_2O$ (10 mL) and acidified to pH 2 with concentrated HCl. The solution is concentrated to dryness, suspended in a smaller amount of water (7 mL) and the resulting solid collected via filtration (lot A). The filtrate is concentrated, triturated with water (3 mL) and the resulting solid collected via filtration (lot B). The filtrate from lot B is concentrated and carried on without further purification as an acid/salt mixture (lot C). Both lots A and B are dried in a vacuum oven at 50°C for 18 h to afford 690 mg (48%) for lot A and 591 mg (42%) for lot B of furo[3,2-c]pyridine-6-carboxylic acid as yellow solids. MS (CI) *m/z* : 164 ($M + H^+$).

**Intermediate D3: 7-Chlorofuro[2,3-c]pyridine-5-carboxylic acid**

Oxalyl chloride (3.1 mL, 35 mmol) is dissolved in 200 mL $CH_2Cl_2$ in a dried flask under $N_2$. The flask is placed in a dry-ice/acetone bath at -78°C, DMSO (4.95

mL, 70 mmol) in 10 mL $CH_2Cl_2$ is added drop-wise, and the mixture is stirred for 20 min. (7-Chlorofuro[2,3-c]pyridin-5-yl)methanol (I-4-D) (5.5 g, 30 mmol) in 10 mL $CH_2Cl_2$ is added, and the reaction is stirred 30 min at -78°C. TEA (21.3 mL, 153 mmol) is then added. The reaction is stirred 30 min in the dry-ice/acetone bath, an ice bath replaces the dry-ice/acetone bath, and the reaction is stirred 1 h and is washed with 100 mL 1:1 saturated NaCl/$NaHCO_3$. The organic layer is dried ($K_2CO_3$), filtered, and concentrated *in vacuo* to afford 7-chlorofuro[2,3-c]pyridine-5-carbaldehyde (I-6-D) as a pale yellow solid (97% yield). MS (EI) for $C_8H_4ClNO_2$ *m/z*: 181 $(M)^+$.

I-6-D (3.0 g, 16.5 mmol) is dissolved in 40 mL DMSO. $KH_2PO_4$ (561 mg, 4.1 mmol) in 6.5 mL $H_2O$ is added and then $NaClO_2$ (2.6 g, 23.1 mmol) in 24 mL $H_2O$ is added, and the reaction is stirred overnight at rt. The reaction is diluted with 200 mL $H_2O$, the pH is adjusted to 9 with 2N NaOH, and any remaining aldehyde is extracted into 3 x 50 mL ether. The pH of the aqueous layer is adjusted to 3 with 10% aqueous HCl and is extracted with 4 x 50 mL EtOAc. The combined organic layer is dried ($MgSO_4$), filtered, and concentrated *in vacuo* to a white solid. The solid is washed with ether and dried to afford 7-chlorofuro[2,3-c]pyridine-5-carboxylic acid (I-7-D) (55% yield). MS (CI) for $C_8H_4ClNO_3$, *m/z*: 198 (M+H).

**Intermediate D4: 2,3-Dihydrofuro[2,3-c]pyridine-5-carboxylic acid**

I-7-D (980 mg, 4.98 mmol) is dissolved in 75 mL MeOH containing 500 mg 20% palladium hydroxide on carbon in a 250 mL Parr shaker bottle. The reaction mixture is hydrogenated at 20 PSI for 24 h. The catalyst is removed by filtration and the filtrate is concentrated *in vacuo* to a white solid. The solid is dissolved in MeOH and is loaded onto 20 mL Dowex 50W-X2 ion exchange resin (hydrogen form) which had been prewashed with MeOH. The column is eluted with 50 mL MeOH followed by 150 mL 5% TEA in MeOH to afford 2,3-dihydrofuro[2,3-c]pyridine-5-carboxylic acid (I-8-D) (74% yield). HRMS (FAB) calculated for $C_8H_7NO_3$+H: 166.0504, found 166.0498 (M+H).

**Intermediate D5: 3,3-Dimethyl-2,3-dihydrofuro[2,3-c]pyridine-5-carboxylic acid**

2-Chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-2-D) (6.3 g, 22 mmol) is dissolved in 30 mL DMF in a dry flask under $N_2$. The flask is placed in an ice bath,

and 60% sodium hydride in mineral oil (880 mg, 22 mmol) is added. The reaction is stirred 30 min while the flask is kept in an ice bath. The ice bath is removed for 30 min and then the flask is placed back into the ice bath to cool the reaction. 3-Bromo-2-methylpropene (23.1 mmol) is added, and the reaction is stirred overnight at rt. The reaction is diluted with 150 mL EtOAc and is washed with 4 x 50 mL 50% saturated 1:1 NaCl/$NaHCO_3$. The organic layer is dried ($Na_2SO_4$), filtered, and then concentrated *in vacuo* to a pale oil which is crystallized from hexanes to afford (6-chloro-4-iodo-5-[(2-methyl-2-propenyl)oxy]-2-pyridinyl)methanol (I-19-D) (86% yield). HRMS (FAB) calculated for $C_{10}H_{11}ClINO_2$+H: 339.9603, found 339.9604 (M+H).

I-19-D (6.3 g, 18.9 mmol), sodium formate (1.49 g, 21.8 mmol), TEA (8 mL, 57.2 mmol), palladium acetate (202 mg, 0.9 mmol) and tetra (n-butyl)ammonium chloride (5.25 g, 18.9 mmol) are added to 30 mL DMF in a dry flask under $N_2$. The reaction is warmed to 60°C for 5h, is poured into 150 mL EtOAc, and is washed with 4 x 50 mL 50% saturated 1:1 NaCl/$NaHCO_3$. The organic layer is dried ($Na_2SO_4$), filtered, and concentrated *in vacuo* to a pale oil. The crude material is chromatographed over 40 g silica gel (Biotage), eluting with 30% EtOAc/hexane to afford (7-chloro-3,3-dimethyl-2,3-dihydrofuro[2,3-c]pyridin-5-yl)methanol (I-20-D) (54% yield). MS (EI) for $C_{10}H_{12}ClNO_2$, *m/z*: 213 $(M)^+$.

I-20-D (2.11 g, 9.9 mmol) and 600 mg 10% Pd/C catalyst are placed in 30 mL EtOH in a 250 mL Parr shaker bottle. 2N NaOH (5 mL, 10 mmol) is then added and the mixture is hydrogenated at 20 PSI for 2.5 h. The catalyst is removed by filtration, and the filtrate is concentrated *in vacuo* to an aqueous residue. Saturated $NaHCO_3$ (20 mL) is added to the residue and extracted with 4 x 20 mL $CH_2Cl_2$. The combined organic layer is dried ($K_2CO_3$), filtered, and concentrated *in vacuo* to afford (3,3-dimethyl-2,3-dihydrofuro[2,3-c]pyridin-5-yl)methanol (I-21-D) (92% yield). MS (EI) for $C_{10}H_{13}NO_2$, *m/z*: 179 $(M)^+$.

Oxalyl chloride (869 µL, 9.9 mmol) is dissolved in 50 mL $CH_2Cl_2$ in a dry flask under $N_2$. The flask is placed in a dry-ice/acetone bath at -78°C, DMSO (1.41 mL, 19.8 mmol) in 5 mL $CH_2Cl_2$ is added drop-wise, and the mixture is stirred for 20 min. I-21-D (1.53 g, 8.5 mmol) in 5 mL $CH_2Cl_2$ is then added, and the reaction is stirred 30 min at -78°C. TEA (5.9 mL, 42.5 mmol) is added and the reaction is stirred 20 min at -78°C. The dry-ice/acetone bath is removed, the reaction is stirred 1h, and

the reaction is washed with 25 mL saturated $NaHCO_3$. The organic layer is dried ($K_2CO_3$), filtered, and then concentrated *in vacuo* to an orange solid. The crude material is chromatographed over 40 g silica gel (Biotage) eluting with 25% EtOAc/hexane to afford 3,3-dimethyl-2,3-dihydrofuro[2,3-c]pyridine-5-carbaldehyde (I-22-D) (92% yield). MS (EI) for $C_{10}H_{11}NO_2$, *m/z*: 177 $(M)^+$.

I-22-D (1.35 g, 7.62 mmol) is dissolved in 40 mL THF, 20 mL t-butanol, and 20 mL $H_2O$. $KH_2PO_4$ (3.11g, 22.9 mmol) and $NaClO_2$ (2.58 g, 22.9 mmol) are added, and the reaction is stirred over the weekend at rt. The reaction is concentrated *in vacuo* to a residue. The residue is partitioned between 20 mL water and $CH_2Cl_2$ (2 x 50 mL). The combined organic layer is dried ($Na_2SO_4$), filtered, and then concentrated *in vacuo* to afford crude 3,3-dimethyl-2,3-dihydrofuro[2,3-c]pyridine-5-carboxylic acid (I-23-D) (99% yield). HRMS (FAB) calculated for $C_{10}H_{11}NO_3$+H: 194.0817, found 194.0808 (M+H).

**Intermediate D6: 2-Methylfuro[2,3-c]pyridine-5-carboxylic acid**

2-Chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-2-D) (4.6 g, 16 mmol), propargyl trimethylsilane (2 g, 17.8 mmol), bis(triphenylphosphine) palladium dichloride (156 mg, 0.21 mmol), cuprous iodide (122 mg, 0.64 mmol), and piperidine (3.52 mL, 26.6 mmol) are added to 25 mL DMF in a dry flask under $N_2$. The mixture is warmed to 45°C for 7 h, is stirred overnight at rt, and is diluted with 150 mL EtOAc. The mixture is washed with 4 x 50 mL 50% saturated 1:1 NaCl/$NaHCO_3$. The organic layer is dried ($Na_2SO_4$), filtered, and then concentrated *in vacuo* to an amber oil. The crude material is chromatographed over 40 g silica gel (230-400 mesh) eluting with 35% EtOAc/hexane to afford (7-chloro-2-methylfuro[2,3-c]pyridin-5-yl)methanol (I-24-D) (44% yield). MS (CI) for $C_9H_8ClNO_2$, *m/z*: 198 (M+H).

I-24-D (2.0 g, 10.8 mmol) is added to 500 mg 10% Pd/C catalyst in 25 mL EtOH in a 250 mL Parr shaker bottle. 2N NaOH (6 mL, 12 mmol) is added, and the reaction is hydrogenated at 20 PSI for 6 h. The catalyst is removed by filtration, and the filtrate is concentrated *in vacuo* to an aqueous residue. The residue is partitioned between 50 mL 50% saturated NaCl and 30 mL $CH_2Cl_2$. The organic layer is dried ($K_2CO_3$), filtered, and then concentrated *in vacuo* to afford (2-methylfuro[2,3-

c]pyridin-5-yl)methanol (I-25-D) (77% yield). MS (CI) for $C_9H_9NO_2$, *m/z*: 164 (M+H).

Oxalyl chloride (784 µL, 8.9 mmol) is dissolved in 25 mL $CH_2Cl_2$ in a dry flask under $N_2$. The flask is placed in a dry-ice/acetone bath at -78°C, and DMSO (1.26 mL, 17.8 mmol) in 5 mL $CH_2Cl_2$ is added. The mixture is stirred for 20 min and I-25-D (1.53 g, 8.5 mmol) in 5 mL $CH_2Cl_2$ is added. The reaction is stirred 1 h, TEA (5.9 mL, 42.5 mmol) is added, and the reaction is stirred 30 min at -78°C. The flask is placed in an ice bath, and the reaction is stirred 1 h. The reaction is washed with 50 mL saturated $NaHCO_3$. The organic layer is dried ($K_2CO_3$), filtered, and then concentrated *in vacuo* to a tan solid. The crude material is chromatographed over 40 g silica gel (Biotage) eluting with 25% EtOAc/hexane to afford 2-methylfuro[2,3-c]pyridine-5-carbaldehyde (I-26-D) (99% yield). MS (EI) for $C_9H_7NO_2$, *m/z*: 161 $(M)^+$.

I-26-D (1.15 g, 7.1 mmol) is dissolved in 40 mL THF, 20 mL t-butanol, and 20 mL $H_2O$. 2-Methyl-2-butene (6.5 mL, 57.4 mmol) is added, and then $KH_2PO_4$ (3.11g, 22.9 mmol) and $NaClO_2$ (2.58 g, 22.9 mmol) are added. The reaction is stirred 6 h at rt. The reaction is concentrated *in vacuo*. Water (20 ml) is added to the residue, a white solid remained. The white solid is collected, washed with water and then with ether, and is dried to afford 2-methylfuro[2,3-c]pyridine-5-carboxylic acid (I-27-D) (70% yield). MS (EI) for $C_9H_7NO_3$, *m/z*: 177 $(M)^+$.

**Intermediate D7: 3-Methylfuro[2,3-c]pyridine-5-carboxylic acid**

2-Chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-2-D) (7.14 g, 25.0 mmol) is dissolved in DMF (50 mL) in a dry flask under $N_2$, sodium hydride (60% dispersion in mineral oil) (1.0 g, 25.0 mmol) is added, and the reaction is stirred for 1 h at rt. Allyl bromide (2.38 mL, 27.5 mmol) is added, and the reaction mixture is stirred 48h at rt. The mixture is diluted with EtOAc (50 mL) and washed 4 x 25 mL of a 50% saturated solution of 1:1 NaCl/$NaHCO_3$. The organic layer is dried ($MgSO_4$), filtered and concentrated *in vacuo* to a white solid. The solid is washed with hexane and dried to afford 3-(allyloxy)-2-chloro-6-(hydroxymethyl)-4-iodopyridine (I-50-D) as a white solid (68% yield). MS (EI) for $C_9H_9ClINO_2$, *m/z*: 325 $(M)^+$.

I-50-D (5.51 g, 16.9 mmol) is suspended in benzene (30 mL) in a dry flask under $N_2$. Azo(bis)isobutyryl nitrile (289 mg, 1.8 mmol) is added, the mixture is

rapidly heated to reflux, and tributyltin hydride (4.91 mL, 18.2 mmol) in benzene (10 mL) is added. The solution is refluxed for 1.5 h, allowed to cool to rt and concentrated *in vacuo*. The resulting residue is chromatographed over 125 g slurry-packed silica gel, eluting with a gradient of EtOAc/hexane (20% - 60%) to afford (7-chloro-3-methyl-2,3-dihydrofuro[2,3-c]pyridin-5-yl)methanol (I-51-D) as a white solid (89% yield). MS (ESI) for $C_9H_{10}ClNO_2$+H, *m/z*: 200.1 (M+H).

I-51-D (3.00 g, 15.0 mmol) is added to 20% palladium hydroxide on carbon (800 mg) and 2N NaOH (9.2 mL, 18.2 mmol) in a Parr shaker bottle. The mixture is hydrogenated at 20 PSI for 3 h, is filtered through celite and concentrated *in vacuo* to a residue. The resulting residue is partitioned between $H_2O$ (50 mL) and $CH_2Cl_2$ (4 x 30 mL). The combined organic layer is dried ($MgSO_4$), filtered, and concentrated to a colorless oil which solidified upon standing to afford 2.50 g (greater than 100% yield) of (3-methyl-2,3-dihydrofuro[2,3-c]pyridin-5-yl)methanol (I-52-D) as a white crystalline solid. MS (EI) for $C_9H_{11}NO_2$, *m/z*: 165 $(M)^+$.

I-52-D (2.48 g, 15.03 mmol) is dissolved in pyridine (15 mL), and acetic anhydride (4.18 mL, 45.09 mmol) is added and stirred for 16 h at rt under $N_2$. The reaction is concentrated *in vacuo*, and the residue is diluted with EtOAc (75 mL), washed with 50% saturated $NaHCO_3$ (4 x 30 mL), and dried ($MgSO_4$). The organic layer is filtered and concentrated *in vacuo* to afford (3-methyl-2,3-dihydrofuro[2,3-c]pyridin-5-yl)methyl acetate (I-53-D) as a colorless oil (92% yield). MS (EI) for $C_{11}H_{13}NO_3$, *m/z*: 207 $(M)^+$.

I-53-D (2.85 g, 13.8 mmol) is dissolved in dioxane (100 mL), 2,3,5,6-tertachlorobenzoquinone (3.72 g, 15.1 mmol) is added, and the reaction is heated to reflux for 17 h. The reaction is concentrated *in vacuo*. The resulting brown solid is washed with 1:1 EtOAc/ether (50 mL), and the insoluble material filtered off. The filtrate is concentrated to a brown solid, dissolved in MeOH (50 mL), treated with 2N NaOH (16 mL, 32 mmol), and stirred at rt for 1 h. The mixture is concentrated to dryness, dissolved in 1N NaOH (75 mL), and extracted with $CH_2Cl_2$ (4 x 50 mL). The combined organic layer is dried ($K_2CO_3$), filtered, and concentrated to a white solid (2.0 g). The crude material is adsorbed onto silica gel (4 g) and chromatographed over a standard 40 g Biotage column, eluting with 90% EtOAc/hexane to afford (3-methylfuro[2,3-c]pyridin-5-yl)methanol (I-54-D) as a white solid (84% yield). MS (EI) for $C_9H_9NO_2$, *m/z*: 163 $(M)^+$.

Oxalyl chloride (1.16 mL, 13.2 mmol) is added to $CH_2Cl_2$ (30 mL) in a dry flask under $N_2$ and in a dry-ice/acetone bath at -78°C. DMSO (18.80 mL, 26.5 mmol) is slowly added. The solution is stirred for 20 min, and I-54-D (1.88 g, 11.5 mmol) is added. The mixture is stirred for 1 h at -78°C, then 30 min at 0-5°C. The material is washed with saturated $NaHCO_3$ (75 mL), dried ($K_2CO_3$), filtered, and concentrated *in vacuo* to a yellow solid (3.23 g). The crude material is adsorbed onto silica gel (6 g) and chromatographed over a standard 40 g Biotage column, eluting with 25% EtOAc/hexane to afford 3-methylfuro[2,3-c]pyridine-5-carbaldehyde (I-55-D) as a white solid (72% yield). MS (EI) for $C_9H_7NO_2$, *m/z*: 161 $(M)^+$.

I-55-D (1.33 g, 8.28 mmol) is dissolved in THF (50 mL), *tert*-butylalcohol (25 mL) and $H_2O$ (25 mL), under $N_2$, and $NaClO_2$ (2.81 g, 24.84 mmol) and $KH_2PO_4$ (2.25 g, 16.56 mmol) are added. The reaction mixture is stirred overnight at rt, concentrated to dryness, dissolved in 50% saturated brine (60 mL) and extracted with ether (3 X). TLC of extracts indicates acid as well as residual aldehyde, so the organic and aqueous layers are combined and basified to pH 10 with $NH_4OH$. The layers are separated and the residual aldehyde extracted with additional ether. The aqueous layer is acidified to pH 3 with concentrated HCl, then extracted with $CH_2Cl_2$ (4 X). Large amounts of acid remained in the aqueous layer, so the aqueous layer is concentrated to dryness. The solid is triturated with $CHCl_3$ (4 X), and then 10% MeOH/$CH_2Cl_2$ (4 X) to extract much of the acid into the supernatant. The combined organic layer is dried ($Na_2SO_4$), filtered, and concentrated to a tan solid (1.69 g, greater than 100% isolated yield). The solid is diluted with $CHCl_3$ and is heated to reflux for 3 h. The flask is removed from heat, allowed to cool slightly, then filtered. The filtrate is concentrated to a tan solid (1.02 g). The solid is triturated with ether, filtered and dried to afford 3-methylfuro[2,3-c]pyridine-5-carboxylic acid (I-56-D) as a light tan solid (51% yield). MS (CI) for $C_9H_7NO_3$, *m/z*: 178 (M+H).

**Intermediate D8: 3-Ethylfuro[2,3-c]pyridine-5-carboxylic acid**

From 1-chloro-2-butene and 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-2-D), the corresponding 3-ethylfuro[2,3-c]pyridine-5-carboxylic acid (I-60-D) was prepared. HRMS (FAB) calculated for $C_{10}H_9NO_3$+H: 192.0661, found 192.0659 (M+H).

**Intermediate D10: Furo[2,3-b]pyridine-2-carboxylic**

Ethyl glycolate (35.5 mL, 375 mmol) is slowly added (over 20 min) to a slurry of NaOH (15.8 g, 394 mmol) in 1,2-dimethoxyethane (400 mL) under $N_2$ with the flask being in an ice bath. The mixture is allowed to warm to rt, is stirred for 30 min, and ethyl 2-chloronicotinate (27.84 g, 150 mmol) in 1,2-dimethoxyethane (50 mL) is added over 10 minutes. The reaction is warmed to 65°C for 15h in an oil bath. The mixture is concentrated to dryness, the residue is dissolved in $H_2O$ (500 mL), washed with hexane (500 mL), acidified to pH 3 with 5% HCl, and extracted with $CHCl_3$ (4 x 400 mL). The combined organic layer is dried ($MgSO_4$), filtered, and concentrated to a yellow solid. The solid is suspended in ether (200 mL) and heated on a steam bath until concentrated to a volume of 40 mL. The material is allowed to crystallize overnight, then filtered to afford ethyl 3-hydroxyfuro[2,3-b]pyridine-2-carboxylate (I-40-D) as a pale orange solid (41% yield). Additional material is obtained by concentrating the filtrate. Recrystallization in ether a second time afforded I-40-D as a pale yellow solid (7.3% yield). MS (EI) for $C_{10}H_9NO_4$, *m/z*: 207 $(M)^+$.

I-40-D (207 mg, 1.0 mmol) is added to TEA (139 µL, 1.0 mmol) in $CH_2Cl_2$ (5 mL) at rt and 2-[*N*,*N*-bis(trifluoromethylsulfonyl)amino]-5-chloropyridine (393 mg, 1.0 mmol) is added. The solution is stirred for 1 h at rt, diluted with EtOAc (25 mL) and washed with 50% saturated brine (2 x 15 mL). The organic layer is dried ($Na_2SO_4$), filtered, and concentrated to a yellow oil which solidified upon standing. The crude material is adsorbed onto silica gel (1.2 g) and chromatographed over 25 g slurry-packed silica gel, eluting with 20% EtOAc/hexane to afford ethyl 3-([(trifluoromethyl)sulfonyl]oxy)furo[2,3-b]pyridine-2-carboxylate (I-41-D) as a white crystalline solid (98% yield). Analysis calculated for $C_{11}H_8F_3NO_6S$: C, 38.94; H, 2.38; N, 4.13, found: C, 38.84; H, 2.29; N, 4.11.

I-41-D (1.36 g, 4.0 mmol) is added to 10% Pd/C catalyst (68 mg) and $NaHCO_3$ (336 mg, 4.0 mmol) in EtOH (100 mL)/$H_2O$ (5 mL) in a 250 mL Parr shaker bottle. The mixture is hydrogenated at 10 PSI for 5 h, filtered and concentrated to a residue. The residue is partitioned between 50% saturated $NaHCO_3$ (80 mL) and EtOAc (80 mL). The organic layer is dried ($Na_2SO_4$), filtered, and concentrated *in vacuo* to a colorless oil which solidified upon standing (793 mg). The crude material is chromatographed over 40 g slurry-packed silica gel, eluting with 25% EtOAc/hexane

to afford ethyl furo[2,3-b]pyridine-2-carboxylate (I-42-D) as a white solid (90% yield). MS (EI) for $C_{10}H_9NO_3$, *m/z*: 191 $(M)^+$.

I-42-D (758 mg, 3.96 mmol) is dissolved in MeOH (20 mL) and lithium hydroxide monohydrate (366 mg, 8.7 mmol) in 6mL $H_2O$ is added under $N_2$. The reaction is stirred at rt for 2 h, concentrated to near-dryness, diluted with $H_2O$ (5 mL) and acidified to pH 3 with 10% HCl. The resulting solid is collected by filtration, washed with additional water and dried to afford furo[2,3-b]pyridine-2-carboxylic acid (I-43-D) as a white solid (97% yield). MS (EI) for $C_8H_5NO_3$, *m/z*: 163 $(M)^+$.

**Intermediate D11: 3-Isopropylfuro[2,3-c]pyridine-5-carboxylic acid**

3-Isopropylfuro[2,3-c]pyridine-5-carboxylic acid (I-70-D) is obtained starting with 1-chloro-3-methyl-2-butene and 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-2-D), using the method described for Intermediate C7, making non-critical changes. HRMS (FAB) calculated for $C_{11}H_{11}NO_3$+H: 206.0817, found 206.0817 $(M+H)^+$.

**Intermediate D12: Thieno[2,3-b]pyridine-2-carboxylic acid**

THF (200 mL) in a dry flask under $N_2$ is chilled by placing the flask in a dry-ice/acetone bath at -78°C. Butyllithium (125 mL, 200 mmol) is added drop-wise, followed by the drop-wise addition of iodobenzene (11.19 mL, 100 mmol) in THF (10 mL). The solution is allowed to stir for 30 min at -78°C. Diisopropylamine (0.70 mL, 5 mmol) in THF (3 mL) and 2-chloropyridine (9.46 mL, 100 mmol) in THF (30 mL) are added successively in a drop-wise manner, and the solution is stirred for 1 h at -40°C. Formyl piperidine (11.1 mL, 100 mmol) in THF (25 mL) is added drop-wise, and the solution is stirred for 1 h at -40°C. The reaction is quenched with 40 mL 6N HCl, diluted with 250 mL ether, and a small amount of sodium thiosulfate solution is added to remove the iodine color. The solution is neutralized with saturated $NaHCO_3$, filtered, and extracted with ether (3 x 150 mL). The combined organic layer is dried ($Na_2SO_4$), filtered, and concentrated *in vacuo*. The crude material is chromatographed over 600 g slurry-packed silica, eluting with 20% EtOAc/hexane to afford 2-chloronicotinaldehyde (I-90-D) as a pale orange solid (54% yield). MS (EI) for $C_6H_4ClNO$, *m/z*: 141 $(M)^+$.

I-90-D (1.41 g, 10.01 mmol) is dissolved in DMF (10mL) and $H_2O$ (1 mL) under $N_2$. $K_2CO_3$ (1.56 g, 11.27 mmol) and methyl thioglycolate (1.00 mL, 11.25

mmol) are added portionwise. The reaction is stirred at 35°C for 24 h, quenched with cold $H_2O$ (75 mL), and placed in an ice bath to enhance precipitation. The precipitate is isolated by filtration, affording methyl-thieno[2,3-b]pyridine-2-carboxylate (I-101-D) as an orange powder (40% yield). MS (EI) for $C_9H_7NO_2S$, *m/z*: 193 $(M)^+$.

I-101-D (0.700 g, 3.63 mmol) is dissolved in MeOH (15 mL) and 3 mL $H_2O$. 2N NaOH (1.82 mL, 3.63 mmol) is added drop-wise, and the reaction is stirred at rt for 24 h. The reaction is concentrated *in vacuo*, and $H_2O$ (40 mL) is added to dissolve the residue. The resulting solution is acidified to pH 4 using concentrated HCl, and the precipitate is isolated by filtration, yielding thieno[2,3-b]pyridine-2-carboxylic acid (I-102-D) as a white powder (85% yield). MS (EI) for $C_8H_5NO_2S$, *m/z*: 179 $(M)^+$.

**Intermediate D13: Thieno[2,3-b]pyridine-5-carboxylic acid**

2-Nitrothiophene (33.76 g, 261.4 mmol) is suspended in concentrated HCl (175 mL) and heated to 50°C. Stannous chloride (118.05 g, 523.2 mmol) is added portionwise, maintaining the reaction temperature between 45-50°C with an ice bath, that is removed after the addition. The solution is allowed to cool slowly to 30°C over an hour. The solution is then cooled in an ice bath and filtered. The cake is washed with concentrated HCl (20 mL), dried in a stream of air, and washed with ether (50 mL) to afford the hexachlorostannate salt of 2-aminothiophene as a brown solid (26% yield).

3,3-Dimethyl-2-formyl propionitrile sodium (3.33 g, 20.2 mmol) can readily be prepared from the method described by Bertz, S.H., et al., *J. Org. Chem.*, 47, 2216-2217 (1982). 3,3-Dimethyl-2-formyl propionitrile sodium is dissolved in MeOH (40 mL), and concentrated HCl (4 mL) and the hexachlorostannate salt of 2-aminothiophene (10.04 g, 19.1 mmol) in MeOH (130 mL) is slowly added drop-wise to the mixture. Following addition, the mixture is heated to reflux in an oil bath (80°C) for 4 h, and then MeOH (10 mL) and concentrated HCl (10 mL) are added. The reaction continued refluxing for another 20 h. The solution is cooled to rt, and the reaction is concentrated *in vacuo*. The purple residue is dissolved in $H_2O$ (60 mL), and the slurry is filtered. The cake is pulverized and stirred vigorously with 5% MeOH/$CHCl_3$ (105 mL) while heating to 55°C. The mixture is cooled and filtered, and the organic layer is concentrated to a green oil. The crude material is

chromatographed over 130 g slurry-packed silica, eluting with 30% EtOAc/hexane to afford thieno[2,3-b]pyridine-5-carbonitrile (I-105-D) as a pale yellow solid (24% yield). HRMS (FAB) calculated for $C_8H_4N_2S$+H: 161.0173, found 161.0173 (M+H).

NaOH (0.138 g, 3.45 mmol) is added to a solution of I-105-D (0.503 g, 3.14 mmol) dissolved in 70% EtOH/$H_2O$ (12 mL). The mixture is heated to reflux at 100°C for 3 h. The reaction is concentrated *in vacuo*, and the residue is dissolved in $H_2O$ (8 mL) and neutralized with concentrated HCl. The slurry is filtered and rinsed with ether. An initial NMR of the isolated material indicates the presence of the carboxamide intermediate, so the material is suspended in 1M NaOH (6 mL) and stirred overnight. Water (10 mL) is added, the solution is extracted with ether (3 x 10 mL), and the mixture is neutralized with concentrated HCl. The slurry is filtered and rinsed with ether, affording of thieno[2,3-b]pyridine-5-carboxylic acid (I-106-D) as an off-white solid (48% yield). MS (EI) for $C_8H_5NO_2S$, *m/z*: 179 $(M)^+$.

**Intermediate D14: Thieno[2,3-b]pyridine-6-carboxylic acid**

2-Nitrothiophene (12.9 g, 99.9 mmol) is dissolved in concentrated HCl (200 mL) and stirred vigorously at 30°C. Granular tin (25 g, 210 mmol) is slowly added portionwise. When the tin is completely dissolved, zinc chloride (6.1 g, 44.7 mmol) in EtOH (70 mL) is added drop-wise, the mixture is heated to 85°C, and malondialdehyde diethyl acetal (24 mL, 100 mmol) in EtOH (30 mL) is added. The solution continued stirring at 85°C for 1 h, and is quenched by pouring over ice (100 g). The mixture is adjusted to pH 10 with $NH_4OH$, and the resulting slurry is carefully filtered through celite overnight. The liquor is extracted with $CHCl_3$ (3 x 300 mL), and the combined organic layer is dried ($MgSO_4$), filtered, and concentrated to a brown oil. The crude material is chromatographed over 250 g slurry-packed silica, eluting with 35% EtOAc/hexane to give thieno[2,3-b] pyridine (I-110-D) as an orange oil (26% yield). MS (EI) for $C_7H_5NS$, *m/z*: 135 $(M)^+$.

I-110-D (3.47 g, 25.7 mmol) is dissolved in acetic acid (12 mL) and heated to 85°C. 30% Hydrogen peroxide (9 mL) is added drop-wise and the solution is allowed to stir overnight. The reaction is allowed to cool to rt and quenched with paraformaldehyde until a peroxide test proved negative using starch-iodine paper. The solution is diluted with $H_2O$ (100 mL) and neutralized with $NaHCO_3$, then extracted repeatedly with $CHCl_3$ (12 x 80 mL, 6 x 50 mL). The combined organic

layer is dried ($Na_2SO_4$), filtered, and concentrated to a brown solid. The crude material is chromatographed over 70 g slurry-packed silica eluting with 3.5% MeOH/$CH_2Cl_2$ to afford thieno[2,3-b] pyridine-7-oxide (I-111-D) as a pale yellow solid (22% yield). MS (EI) for $C_7H_5NOS$ *m/z*: 151 $(M)^+$.

A 0.5M solution of I-111-D (5 mL, 2.5 mmol) in $CH_2Cl_2$ is diluted with 8 mL of $CH_2Cl_2$ under $N_2$. Dimethyl carbamyl chloride (0.27 mL, 2.9 mmol) is added drop-wise, followed by the addition of trimethylsilyl cyanide (0.388 mL, 2.9 mmol) via syringe. The reaction is allowed to stir for 9 days and is quenched with 10% $K_2CO_3$ (10 mL). The layers are allowed to separate, the organic layer is isolated and dried ($K_2CO_3$), filtered, and concentrated to a brown solid. The crude material is chromatographed over 25 g slurry-packed silica, eluting with 35% EtOAc/hexane to afford thieno[2,3-b]pyridine-6-carbonitrile (I-112-D) as a pale yellow solid (100% yield). Analysis calculated for $C_8H_4N_2S$: C, 59.98; H, 2.52; N, 17.49, found: C, 59.91; H, 2.57; N, 17.43.

NaOH (398 mg, 9.95 mmol) is added portionwise to a solution of I-112-D (674 mg, 4.2 mmol) in 70% EtOH/$H_2O$ (20 mL). The solution is heated to reflux at 100°C for 24 h, and the reaction is concentrated *in vacuo*. The residue is dissolved in $H_2O$ (15 mL) and washed with ether (3 x 10 mL). Concentrated HCl is used to adjust the pH to 3.5, creating a precipitate. The slurry is filtered, giving thieno[2,3-b]pyridine-6-carboxylic acid (I-113-D) as a white solid (45% yield). MS (EI) for $C_8H_5NO_2S$, *m/z*: 179$(M)^+$.

**Intermediate D15: Thieno[2,3-c]pyridine-2-carboxylic acid**

THF (200 mL) is chilled to –70°C in a dry flask under $N_2$, and N-butyllithium (24.4 mL, 55.0 mmol) is added drop-wise. The reaction is placed in an ice bath and DIA (7.71 mL, 55.0 mmol) in THF (20 mL) is added drop-wise. The solution is again chilled to –70°C, and 3-chloropyridine (4.75 mL, 50.0 mmol) in THF (20 mL) is added drop-wise. The reaction is allowed to stir for 4 h at –70°C and ethyl formate (4.44 mL, 55.0 mmol) in THF (20 mL) is added. The reaction is stirred for an additional 3 h at –70°C and quenched with $H_2O$ (500 mL). The layers are allowed to separate, and the aqueous layer is extracted with EtOAc (3 x 250 mL). The combined organic layer is dried ($MgSO_4$), filtered, and concentrated to a dark brown solid. The crude material is chromatographed over 250 g slurry-packed silica, eluting with 50%

EtOAc/hexane to give 3-chloroisonicotinaldehyde (I-120-D) as an off-white solid (55% yield). MS (EI) for $C_6H_4ClNO$, *m/z*: 141 $(M)^+$.

I-120-D (2.12 g, 14.9 mmol) is dissolved in DMF (75 mL) with a small amount of $H_2O$ (7.5 mL). Methyl thioglycolate (1.67 mL, 18.7 mmol) and $K_2CO_3$ (2.59 g, 18.7 mmol) are added portionwise, and the mixture is stirred at 45°C for 24 h. The reaction is quenched with cold $H_2O$ (200 mL) and extracted with EtOAc (3 x 150 mL). The combined organic layer is washed with 50% NaCl solution (3 x 150 mL), dried ($MgSO_4$), filtered, and concentrated to an orange solid. The crude material is chromatographed over 40 g slurry-packed silica, eluting with 50% EtOAc/hexane to afford ethyl thieno[2,3-c]pyridine-2-carboxylate (I-121-D) as a pale yellow solid (22% yield).

I-121-D (577 mg, 2.99 mmol) is combined with 2M NaOH (1.5 mL, 3.0 mmol) in MeOH (15 mL) and $H_2O$ (1.5 mL). The reaction is stirred at rt for 24 h. The reaction is concentrated *in vacuo* and the residue is dissolved in $H_2O$ (75 mL). Concentrated HCl is used to acidify the solution to pH 3. The slurry is filtered, washed with $H_2O$ and ether, and dried, affording thieno[2,3-c]pyridine-2-carboxylic acid (I-122-D) as an off-white solid (38% yield). HRMS (FAB) calculated for $C_8H_5NO_2S$+H: 180.0119, found 180.0119 (M+H).

**Intermediate D16: Thieno[3,2-b]pyridine-2-carboxylic acid**

3-Chloropyridine (9.5 mL. 99.9 mmol) is dissolved in acetic acid (35 mL) and heated to 98°C. 30% Hydrogen peroxide (28 mL) is added drop-wise, and the reaction stirred for 5 h at 98°C. The reaction is cooled and paraformaldehyde is added so that a negative peroxide test is achieved using starch-iodine paper. The solution is concentrated *in vacuo* and the crude paste is chromatographed over 600 g slurry-packed silica eluting with 4 L of 2% MeOH/$CH_2Cl_2$, 2 L of 4% MeOH/$CH_2Cl_2$, and finally 1 L of 10% MeOH/$CH_2Cl_2$ to afford 3-chloropyridine 1-oxide (I-125-D) as a pale oil (100% yield).

A 2M solution of I-125-D (10 mL, 20 mmol) is combined with an additional 90 mL of $CH_2Cl_2$. Dimethylcarbamoyl chloride (2.03 mL, 22.0 mmol) is added drop-wise, followed by the addition of trimethyl silylcyanide (2.93 mL, 22.0 mmol) via syringe. The reaction is stirred at rt for 10 days and is quenched with 10% $K_2CO_3$ (100 mL). The layers are allowed to separate, and the organic layer is dried ($K_2CO_3$),

filtered, and concentrated to an orange solid. The crude material is chromatographed over 160 g slurry-packed silica eluting with 40% EtOAc/hexane to yield 3-chloropyridine-2-carbonitrile (I-126-D) as a white solid (59% yield). MS (EI) for $C_6H_3ClN_2$, *m/z*: 138 $(M)^+$.

I-126-D (1.01 g, 7.29 mmol) and $K_2CO_3$ (1.10 g, 7.96 mmol) are added to DMF (10 mL) and $H_2O$ (1 mL). Methyl thioglycolate (0.709 mL, 7.93 mmol) is added drop-wise, and the solution is heated to 40°C and stirred for 3 h. The reaction is quenched with cold $H_2O$ (70 mL) and placed on ice to enhance precipitation. The slurry is filtered and the cake is dissolved in $CHCl_3$. This organic solution is dried ($MgSO_4$), filtered, and concentrated, affording methyl 3-aminothieno[3,2-b]pyridine-2-carboxylate (I-127-D) as a yellow solid (84% yield). HRMS (FAB) calculated for $C_9H_8N_2O_2S$+H: 209.0385, found 209.0383 (M+H).

I-127-D (0.919 g, 4.42 mmol) is dissolved in 50% hypophosphorous acid (35 mL) and chilled in an ice bath. Sodium nitrite (0.61 g, 8.84 mmol) is dissolved in a minimal amount of $H_2O$ and added drop-wise to the previous solution, and the reaction is stirred for 3 h in an ice bath. 3M NaOH is used to adjust the pH to 7.9, and the solution is extracted with EtOAc (3 x 100 mL). The combined organic layer is dried ($MgSO_4$), filtered, and concentrated to afford methyl thieno[3,2-b]pyridine-2-carboxylate (I-128-D) as a yellow solid (44% yield). MS (EI) for $C_9H_7NO_2S$, *m/z*: 193 $(M)^+$.

2M NaOH (0.8 mL, 1.6 mmol) and I-128-D (300 mg, 1.55 mmol) are added to MeOH (8 mL) and $H_2O$ (1 mL) and is stirred for 24 h. The reaction is concentrated *in vacuo*, and the residue is dissolved with $H_2O$ (5 mL). 5% HCl is used to adjust the pH to 3.5, creating a precipitate. The slurry is filtered and washed with ether, affording thieno[3,2-b]pyridine-2-carboxylic acid (I-129-D) as a brown solid (67% yield). HRMS (FAB) calculated for $C_8H_5NO_2S$+H: 180.0119, found 180.0121 (M+H).

**Intermediate D17: Thieno[3,2-b]pyridine-6-carboxylic acid**

Methyl 3-aminothiophene-2-carboxylate (1.52 g, 9.68 mmol) is dissolved in 2M NaOH (10 mL, 20 mmol) and heated to reflux in a 115°C oil bath for 30 min. The mixture is cooled to rt, placed in an ice bath, and carefully acidified with concentrated HCl. The slurry is filtered and rinsed with $H_2O$ (25 mL). The cake is then dissolved in acetone (50 mL), dried ($MgSO_4$), filtered, and concentrated to a thick paste. The

crude material is dissolved in 1-propanol (25 mL), and oxalic acid (0.90 g, 10.0 mmol) is added portionwise. The mixture is heated at 38°C for 45 min, cooled to rt, and diluted with ether. The precipitate is isolated via filtration, and washed with ether, affording 3-amino-thiophene oxalate (I-135-D) as a fluffy white solid (70% yield). HRMS (FAB) calculated for $C_4H_5NS$+H: 100.0221, found 100.0229 (M+H).

3,3-Dimethyl-2-formyl propionitrile sodium (5.38 g, 32.6 mmol) is dissolved in MeOH (60 mL) with concentrated HCl (6 mL). I-135-D (6.16 g, 32.6 mmol) is suspended in MeOH (200 mL) and added drop-wise to the acidic solution. The mixture is heated to reflux at 80°C for 5 h when an additional 20 mL concentrated HCl and 20 mL $H_2O$ are added; the mixture continues refluxing for another 12 h. The mixture is concentrated *in vacuo*, and the residue is dissolved with cold $H_2O$ (100 mL). The resulting precipitate is filtered off and dried, giving thieno[3,2-b]pyridine-6-carbonitrile (I-136-D) as a brown solid (44% yield). HRMS (FAB) calculated for $C_8H_4N_2S$+H: 161.0173, found 161.0170 (M+H).

I-136-D (1.99 g, 12.5 mmol) is dissolved in 70% EtOH/$H_2O$ (20 mL), and NaOH (0.52 g, 13.0 mmol) is added portionwise. The mixture is heated at 100°C for 15 h and then allowed to cool to rt. The mixture is concentrated *in vacuo*. The residue is dissolved in cold $H_2O$ (30 mL), and the solution is rinsed with ether (3 x 10 mL). The pH is adjusted to 3.5 with concentrated HCl to precipitate the desired product that is removed by filtration to give thieno[3,2-b]pyridine-6-carboxylic acid (I-137-D) as a tan solid (77% yield). HRMS (FAB) calculated for $C_8H_5NO_2S$+H: 180.0119, found 180.0118 (M+H).

**Intermediate D18: Thieno[3,2-c]pyridine-2-carboxylic acid**

4-Chloropyridine hydrochloride (15 g, 99.9 mmol) is free-based by stirring in 1000mL 1:1 saturated $NaHCO_3$/ether for 1 h. The layers are allowed to separate, the aqueous layer is extracted with ether (2 x 175 mL), and the combined organic layer is dried ($MgSO_4$), filtered, and concentrated to an oil. THF (300 mL) is chilled to -70°C in a dry flask. N-butyllithium (105.1 mL, 168.2 mmol) is added drop-wise, and the mixture is placed in an ice bath. Diisopropylamine (23.6mL. 168.4 mmol) in THF (50 mL) is added drop-wise, the yellow solution is stirred for 30 min, and the reaction is cooled to -70°C. The free-based 4-chloropyridine oil (9.55 g, 84.1 mmol) is dissolved in THF (50 mL) and added drop-wise to the chilled yellow solution, that turned dark

red after the addition. The reaction is stirred at –70°C for 2 h. Ethyl formate (13.6 mL, 168.3 mmol) in THF (25 mL) is then added drop-wise to the dark solution at -70°C. After 2 hours, the reaction is warmed to -10°C and quenched with water (450 mL). The layers are allowed to separate, and the aqueous layer is extracted with ether (3 x 200 mL). The combined organic layer is dried ($MgSO_4$), filtered, and concentrated *in vacuo* to an oil. The crude material is chromatographed over 320 g slurry-packed silica eluting with 30% EtOAc/hexane to afford 4-chloropyridine-3-carboxaldehyde (I-140-D) an orange oil which solidified under vacuum to an orange solid (21% yield).

I-140-D (2.53 g, 17.9 mmol) is dissolved in DMF (20 mL) and $H_2O$ (2 mL). $K_2CO_3$ (2.97 g, 21.5 mmol) and methyl thioglycolate (1.92 mL, 21.5 mmol) are added portionwise. The reaction is stirred at 45°C for 24 h, then quenched with cold $H_2O$ (100 mL), and the flask is placed on ice to enhance precipitation. The precipitate is isolated by filtration and dried, affording methyl thieno[3,2-c]pyridine-2-carboxylate (I-141-D) as a white solid (92% yield). MS (EI) for $C_9H_7NO_2S$, *m/z*: 193 $(M)^+$.

I-141-D (2.65 g, 13.7 mmol) is dissolved in MeOH (70 mL) and $H_2O$ (5 mL). 2N NaOH (6.86 mL, 13.7 mmol) is added drop-wise, and the reaction is stirred at rt for 24 h. The reaction is concentrated *in vacuo*, and $H_2O$ (150 mL) is added to dissolve the residue. The resulting salt solution is acidified to pH 3.5 using concentrated HCl, and the precipitate is isolated by filtration and dried, affording thieno[3,2-c]pyridine-2-carboxylic acid (I-142-D) as a white powder (57% yield). HRMS (FAB) calculated for $C_8H_5NO_2S$+H: 180.0119, found 180.0124 (M+H).

**Intermediate D19: Thieno[2,3-c]pyridine-5-carboxylic acid**

Glyoxylic acid monohydrate (20.3 g, 221 mmol) and benzyl carbamate (30.6 g, 202 mmol) are added to ether (200 mL). The solution is allowed to stir for 24 h at rt. The resulting thick precipitate is filtered, and the residue is washed with ether, affording ([(benzyloxy)carbonyl]amino)(hydroxy)acetic acid (I-150-D) as a white solid (47% yield). MS (CI) for $C_{10}H_{11}NO_5$+H *m/z*: 226 (M+H).

I-150-D (11.6 g, 51.5 mmol) is dissolved in absolute MeOH (120 mL) and chilled in an ice bath. Concentrated sulfuric acid (2.0 mL) is carefully added drop-wise. The ice bath is allowed to expire as the solution stirred for 2 days. The reaction is quenched by pouring onto a mixture of 500 g ice with saturated $NaHCO_3$ solution

(400 mL). The solution is extracted with EtOAc (3 x 300 mL), and the combined organic layer is dried ($MgSO_4$), filtered, and concentrated to a pale oil that crystallized upon standing, giving methyl([(benzyloxy)carbonyl]amino)(methoxy)-acetate (I-151-D) as a white solid (94% yield). Analysis calculated for $C_{12}H_{15}NO_5$: C, 56.91; H, 5.97; N, 5.53, found: C, 56.99; H, 6.02; N, 5.60.

I-151-D (11.76 g, 46.4 mmol) is dissolved in toluene (50 mL) under $N_2$ and heated to 70°C. Phosphorous trichloride (23.2 mL, 46.4 mmol) is added drop-wise via syringe, and the solution is stirred for 18 h at 70°C. Trimethyl phosphite (5.47 mL, 46.4 mmol) is then added drop-wise, and stirring continued for an additional 2 h at 70°C. The mixture is concentrated *in vacuo* to an oil, and the crude material is dissolved in EtOAc (100 mL) and washed with saturated $NaHCO_3$ (3 x 50 mL). The organic layer is dried ($Na_2SO_4$), filtered, and concentrated to a volume of 30 mL. This remaining solution is stirred vigorously while hexane is added until a precipitate formed. The precipitated solid is removed by filtration, affording methyl ([(benzyloxy)carbonyl]amino) (dimethoxyphosphoryl)acetate (I-152-D) as a white solid (84% yield). MS (EI) for $C_{13}H_{18}NO_7P$, *m/z*: 331 $(M)^+$.

I-152-D (12.65 g, 38.2 mmol) and acetic anhydride (9.02 mL, 95.5 mmol) in MeOH (100 mL) were added to a Parr flask. The solution is hydrogenated with 10% Pd/C catalyst (0.640 g) at 45 PSI for 3h. The catalyst is filtered off, and the filtrate is concentrated *in vacuo* to an oil. The oil is placed under reduced pressure and solidified as the reduced pressure is applied. The white residue is dissolved in a small amount of EtOAc and stirred vigorously while pentane is added until a precipitate began to form. The precipitate is removed by filtration to give methyl (acetylamino)(dimethoxyphosphoryl)acetate (I-153-D) as a white powder (87% yield). MS (CI) for $C_7H_{14}NO_6P$, *m/z*: 240 (M+H).

2,3-Thiophene dicarboxaldehyde (1.40 g, 9.99 mmol) is dissolved in $CH_2Cl_2$ (100 mL) and the flask is placed in an ice bath. I-152-D (2.63 g, 11.0 mmol) is dissolved in $CH_2Cl_2$ (50 mL), 1,8-diazabicyclo[5.4.0]undec-7-ene (1.65 mL, 11.0 mmol) is added, and this solution is added drop-wise to the chilled thiophene solution. The reaction mixture is stirred for 1 h while the flask is in an ice bath and then over night at rt. The reaction is concentrated *in vacuo*, and the crude material is chromatographed over 300 g slurry-packed silica eluting with 50% EtOAc/hexane. The fractions were collected in two different groups to obtain the desired compounds.

Each group of fractions is combined and concentrated separately. The first group of fractions affords methyl thieno[2,3-c]pyridine-5-carboxylate (I-154-D) as a white solid (41% yield), and the second group of fractions affords methyl thieno[3,2-c]pyridine-6-carboxylate (I-155-D) as a yellow solid (38% yield). MS (EI) for I-154-D for $C_9H_7NO_2S$, *m/z*: 193 $(M)^+$. MS (EI) for I-155-D for $C_9H_7NO_2S$, *m/z*: 193 $(M)^+$.

I-154-D (736 mg, 3.8 mmol) is dissolved in MeOH (16 mL) with water (2 mL). 2M NaOH (2.0 mL, 4.0 mmol) is added drop-wise and the solution stirred at rt. After 2 days (complete disappearance of ester by TLC), the reaction is concentrated *in vacuo*. The residue is dissolved in $H_2O$ (12 mL), and the pH is adjusted to 3.5 with 10% HCl. The precipitated solid is removed by filtration, and the solid is rinsed with ether, affording thieno[2,3-c]pyridine-5-carboxylic acid (I-156-D) as a white solid (58% yield). HRMS (FAB) calculated for $C_8H_5NO_2S$+H: 180.0119, found 180.0123 (M+H).

**Intermediate D20: Thieno[3,2-c]pyridine-6-carboxylic acid**

Methyl thieno[3,2-c]pyridine-6-carboxylate (I-155-D) (678 mg, 3.5 mmol) is dissolved in MeOH (16 mL) and $H_2O$ (2 mL). 2M NaOH (1.8 mL, 3.6 mmol) is added drop-wise, and the solution stirred at rt. After 2 days (complete disappearance of ester by TLC), the solution is concentrated *in vacuo*. The residue is dissolved in $H_2O$ (12 mL), and the pH is adjusted to 3.5 with 10% HCl. The precipitated solid is removed by filtration, and the solid is rinsed with ether, affording thieno[3,2-c]pyridine-6-carboxylic acid (I-160-D) as a white solid (43% yield). HRMS (FAB) calculated for $C_8H_5NO_2S$+H: 180.0119, found 180.0123 (M+H).

**Intermediate D21: 1*H*-Pyrrolo[2,3-c]pyridine-5-carboxylic acid**

2,4-Lutidine (51.4 mL, 0.445 mole) is added drop-wise to 250 mL fuming sulfuric acid in a flask under $N_2$ in an ice bath. The solution is treated portionwise with potassium nitrate (89.9 g, 0.889 mole) over a 15 min period. The reaction is stirred 1h in an ice bath, 2 h at rt, is gradually warmed in a 100°C oil bath for 5 h, and then in a 130°C oil bath for 4 h. The mixture is cooled, is poured into 1000 mL ice, and the mixture is neutralized with $NaHCO_3$ (1,100 g, 13.1 mole). The precipitated $Na_2SO_4$ is removed by filtration, the solid is washed with 500 mL $H_2O$ and the filtrate is extracted with 4 x 500 mL ether. The combined organic layer is dried ($MgSO_4$) and

is concentrated *in vacuo* to a yellow oil (50 g). The crude oil is distilled under vacuum to provide three fractions: 16 g recovered 2,4-lutidine (85°C), 16 g 2,4-dimethyl-3-nitro-pyridine (I-169-D) contaminated with 25% 2,4-dimethyl-5-nitro-pyridine (135-145°C), and 16 g 2,4-dimethyl-5-nitro-pyridine (I-170-D) contaminated with 2,4-dimethyl-3-nitropyridine (145-153°C). $^1$H NMR of C169 ($CDCl_3$) δ 2.33, 2.54, 7.10, 8.43 ppm. $^1$H NMR of C170 ($CDCl_3$) δ 2.61, 2.62, 7.16, 9.05 ppm.

I-170-D/I-169-D (75:25) (5.64 g, 37 mmol) is combined with benzeneselenic anhydride (8.2 g, 22.8 mmol) in 300 mL dioxane in a flask under $N_2$. The reaction is warmed to reflux for 10 h, is cooled, and is concentrated to a dark yellow oil. The oil is chromatographed over 250 g silica gel (230-400 mesh) eluting with 15% EtOAc/hexane to afford 2-formyl-4-methyl-5-nitropyridine (I-171-D) (66% yield). HRMS (EI) calculated for $C_7H_6N_2O_3$: 166.0378, found 166.0383 ($M^+$).

I-171-D (1.15 g, 6.9 mmol), p-toluene sulfonic acid (41 mg, 0.22 mmol), and ethylene glycol (1.41 mL, 25 mmol) are added to 25 mL toluene in a flask equipped with a Dean-Starke trap. The reaction is warmed to reflux for 2 h, is cooled to rt, and is concentrated *in vacuo* to an oily residue. The crude oil is chromatographed over 40 g silica gel (Biotage), eluting with 20% EtOAc/hexane to afford 2-(1,3-dioxolan-2-yl)-4-methyl-5-nitropyridine (I-172-D) (90% yield). MS (EI) for $C_9H_{10}N_2O_4$, *m/z*: 210 $(M)^+$.

I-172-D (1.3 g, 6.2 mmol) and DMF dimethyl acetal (1.12 mL, 8.4 mmol) are added to 15 mL DMF under $N_2$. The reaction is warmed to 90°C for 3 h, is cooled, and the reaction is concentrated *in vacuo*. The residue is combined with 1.25 g 5% $Pd/BaSO_4$ in 20 mL EtOH in a 250 mL Parr shaker bottle and the mixture is hydrogenated at ambient pressure until uptake ceased. The catalyst is removed by filtration, and the filtrate is combined with 500 mg 10% Pd/C catalyst in a 250 mL Parr shaker bottle. The mixture is hydrogenated at ambient pressure for 1 h. No additional hydrogen uptake is observed. The catalyst is removed by filtration, and the filtrate is concentrated *in vacuo* to a tan solid. The crude material is chromatographed over 50 g silica gel (230-400 mesh), eluting with 7% MeOH/$CH_2Cl_2$. The appropriate fractions are combined and concentrated to afford 5-(1,3-dioxolan-2-yl)-1H-pyrrolo[2,3-c]pyridine (I-173-D) (69%yield). MS for $C_{10}H_{10}N_2O_2$, (EI) *m/z*: 190 $(M)^+$.

I-1730-D (800 mg, 4.21 mmol) is dissolved in 44 mL 10% aqueous acetonitrile. p-Toluene sulfonic acid (630 mg, 3.3 mmol) is added, and the mixture is heated to reflux for 5 h. The mixture is cooled to rt, is concentrated *in vacuo,* and the resultant residue is diluted with 15 mL saturated $NaHCO_3$. A pale yellow solid is collected, washed with water, and is dried to afford 1H-pyrrolo[2,3-c]pyridine-5-carbaldehyde (I-174-D) (81% yield). HRMS (FAB) calculated for $C_8H_6N_2O$+H: 147.0558, found 147.0564 (M+H).

I-174-D (500 mg, 3.42 mmol) is dissolved in 1.5 mL formic acid. The solution is cooled in an ice bath, 30% aqueous hydrogen peroxide (722 µL, 6.8 mmol) is added drop-wise, and the reaction is stirred 1 h in an ice bath, and allowed to stand overnight at 5°C. The mixture is diluted with $H_2O$, the solid is collected, washed with $H_2O$ and is dried to give 522 mg of an off-white solid. The formate salt is added to 7 mL $H_2O$, 3 mL 2N NaOH is added, and the pH is adjusted to 3 with 5% aqueous HCl. The precipitate is collected and is dried to afford 1*H*-pyrrolo[2,3-c]pyridine-5-carboxylic acid (I-176-D) (67% yield). HRMS (FAB) calculated for $C_8H_6N_2O_2$+H: 163.0508, found 163.0507 (M+H).

**Intermediate D22: 1-Methyl-pyrrolo[2,3-c]pyridine-5-carboxylic acid**

5-(1,3-Dioxolan-2-yl)-1H-pyrrolo[2,3-c]pyridine (I-173-D) (1.05 g, 5.52 mmol) is dissolved in 20 mL THF in a dried flask under $N_2$. 60% Sodium hydride (243 mg, 6.07 mmol) is added, the reaction is stirred 30 min, methyl iodide (360 µL, 5.8 mmol) is added, and the reaction is stirred overnight at rt. The reaction is concentrated *in vacuo* and the residue is partitioned between 10 mL saturated NaCl and $CH_2Cl_2$ (4 x 10 mL). The combined organic layer is dried ($K_2CO_3$) and is concentrated *in vacuo* to a tan paste. The crude material is chromatographed over 50 g silica gel (230-400 mesh) eluting with 5% MeOH/$CH_2Cl_2$. The appropriate fractions are combined and concentrated to afford 5-(1,3-dioxolan-2-yl)-1-methyl-1H-pyrrolo[2,3-c]pyridine (I-175-D) (86% yield). HRMS (FAB) calculated for $C_{11}H_{12}N_2O_2$+H: 205.0977, found 205.0983.

I-175-D (920 mg, 4.5 mmol) is dissolved in 25 mL 10% aqueous acetonitrile in a flask. p-Toluene sulfonic acid (630 mg, 3.3 mmol) is added, and the mixture is heated to 90°C for 8 h. The mixture is cooled to rt, concentrated *in vacuo*, and the residue is partitioned between 15 mL saturated $NaHCO_3$ and $CH_2Cl_2$ (4 x 10 mL).

The combined organic layer is dried ($K_2CO_3$) and is concentrated *in vacuo* to afford 1-methyl-pyrrolo[2,3-c]pyridine-5-carbaldehyde (I-177-D) (99% yield). HRMS (FAB) calculated for $C_9H_8N_2O$+H: 161.0715, found 161.0711.

I-177-D (690 mg, 4.3 mmol) is dissolved in 2 mL formic acid. The solution is cooled in an ice bath, 30% aqueous hydrogen peroxide (970 µL, 8.6 mmol) is added drop-wise, and the reaction is stirred 1 h in an ice bath, and allowed to stand overnight at 5°C. The mixture is concentrated to dryness, is suspended in $H_2O$, and the pH is adjusted to 7 with 2N NaOH. The mixture is concentrated to dryness, is dissolved in MeOH, and is passed over 15 mL 50W-X2 ion exchange resin (hydrogen form) eluting with 200 mL MeOH followed by 200 mL 5% $Et_3N$/MeOH. The basic wash is concentrated to dryness to afford 1-methyl-pyrrolo[2,3-c]pyridine-5-carboxylic acid (I-178-D) (78% yield). HRMS (FAB) calculated for $C_9H_8N_2O_2$+H: 177.0664, found 177.0672 (M+H).

**Intermediate D23: 3-Bromofuro[2,3-c]pyridine-5-carboxylic acid**

Furo[2,3-c]pyridin-5-ylmethyl acetate (5.17 g, 27.05 mmol) is dissolved in $CH_2Cl_2$ (130 mL), layered with saturated $NaHCO_3$ (220 mL), treated with $Br_2$ (8.36 mL, 162.3 mmol) and stirred very slowly for 4.5 h at rt. The mixture is stirred vigorously for 30 min, is diluted with $CH_2Cl_2$ (100 mL) and the layers separated. The aqueous layer is extracted with $CH_2Cl_2$ (2 x 100 mL) and the combined organics are concentrated to a small volume under a stream of nitrogen. The solution is diluted with EtOH (200 mL), treated with $K_2CO_3$ (22.13 g, 160.1 mmol) and stirred for 2.5 days at rt. The mixture is concentrated to dryness, partitioned between 50% saturated NaCl (200 mL) and $CH_2Cl_2$ (5 x 200 mL), dried ($Na_2SO_4$) and concentrated *in vacuo* to a yellow solid (6.07 g). The crude material is adsorbed onto silica gel (12 g) and chromatographed over 250 g slurry-packed silica gel, eluting with a gradient of 50% EtOAc / hexane to 100% EtOAc. The appropriate fractions are combined and concentrated *in vacuo* to afford 5.02 g (81%) of (3-bromofuro[2,3-c]pyridin-5-yl)methanol as a white solid. MS (EI) *m/z*: 227 ($M^+$).

Oxalyl chloride (1.77 mL, 20.1 mmol) is combined with $CH_2Cl_2$ (60 mL) in a dried flask under nitrogen, cooled to -78°C, treated dropwise with DMSO (2.86 mL, 40.25 mmol) and stirred for 20 min. The cooled solution is treated drop-wise with a solution of (3-bromofuro[2,3-c]pyridin-5-yl)methanol (4.0 mg, 17.5 mmol) in THF

(50 mL), stirred for 1 h, then treated drop-wise with $Et_3N$ (12.2 mL, 87.5 mmol). The mixture is stirred for 30 min at -78°C, then 30 min at 0°C. The mixture is washed with saturated $NaHCO_3$ (120 mL) and the organics dried ($K_2CO_3$) and concentrated *in vacuo* to a dark yellow solid (3.91 g). The crude material is chromatographed over 150 g slurry-packed silica gel, eluting with 30% EtOAc / hexane. The appropriate fractions are combined and concentrated *in vacuo* to afford 3.93 g (99%) of 3-bromofuro[2,3-c]pyridine-5-carbaldehyde as a white solid. MS (EI) *m/z*: 225 ($M^+$).

3-Bromofuro[2,3-c]pyridine-5-carbaldehyde (3.26 g, 14.42 mmol) is dissolved in THF (100 mL)/t-BuOH (50 mL)/$H_2O$ (50 mL), treated with a single portion of $NaOCl_2$ (4.89 g, 43.3 mmol) and $KH_2PO_4$ (3.92 g, 28.8 mmol) and stirred at rt for 18 h. The white solid is collected via filtration and the filtrate is concentrated *in vacuo* to dryness. The residue is suspended in water (25 mL), acidified to pH 2 with concentrated HCl and the resulting solid collected via filtration. The collected solids are dried in a vacuum oven at 50°C for 18 h and combined to afford 3.52g (99%) of 3-bromofuro[2,3-c]pyridine-5-carboxylic acid as a white solid. MS (EI) *m/z*: 241 ($M^+$).

**Intermediate D24: 3-Chlorofuro[2,3-c]pyridine-5-carboxylic acid**

Furo[2,3-c]pyridin-5-ylmethanol (7.70 g, 51.63 mmol) is dissolved in pyridine (45 mL), treated with acetic anhydride (14.36 mL, 154.9 mmol) and stirred for 18 h at rt. The pyridine is removed *in vacuo* and the resulting residue dissolved in EtOAc (200 mL), washed with 50% saturated sodium bicarbonate (4 x 90 mL), dried ($MgSO_4$) and concentrated *in vacuo* to afford 9.32 g (94%) of furo[2,3-c]pyridin-5-ylmethyl acetate as a yellow oil. MS (EI) *m/z*: 191 ($M^+$), 277, 148, 119, 118, 86, 84, 77, 63, 51, 50. .

Furo[2,3-c]pyridin-5-ylmethyl acetate (956 mg, 5 mmol) is dissolved in $CH_2Cl_2$ (40 mL) and cooled to 0°C. Chlorine gas is bubbled through the solution for 15 min, the cooling bath is immediately removed and the mixture stirred for 2 h. The mixture is re-cooled to 0°C, saturated with chlorine gas, the cooling bath removed and the solution warmed to rt. The solution is layered with saturated $NaHCO_3$ (20 mL), stirred gently for 2 h then stirred vigorously for 15 min. The mixture is diluted with saturated $NaHCO_3$ (50 mL), extracted with $CH_2Cl_2$ (1 x 40 mL then 1 x 20 mL), dried ($K_2CO_3$) and concentrated to a volume of 20 mL under a stream of nitrogen. The solution is diluted with EtOH (35 mL), treated with $K_2CO_3$ (4.09 g, 29.6 mmol) and

stirred for 18 h at rt. Water (7 mL) is added and the mixture stirred for 2 days. The mixture is concentrated to dryness, partitioned between 50% saturated NaCl (50 mL) and $CH_2Cl_2$ (4 x 50 mL), dried ($K_2CO_3$) and concentrated *in vacuo* to a brown solid (833 mg). The crude material is chromatographed over a standard 40 g Biotage column, eluting with 50% EtOAc / hexane. The appropriate fractions are combined and concentrated to afford 624 mg (68%) of (3-chlorofuro[2,3-c]pyridin-5-yl)methanol as a yellow oil. $^1$H NMR (DMSO-$d_6$): δ 4.69, 5.56, 7.69, 8.55, 8.93 ppm.

Oxalyl chloride (231 μL, 2.6 mmol) is combined with $CH_2Cl_2$ (10 mL), cooled to -78°C, treated dropwise with DMSO (373 μL, 5.3 mmol) and stirred for 20 min. The cooled solution is treated dropwise with a solution of (3-chlorofuro[2,3-c]pyridin-5-yl)methanol (420 mg, 2.3 mmol) in THF (5 mL) / $CH_2Cl_2$ (5 mL), stirred for 1 h, then treated dropwise with $Et_3N$ (1.59 mL, 11.45 mmol). The mixture is stirred for 30 min at -78°C, then 30 min at 0°C. The mixture is washed with saturated $NaHCO_3$ (20 mL) and the organics dried ($K_2CO_3$) and concentrated *in vacuo* to a yellow solid (410 mg). The crude material is chromatographed over 20 g slurry-packed silica gel, eluting with 15% EtOAc / hexane. The appropriate fractions are combined and concentrated *in vacuo* to afford 322 mg (77%) of 3-chlorofuro[2,3-c]pyridine-5-carbaldehyde as a white solid. $^1$H NMR ($CDCl_3$): δ 7.89, 8.33, 9.02, 10.18 ppm.

3-Chlorofuro[2,3-c]pyridine-5-carbaldehyde (317 mg, 1.74 mmol) is dissolved in THF (10 mL)/t-BuOH (5 mL)/$H_2O$ (5 mL), treated with a single portion of sodium chlorite (592 mg, 5.24 mmol) and $KH_2PO_4$ (473 mg, 3.48 mmol) and stirred at rt for 18 h. The reaction mixture is concentrated *in vacuo* to dryness, suspended in water (10 mL), acidified to pH 3.5 with concentrated HCl and stirred at rt for 2 h. The resulting solid is filtered, washed with water and dried in a vacuum oven at 40°C for 18 h to afford 364 mg of 3-chlorofuro[2,3-c]pyridine-5-carboxylic acid as a white solid. MS (EI) *m/z*: 197 ($M^+$).

**Intermediate D25: Benzothieno[3,2-c]pyridine-3-carboxylic acid**

N-butyl lithium (150.6 ml, 241 mmol) is added dropwise to ether (100 ml) at -20°C under $N_2$. 3-Bromothianaphthene (10.5 ml, 80.3 mmol) is dissolved in ether (50 ml) and also added dropwise to the chilled solution, stirring cold for 0.5 h. DMF (16.3 ml, 210 mmol) is dissolved in ether (75 ml) and added dropwise, and the solution stirred an additional 15 h at –20°C. The reaction is quenched onto ice (300 g)

in 10% $H_2SO_4$ (200 ml) and stirred until both layers turn yellow in color. The resulting slurry is filtered, and the cake is allowed to dry in the air stream, affording 1-benzothiophene-2,3-dicarbaldehyde (I-180-D) as a yellow solid (60% yield). HRMS (FAB) calculated for $C_{10}H_6O_2S$+H: 191.0167, found 191.0172 (M+H).

1-Benzothiophene-2,3-dicarbaldehyde (I-180-D) (1.91 g, 10.0 mmol) is dissolved in $CH_2Cl_2$ (100 ml) and chilled in an ice bath. Methyl (acetylamino)(dimethoxyphosphoryl) acetate (I-152-D) (2.63 g, 11.0 mmol) is dissolved in $CH_2Cl_2$ (50 ml) and added to 1,8-diazabicyclo[5.4.0]undec-7-ene (1.65 ml, 11.0 mmol), stirring for 5 minutes. This solution is added dropwise to the chilled thiophene solution. The reaction mixture is stirred in the ice bath for 1 h and then over night at rt. The reaction is concentrated *in vacuo* and the crude material is chromatographed over 500 g slurry-packed silica eluting with 50% ethyl acetate/hexane to afford methyl benzothieno[3,2-c]pyridine-3-carboxylate (I-181-D) as a white solid (73% yield). MS for $C_{13}H_9NO_2S$, (EI) *m/z*: 243 $(M)^+$.

I-181-D (1.43 g, 5.87 mmol) is dissolved in MeOH (25 ml) with $H_2O$ (3 ml). 2M NaOH (3.0 ml, 6.0 mmol) is added dropwise and the solution stirred at rt. After 4 days (complete disappearance of ester by TLC), the reaction is concentrated *in vacuo*. The residue is dissolved in $H_2O$ (5 ml) and the pH is adjusted to 3 with 10% HCl. The solution is stirred over night before precipitation is complete. The slurry is filtered and the cake is rinsed with ether, giving a 100% yield of benzothieno[3,2-c]pyridine-3-carboxylic acid (I-182-D)as a white solid. HRMS (FAB) calculated for $C_{12}H_7NO_2S$+H 230.0276, found 230.0275 (M+H).

**Intermediate D26: Thieno[3,4-c]pyridine-6-carboxylic acid**

3,4-Dibromothiophene (12.5 ml, 113 mmol) is combined with CuCN (30.4 g, 339 mmol) in DMF (40 ml) in a dry flask under nitrogen utilizing an over-head stirrer. The reaction is allowed to reflux at 180°C for 5 h. The dark mixture is then poured into a solution of $FeCl_3$ (113.6 g, 700 mmol) in 1.7M HCl (200 ml) and heated at 65°C for 0.5 h, again using the over-head stirrer. The reaction is cooled to rt and extracted with $CH_2Cl_2$ (7 x 300 ml). Each extract is washed individually with 200 ml *each* 6M HCl (2X), water, saturated $NaHCO_3$, and water. The organics are then combined, dried ($MgSO_4$), filtered, and concentrated, affording 10.49 g (69%) of 3,4-

dicyanothiophene as a fluffy tan solid. HRMS (EI) calcd for $C_6H_2N_2S$: 133.9939, found 133.9929 ($M^+$).

3,4-Dicyanothiophene (5.0 g, 37.2 mmol) is suspended in benzene (150 ml) in a dry flask under nitrogen utilizing an over-head stirrer. Diisobutyl aluminum hydride (1.0M in toluene) (82.0 ml, 82.0 mmol) is added dropwise, and the reaction stirred at rt for 2 h. The reaction is then carefully quenched with MeOH (5 ml) and poured onto 30% $H_2SO_4$ (60 ml) with ice (200 g). The slurry is stirred until all lumps are dissolved, and the layers are allowed to separate. The aqueous layer is extracted with $Et_2O$ (4 x 200 ml), and the combined organics are dried ($MgSO_4$), filtered, and adsorbed onto silica. The crude material is chromatographed over 225 g slurry-packed silica, eluting with 40% EtOAc/hexane. The appropriate fractions are combined and concentrated to afford 1.88 g (36%) of 3,4-thiophene dicarboxaldehyde as a pale yellow solid. MS (EI) m/z: 140 ($M^+$).

3,4-Thiophene dicarboxaldehyde (1.0 g, 7.13 mmol) is dissolved in $CH_2Cl_2$ (40 ml) and chilled to 0°C. Methyl (acetylamino)(dimethoxyphosphoryl)acetate (1.88 g, 7.85 mmol) is dissolved in $CH_2Cl_2$ (30 ml) and combined with DBU (1.1 ml, 7.85 mmol). This solution is added dropwise to the chilled thiophene solution after stirring for 5 min. The reaction mixture is stirred at 0°C for 1 h and then overnight at rt. The volatiles are removed *in vacuo* and the crude material is chromatographed over 68 g slurry-packed silica eluting with 70% EtOAc/hexane. The appropriate fractions are combined and concentrated to yield 2.09 g of the carbinol intermediate as a white foam. The intermediate is dissolved in $CHCl_3$ (50 ml) and treated with DBU (1.32 ml, 8.8 mmol) and trifluoracetic anhydride (1.24 ml, 8.8 mmol) in a drop-wise fashion. The reaction is stirred overnight at rt and is then quenched with saturated $NaHCO_3$ solution (50ml). The layers are separated, and the aqueous layer is extracted with $CHCl_3$ (2 x 50 ml). The combined organics are dried ($MgSO_4$), filtered, and concentrated to a yellow oil. This oil is chromatographed over 50 g slurry-packed silica, eluting with 90% EtOAc/hexane. The appropriate fractions are combined and concentrated to afford 1.2 g (88%) of methyl thieno[3,4-c]pyridine-6-carboxylate as a yellow solid. MS (EI) *m/z*: 193 ($M^+$).

Methyl thieno[3,4-c]pyridine-6-carboxylate (250 mg, 1.3 mmol) is dissolved in MeOH (7 ml) and water (1 ml). 2M NaOH (0.72 ml, 1.43 mmol) is added drop-wise. The reaction is stirred overnight at rt and is monitored by TLC. The volatiles

are removed *in vacuo* and the residue is dissolved in water (2 ml). 10% HCl is used to adjust the pH to 3, and the reaction again stirred overnight at rt. The aqueous solution is extracted repeatedly with EtOAc (20 x 10 ml). The combined organics are dried ($MgSO_4$), filtered, and concentrated to a yellow solid. The amount of isolated product via extraction is minimal (67 mg), so the aqueous layer is concentrated and found to contain the majority of product. Extraction of the solid aqueous residue with EtOAc provided 225 mg (97%) of thieno[3,4-c]pyridine-6-carboxylic acid as a yellow solid. MS (EI) *m/z*: 179 ($M^+$).

**Intermediate D27: Benzofuran-5-carboxylic acid**

1-(2,3-Dihydrobenzofuran-5-yl)ethanone is made using a procedure, making non-critical changes, as described in Dunn, J.P.; Ackerman, N.A.; Tomolois, A.J. *J. Med. Chem.* **1986,** *29*, 2326. Similar yield (82%) and similar purity (95%) are obtained. $^1$H NMR (400 MHz, $CDCl_3$) δ 7.89, 7.83, 6.84, 4.70, 3.29 , 2.58.

A mixture of 1-(2,3-dihydrobenzofuran-5-yl)ethanone (4.0 g, 25 mmol) and sodium hypochlorite [160 mL of a 6.0% aqueous solution, (Clorox brand of bleach)] at 55°C is stirred for 1 h. The mixture (now homogeneous) is cooled to rt and solid sodium bisulfite is added until a clear color persists. Hydrochloric acid (80 mL of a 1.0 N aqueous solution) is added, followed by extraction with EtOAc. The organic layer is washed with brine, dried ($MgSO_4$), filtered, and concentrated *in vacuo* to afford 3.93 g (97%) of 2,3-dihydrobenzofuran-5-carboxylic acid as a white solid. $^1$H NMR (400 MHz, $CDCl_3$) δ 11.0–10.3, 8.00, 6.87, 4.72, 3.31.

To a stirred solution of 2,3-dihydrobenzofuran-5-carboxylic acid (3.96 g, 24.1 mmol) in MeOH (200 mL) is added concentrated sulfuric acid (0.5 mL). The mixture is heated to reflux for 24 h. The mixture is cooled to rt, followed by the addition of solid sodium bicarbonate. The reaction mixture is concentrated *in vacuo*, and the remaining residue is partitioned between EtOAc and water. The aqueous layer is extracted with EtOAc, and the combined organic layers are dried ($MgSO_4$), filtered and concentrated *in vacuo* to afford 4.22 g (98%) of methyl 2,3-dihydrobenzofuran-5-carboxylate as a white solid. $^1$H NMR (400 MHz, $CDCl_3$) δ 7.93-7.89, 6.82, 4.69, 3.86, 3.28.

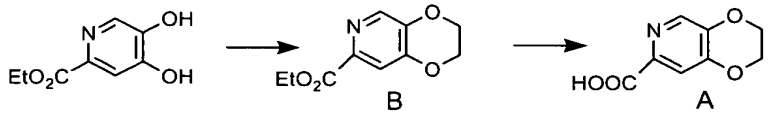
To a stirred solution of methyl 2,3-dihydrobenzofuran-5-carboxylate (4.2 g, 24 mmol) in anhydrous *p*-dioxane (150 mL) under argon atmosphere is added 2,3-

dichloro-5,6-dicyano-1,4-benzoquinone (6.42 g, 28 mmol). The mixture is heated to reflux for 24 h, followed by cooling to rt. The reaction mixture is partitioned between ether and ½ saturated aqueous sodium carbonate solution. The organic layer is extracted several times with ½ saturated aqueous sodium carbonate solution. The organic layer is washed with water, dried ($MgSO_4$), filtered, and concentrated *in vacuo* to give a mixture (92%) of recovered starting material methyl 2,3-dihydrobenzofuran-5-carboxylate and methyl benzofuran-5-carboxylate in a ratio of 1:3. The crude product is purified by preparative HPLC using a Chiralcel OJ column. Elution with heptane-*iso*-propyl alcohol, (80:20, flow rate = 70 mL/min) gives 0.75 g (18%) of methyl 2,3-dihydrobenzofuran-5-carboxylate as a white solid and 2.5 g (61%) of methyl benzofuran-5-carboxylate as a white solid. $^1$H NMR for methyl benzofuran-5-carboxylate (400 MHz, $CDCl_3$) δ 8.40, 8.07, 7.73, 7.57, 6.89, 3.99.

A stirred mixture of methyl benzofuran-5-carboxylate (1.3 g, 7.38 mmol) in MeOH (51 mL) and sodium hydroxide (41 mL of a 5 % aqueous solution) is heated to 65°C for 4 h. The mixture is cooled to rt, and MeOH was removed *in vacuo*. The remaining aqueous layer is extracted with $CH_2Cl_2$. The $CH_2Cl_2$ layer is discarded, and the aqueous layer is acidified to pH=1 with concentrated hydrochloric acid. The aqueous layer is extracted with $CHCl_3$. The organic layer is washed with water, dried ($MgSO_4$), filtered and concentrated *in vacuo* to afford 1.2 g (98%) of benzofuran-5-carboxylic acid as a white solid. $^1$H NMR (400 MHz, DMSO-$d_6$) δ 12.9, 8.30, 8.11, 7.92, 7.69, 7.09.

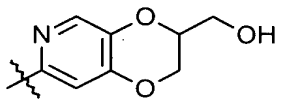
Compounds of Formula I where W is (E) are made using the coupling procedures discussed herein and in cited references, making non-critical changes to obtain the desired compounds. The following intermediates to provide W of formula I are for exemplification only and are not intended to limit the scope of the present invention. Other intermediates within the scope of the present invention can be obtained using known procedures or by making slight modifications to known procedures.

It will be apparent to those skilled in the art that the requisite carboxylic acids can be obtained through synthesis via literature procedures or through the slight modification thereof. For example, compounds of Formula I where $E^0$ is N and $E^1$ and $E^2$ are O, can be obtained as follows:



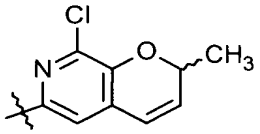

Acid A can be prepared from ethyl 4,5-dihydroxypyridine-2-carboxylate (see *Z. Naturfirsch*, **34b**, 1729-1736, 1979). Alkylation with 1,2-dibromoethane gives B. Saponification of B with aqueous NaOH would provide the requisite carboxylic acid A. The resulting acid is coupled with an Azabicyclo using conditions described herein.

Substituents can be introduced for $R_{E-1}$ or $R_{E-2}$ where $E^0$ is CH and $E^1$ and $E^2$ are each Oais described in Taniguchi, Eiji, et al., *Biosci. Biotech. Biochem.*, **56** (4), 630-635, 1992. See also Henning, R.; Lattrell, R.; Gerhards, H. J.; Leven, M.; *J.Med.Chem.*; 30; 5; 1987; 814-819. This is also applicable to make the final compounds where $E^0$ is N, starting with ethyl 4,5-dihydroxypyridine-2-carboxylate to obtain the ester intermediate which could be saponified:

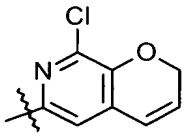



Furthermore, where $E^0$ is N, the compounds where one $R_{E-1}$ is a bond to $CR_{E-1-1}$ or where one $R_{E-2}$ is a bond to $CR_{E-2-2}$, the compounds can be obtained using methods described herein for $E^0$ is CH, making non-critical changes. Moreover, where at least one $R_{E-1}$ and/or at least one $R_{E-2}$ is other than H and is not a bond, the compounds can be obtained using methods described herein for where $E^0$ is CH.

Compounds where $E^0$ is N, only one of $E^1$ or $E^2$ is O, $R_{E-0}$ is other than H, and one of $R_{E-1}$ or $R_{E-2}$ is a bond, can be obtained as discussed herein using procedures for where $E^0$ is CH. For example, 2-chloro-6-(hydroxymethyl)-4-vinylpyridin-3-ol could be converted into (8-chloro-2-methyl-2*H*-pyrano[2,3-c]pyridin-6-yl)methanol using the procedures discussed herein. The alcohol could be oxidized to the corresponding carboxylic acid:




Similarly, (8-chloro-2*H*-pyrano[2,3-c]pyridin-6-yl)methanol can be oxidized to give 8-chloro-2*H*-pyrano[2,3-c]pyridin-6-carboxylic acid:




Some specific examples are provided for exemplification and are not intended to limit the scope of the present invention:

**Intermediate E1: 2,3-Dihydro-1,4-benzodioxine-6-carboxylic acid**

A suspension of calcium ethoxide (816mg, 6.3mmol), butene oxide (5.2mL, 93mmol) and 2,4-diiodophenol (2.17g, 6.3mmol) is heated in a sealed flask at 80°C for 18 h. The reaction mixture is allowed to cool, poured into 1N HCl and extracted three times with $CH_2Cl_2$. The combined organic extracts are dried ($Na_2SO_4$), filtered and concentrated *in vacuo*. The resulting material is purified by column chromatography (two columns, step gradient of 30-40-50% $CH_2Cl_2$ in hexanes) to give 1-(2,4-diiodophenoxy)butan-2-ol as a clear oil (1.73g, 67%).$^1$H NMR (400 MHz, $CDCl_3$) δ 8.04, 7.56, 6.57, 4.03, 3.9, 3.84, 2.42, 1.65, 1.04.

A solution of 1-(2,4-diiodophenoxy)butan-2-ol (1.27g, 3.0) in pyridine (12mL) is degassed by repeatedly evacuating the flask then filling with $N_2$. Sodium hydride (60% suspension, 153mg, 3.8mmol) is added and the resulting mixture is stirred for 15 min. Copper (I) chloride (15mg, 0.15mmol) is added, and the resulting mixture is heated at 80°C for 2 h. The reaction is allowed to cool, poured into 1M HCl and extracted three times with $CH_2Cl_2$. The combined organic extracts are dried ($Na_2SO_4$), filtered and concentrated *in vacuo*. The resulting material is purified by column chromatography (10% $CH_2Cl_2$ in hexanes) to give 2-ethyl-7-iodo-2,3-dihydro-1,4-benzodioxine as a clear oil (493mg, 57%). $^1$H NMR (400 MHz, $CDCl_3$) δ 7.20, 7.10, 6.61, 4.22, 4.01, 3.85, 1.7, 1.6, 1.06 .

A solution of 2-ethyl-7-iodo-2,3-dihydro-1,4-benzodioxine (486mg, 1.68mmol) in DMF (3mL) is degassed by repeatedly evacuating the flask and filling with $N_2$. $Zn(CN)_2$ (117mg, 1.0mmol), and $Pd(PPh_3)_4$ (97mg, 0.084mmol) are added, and the resulting solution is degassed, and is then heated to 80°C for 1.5 h. The reaction is allowed to cool, poured into water and extracted two times with ether. The combined organic extracts are dried ($Na_2SO_4$), filtered and concentrated *in vacuo*. The resulting material is purified by column chromatography (step gradient, 25-50% $CH_2Cl_2$ in hexanes) to give 3-ethyl-2,3-dihydro-1,4-benzodioxine-6-carbonitrile as a

clear oil (296mg, 92%). $^1$H NMR (400 MHz, $CDCl_3$) δ 7.16, 7.13, 6.91, 4.31, 4.05, 3.93, 1.7, 1.6, 1.08.

KOH (218mg, 3.9mmol) is added to a mixture of 3-ethyl-2,3-dihydro-1,4-benzodioxine-6-carbonitrile (247mg, 1.3mmol), ethanol (3mL) and water (1mL). The resulting mixture is heated to 80°C for 24 hours. The reaction is allowed to cool, diluted with water (2mL) and acidified to pH<2 with concentrated HCl. The resulting solid is filtered, washed with water and dried at 60°C under vacuum to give 3-ethyl-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid as a white solid (249mg, 92%). $^1$H NMR (400 MHz, DMSO-$d_6$) δ 12.66, 7.43, 7.37, 6.95, 4.38, 4.10, 3.95, 1.64, 1.01.

**Intermediate E2: 2-(Phenoxymethyl)-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid**

6-Bromo-2,3-dihydro-1,4-benzodioxin-2-yl)methanol is prepared according to literature reports for 6-fluoro-2,3-dihydro-benzo-1,4-dioxin-2-yl)-methanol. See Henning, R.; Lattrell, R.; Gerhards, H. J.; Leven, M.; *J.Med.Chem.*; 30; 5; 1987; 814-819. The intermediate is obtained in 70% yield as a solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.08, 7.00, 6.81, 4.25-4.40, 4.10-4.20, 3.85-4.00, 1.95; MS (EI) *m/z* 244 ($M^+$).

A mixture of (6-bromo-2,3-dihydro-1,4-benzodioxin-2-yl)methanol (3.94 g, 16.1 mmol) and DMF (35 mL) at rt is treated with a 60% dispersion of NaH in mineral oil (0.706 g, 17.7 mmol). After 15 min, the mixture is treated with benzyl bromide (2.10 mL, 17.7 mmol). After 2 h, the mixture is poured into $H_2O$ and extracted with EtOAc (2 x 125 mL). The combined organics are washed with $H_2O$ (3 x 100 mL), brine, dried ($MgSO_4$), filtered, and concentrated. The resulting oil is adsorbed onto $SiO_2$ and chromatographed (Biotage 40M + SIM, 5% EtOAc/Hexane). The product fractions are pooled and concentrated to give an oil which solidified (upon standing) to give 3.91 g (73%) of 2-[(benzyloxy)methyl]-6-bromo-2,3-dihydro-1,4-benzodioxine: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.30-7.45, 7.06, 6.99, 6.81, 4.60-4.70, 4.30-4.40, 4.05-4.15, 3.65-3.85; MS (EI) *m/z* 244 ($M^+$).

A mixture of 2-[(benzyloxy)methyl]-6-bromo-2,3-dihydro-1,4-benzodioxine (3.63 g, 10.8 mmol) in THF (60, mL) is cooled in a $CO_2$/acetone bath under $N_2$. A solution of *t*-butyl lithium in pentane (1.3 M, 17.5 mL, 22.8 mmol) is added. After 5 min, $CO_2$ (g) is bubbled through the mixture and the mixture is warmed to rt. A

solution of HCl in methanol is added and the mixture concentrated. The residue is extracted between NaOH (1 N) and EtOAc. The organic layer is discarded. The pH of the aqueous layer is adjusted to ~ 4 and is extracted with EtOAc (2 x 100 mL). The combined organics are washed with $H_2O$ (3 x 100 mL), brine, dried ($MgSO_4$), filtered, and concentrated. The resulting oil is chromatographed (Biotage 40M, 2% $MeOH/CH_2Cl_2$). The product fractions are pooled and concentrated to an give oil 1.66 g (51%) of 2-(phenoxymethyl)-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid.

**Intermediate E3: 3-[(Benzyloxy)methyl]-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid**

(*R*) and (*S*)-(7-Bromo-2,3-dihydro-benzo-1,4-dioxin-2-yl)-methanol are prepared according to the literature example. The racemic mixture is obtained starting with racemic epichlorohydrin. See Aiba, Y.; Hasegawa, et al., Bioorg.Med.Chem.Lett.; 11; 20; 2001; 2783-2786.

A mixture of 7-bromo-2,3-dihydro-1,4-benzodioxin-2-yl)methanol (2.73 g, 11.1 mmol) and DMF (25 mL) at 0°C is treated with a 60% dispersion of NaH in mineral oil (0.49 g, 12.3 mmol). After 15 min, the mixture is treated with benzyl bromide (1.46 mL, 12.37 mmol). After 2 h, the mixture is poured into $H_2O$ and extracted with EtOAc (2 x 125 mL). The combined organic layers are washed with $H_2O$ (3 x 100 mL), brine, dried ($MgSO_4$), filtered, and concentrated. The resulting oil is adsorbed onto $SiO_2$ and chromatographed (Biotage 40M + SIM, 5% EtOAc/Hexane). The product fractions are pooled and concentrated to provide an oil, which solidified (upon standing) to give 3.48 g (93%) of 2-[(benzyloxy)methyl]-7-bromo-2,3-dihydro-1,4-benzodioxine.

A mixture of 2-[(benzyloxy)methyl]-7-bromo-2,3-dihydro-1,4-benzodioxine (3.35 g, 10.0 mmol) in THF (60, mL) is cooled in a $CO_2$/acetone bath under $N_2$. A solution of *t*-butyl lithium in pentane (1.7 M, 6.0 mL, 10.2 mmol) is added. After 5 min, $CO_2$ (g) is bubbled through the mixture and the mixture is warmed to rt. A solution of HCl in methanol is added and the mixture concentrated. The residue is chromatographed (Biotage 40M, 3% $MeOH/CH_2Cl_2$). The product fractions are pooled and concentrated to give 1.19 g (40%) of 3-[(benzyloxy)methyl]-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid as an oil.

**Intermediate E4: (3*S*)-3-[(Benzyloxy)methyl]-2,3-dihydro-1,4-benzodioxine-6-carboxyl acid**

Intermediate E4 is obtained following the procedures discussed for Intermediate E3, making non-critical changes, and starting with [(2*S*)-7-bromo-2,3-dihydro-1,4-benzodioxin-2-yl]methanol

**Intermediate E5: (3*R*) 3-[(Benzyloxy)methyl]-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid**

Intermediate E5 is obtained following the procedures discussed for Intermediate E3, making non-critical changes, and starting with (3*R*)-3-[(benzyloxy)methyl]-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid.

**Intermediate E6: (3S)-3-(Phenoxymethyl)-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid**

A mixture of [(2*S*)-7-bromo-2,3-dihydro-1,4-benzodioxin-2-yl]methanol (2.26 g, 9.20 mmol), phenol (0.87 g, 9.2 mmol), triphenylphosphine (2.42 g, 9.20 mmol) and THF (80 mL) is cooled in a 0°C bath under $N_2$. Diethylazodicarboxylate (1.50 ml, 9.5 mmol) is added, and the mixture is allowed to warm to rt overnight. The mixture is adsorbed onto $SiO_2$ and chromatographed (Biotage 40S+SIM, (1:19) EtOAc:hexane). The product fractions are pooled and concentrated to afford 1.45 g (49%) of (2S)-7-bromo-2-(phenoxymethyl)-2,3-dihydro-1,4-benzodioxine as a clear oil.

**Intermediate E7: (3*R*)-3-(Phenoxymethyl)-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid**

A mixture of [(2R)-7-bromo-2,3-dihydro-1,4-benzodioxin-2-yl]methanol (0.648 g, 2.64 mmol), phenol (0.248 g, 2.64 mmol), triphenylphosphine (0.692 g, 2.64 mmol) and THF (26 mL) is cooled in a 0°C bath under $N_2$. Diethylazodicarboxylate (0.42 ml, 2.7 mmol) is added and the mixture allowed to warm to rt overnight. The mixture is concentrated, partitioned between EtOAc and $H_2O$, the organic layer dried ($MgSO_4$), adsorbed onto $SiO_2$, and chromatographed (Biotage 40S+SIM, (1:19) EtOAc:hexane). The product fractions are pooled and concentrated to afford 0.315 g, (37%) of (2R)-7-bromo-2-(phenoxymethyl)-2,3-dihydro-1,4-benzodioxine as an oil.

A solution of this oil (0.280 g, 0.87 mmol) and THF (30 ml) is cooled in a $CO_2$ (s)/acetone bath under $N_2$. To this is added a solution of *tert*-butyl lithium in pentane (1.7 M, 1.10 ml, 1.9 mmol). After stirring for 5 min, $CO_2$ (g) is bubbled through the solution for an additional 10 min. The mixture is treated with MeOH/HCl and allowed to warm to rt. The mixture is concentrated, and the residue is chromatographed (Biotage 40S, (1:499) MeOH:$CH_2Cl_2$). The product fractions are pooled and concentrated to afford 0.103 g (41%) of (3*R*)-3-(phenoxymethyl)-2,3-dihydro-1,4-benzodioxine-6-carboxylic acid as a solid.

**Intermediate E8: 2,3-Dihydro-1,4-dioxino[2,3-c]pyridine-7-carboxylic acid**

To a stirred solution of 4,5-hydroxypyridine-2-carboxylic acid [see:Kenichi Mochida, *et al. J. Antibiot.* **1987,** 182] (800 mg, 4.18 mmol) in MeOH (30 mL) is added concentrated sulfuric acid (1 mL). The mixture is heated to reflux for 2 days. The mixture is cooled to rt, followed by addition of solid sodium bicarbonate. The mixture is diluted with water and the precipitate is filtered and dried to give 527 mg (75%) of methyl 4,5-dihydroxypyridine-2-carboxylate: $^1$H NMR (400 MHz, MeOH-$d_4$) δ 7.68, 7.24, 3.97.

To a stirred solution of methyl 4,5-dihydroxypyridine-2-carboxylate (348 mg, 2.06 mmol) in DMF (20 mL) is added solid $K_2CO_3$ (3.1 g, 22 mmol) and 1,2-dibromoethane (386 μL, 4.5 mmol). The mixture is heated at 115°C for 2 h. DMF is removed *in vacuo*, the residue is partitioned between water and EtOAc. The aqueous layer is again extracted with EtOAc. The combined organic layers are dried ($MgSO_4$) and concentrated *in vacuo* to give a yellow solid for methyl 2,3-dihydro-1,4-dioxino[2,3-c]pyridine-7-carboxylate (348 mg, 86%): $^1$H NMR (400 MHz, $CDCl_3$) δ 8.29, 7.71, 4.39, 3.99.

To a stirred solution of methyl 2,3-dihydro-1,4-dioxino[2,3-c]pyridine-7-carboxylate (300 mg, 1.54 mmol) in MeOH (10 mL) is added NaOH (10 mL of a 5% aqueous solution). The mixture is heated to reflux for 3 h, followed by cooling to rt. The methanol is removed *in vacuo* and the remaining aqueous layer is acidified to pH=5 with 1N HCl, extracted with $CH_2Cl_2$ continuously for 2 days. The organic layer is concentrated to a white solid (245 mg, 88%) for 2,3-dihydro-1,4-dioxino[2,3-c]pyridine-7-carboxylic acid: $^1$H NMR (400 MHz, DMSO-$d_6$) δ 13-12, 8.21, 7.52, 4.39.

**Intermediate E9: Chromane-6-carboxylic acid**

A mixture of chromene (see: Chatterjea, *J. Indian Chem. Soc.* **1959,** *35*, 78.) (5.00 g, 37.8 mmol) and 10% palladium on activated carbon (250 mg) in glacial acetic acid (100 mL) is placed in a Parr bottle. The mixture is shaken under an atmosphere of hydrogen (45 psi) for 3 h at rt. The mixture is filtered through Celite and the filtrate is concentrated *in vacuo* to afford 5.00 g (98%) of chromane as light yellow oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.15-7.05, 6.89, 6.80, 4.23, 2.84, 2.08-2.02.

To a stirred solution of acetyl chloride (4.78 mL, 67.1 mmol) in dry $CH_2Cl_2$ (20 mL) in a –10°C bath is added aluminum trichloride (4.76 g, 35.7 mmol) in small portions. The mixture is stirred for 15 min until the solution became homogeneous. The solution is added via canula to a separate solution of chromane (4,79 g, 35.7 mmol) in $CH_2Cl_2$ (30 mL) all at –10 °C. After complete addition, the solution is stirred at –10°C for 30 min. The solution is poured over a mixture of crushed ice and concentrated HCl. The mixture is extracted with $CH_2Cl_2$. The combined organic layers are washed with brine, dried ($MgSO_4$), filtered and concentrated *in vacuo*. The remaining residue is purified via crystallization from hexanes to give 4.0 g (64%) of 1-(3,4-dihydro-2H-chromen-6-yl)ethanone as a white solid. $^1$H NMR (400 MHz, $CDCl_3$) δ7.76-7.73, 6.75, 4.27, 2.86, 2.57, 2.09-2.03.

A mixture of 1-(3,4-dihydro-2H-chromen-6-yl)ethanone (3.80 g, 22.0 mmol) and sodium hypochlorite [150 mL of a 6.0% aqueous solution, (Clorox brand of bleach)] in a 55°C oil bath is stirred for 2 h. The mixture (now homogeneous) is cooled to rt and solid sodium bisulfite is added until a clear color persisted. HCl (ca 15 mL of a 6.0 M aqueous solution) is added, followed by extraction with EtOAc. The organic layer is washed with brine, dried ($MgSO_4$), filtered, and concentrated *in vacuo* to afford 3.10 g (82%) of chromane-6-carboxylic acid as a white solid. $^1$H NMR (400 MHz, DMSO-$d_6$) δ 12.55, 7.67, 7.6, 6.79, 4.20, 2.77, 1.96-1.90.

**Intermediate E10: Chromane-7-carboxylic acid**

To a stirred solution of methyl 4-formyl-3-hydroxybenzoate [see: Harayama, *Chem. Pharm. Bull.* **1994,** 2170] (0.8 g, 4.1 mmol) and anhydrous $K_2CO_3$ (1.1 g, 8.0 mmol) in acetone (12 mL) is added allyl bromide (0.70 mL, 8.1 mmol). The mixture is heated in a 48°C oil bath for 2 h. The reaction mixture is cooled to rt and filtered.

The mother liquor is concentrated *in vacuo* to a brown oil. The crude product is purified by flash chromatography on $SiO_2$. Elution with hexanes-EtOAc (85:15) gives 0.85 g (49%) of methyl 3-(allyloxy)-4-formylbenzoate as a clear solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 10.6, 7.9, 7.7, 6.1, 5.5, 5.4, 4.8, 4.0.

Sodium hydride [220 mg (60% oil dispersion), 5.4 mmol], is washed with pentane (3x) and is suspended in THF (12 mL) in a 0°C ice bath. Methyl triphenylphosphonium bromide (1.7 g, 4.7 mmol) is added. The suspension is allowed to warm to rt and stir for 30 min. A solution of methyl 3-(allyloxy)-4-formylbenzoate (0.85 g, 3.8 mmol) in THF (5 mL) is added via canula. The mixture is stirred at rt for 2 h. The mixture is diluted with EtOAc and washed with brine. The organic layer is dried with $MgSO_4$, filtered and concentrated *in vacuo* to a yellow residue. The crude product is triturated with hexanes, filtered and dried *in vacuo* to a clear oil for methyl 3-(allyloxy)-4-vinylbenzoate (680 mg, 81%): $^1$H NMR (400 MHz, $CDCl_3$) δ 7.65-7.54, 7.13, 6.13, 5.88, 5.49-5.29, 4.65, 3.93.

To a stirred solution of methyl 3-(allyloxy)-4-vinylbenzoate (0.67 g, 3.1 mmol) in $CH_2Cl_2$ (20 mL) at rt is added benzylidene-bis(tricyclohexylphosphine)-dichlororuthenium (63 mg, 0.076 mmol). The mixture is stirred at rt for 2 h. The reaction mixture is concentrated *in vacuo* to a dark residue. The crude product is purified by flash chromatography on $SiO_2$. Elution with hexanes-EtOAc (95:5) gives 372 mg (64%) of methyl 2H-chromene-7-carboxylate as a clear oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.56, 7.46, 7.01, 6.46, 5.91, 4.89, 3.91.

A mixture of methyl 2H-chromene-7-carboxylate (372 mg, 1.96 mmol) and 10% Pd/C (25 mg) in methanol (15 mL) is stirred under 1 atm of hydrogen at rt for 3 h. The mixture is filtered through Celite and the filtrate is concentrated to a yellow residue. The crude product is purified by flash chromatography on $SiO_2$. Elution with hexanes-EtOAc (95:5) gives 140 mg (37%) of methyl chromane-7-carboxylate as a clear oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.51, 7.47, 7.10, 4.23, 3.91, 2.85, 2.04.

To a stirred solution of methyl chromane-7-carboxylate (140 mg, 0.73 mmol) in MeOH (5 mL) is added NaOH (5 mL of a 5% aqueous solution). The mixture is heated in a 85°C oil bath for 3 h, followed by cooling to rt. The methanol is removed *in vacuo* and the remaining aqueous layer is acidified to pH=1 with concentrated HCl, extracted with EtOAc (3X). The combined organic layers are dried ($MgSO_4$) and

concentrated to a white solid for chromane-7-carboxylic acid (130 mg, 100%): $^1$H NMR (400 MHz, DMSO-$d_6$) δ 13-12, 7.37, 7.24, 7.16, 4.16, 2.79, 1.92.

**Intermediate E11: 2*H*-chromene-6-carboxylic acid**

To a stirred solution of ethyl 3-formyl-4-hydroxybenzoate [see: Skattebol, *Acta. Chemica. Scandinavica* **1999,** *53*, 258] (1.9 g, 10.0 mmol) and anhydrous $K_2CO_3$ (2.7 g, 19.5 mmol) in acetone (30 mL) is added allyl bromide (1.7 mL, 19.8 mmol). The mixture is heated in a 60°C oil bath for 2 h. The mixture is cooled to rt, filtered and concentrated *in vacuo* to afford 2.1 g (92%) of ethyl 4-(allyloxy)-3-formylbenzoate as a white solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 10.5, 8.5, 8.2, 7.1, 6.1, 5.5, 5.4, 4.8, 4.4, 1.4.

To a stirred suspension of sodium hydride [588 mg (60% oil dispersion), 15 mmol), which had been previously washed with pentane (3x), in THF (30 mL) in a 0°C ice bath is added methyl triphenylphosphonium bromide (4.6 g, 13 mmol). The suspension is allowed to warm to rt and stir for 30 min. A solution of ethyl 4-(allyloxy)-3-formylbenzoate (2.3 g, 9.8 mmol) in THF (10 mL) is added via canula. The mixture is stirred at rt 2 h. The mixture is diluted with EtOAc and washed with brine. The organic layer is dried of $MgSO_4$, filtered and concentrated *in vacuo* to a yellow residue. The crude product is purified by flash chromatography on $SiO_2$. Elution with hexanes-EtOAc (95:5) gives 1.8 g (79%) of ethyl 4-(allyloxy)-3-vinylbenzoate as a clear oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 8.2, 7.9, 7.1, 6.9, 6.1, 5.9, 5.5, 5.3, 4.7, 4.4, 1.4.

To a stirred solution of ethyl 4-(allyloxy)-3-vinylbenzoate (1.8 g, 7.7 mmol) in $CH_2Cl_2$ (40 mL) at rt is added benzylidene-bis(tricyclohexylphosphine)-dichlororuthenium (127 mg, 0.15 mmol). The mixture is stirred at rt for 2.5 h. The reaction mixture is concentrated *in vacuo* to a dark residue. The crude product is purified by flash chromatography on $SiO_2$. Elution with hexanes-EtOAc (95:5) gives 1.3 g (80%) of ethyl 2H-chromene-6-carboxylate as a clear oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.8, 7.7, 6.8, 6.4, 5.8, 4.9, 4.4, 1.4.

To a stirred solution of ethyl 2H-chromene-6-carboxylate in MeOH (80 mL) is added NaOH (40 mL of a 5% aqueous solution). The mixture is heated in a 60°C oil bath for 30 min, followed by cooling to rt. The methanol is removed *in vacuo* and the remaining aqueous layer is acidified to pH=1 with concentrated HCl. The solid

precipitate is filtered and washed with water to afford 130 mg (13%) of 2*H*-chromene-6-carboxylic acid as a white solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 12-11, 7.9, 7.7, 6.8, 6.5, 5.8, 5.0.

**Intermediate E12: 2-Methyl-2*H*-chromene-6-carboxylic acid**

To a stirred solution of lithium bis(trimethylsilyl)amide (1.0 M solution in tetrahydrofuran) (8 mL) in a 0°C ice bath is added methyl triphenylphonium bromide (1.92 g, 5.38 mmol). The mixture is allowed to warm to rt and stir for 10 min. A solution of methyl 3-formyl-4-hydroxybenzoate (200 mg, 1.11 mmol) in THF (3 mL) is added to the above solution. The mixture is stirred at rt for 5 h. The reaction mixture is acidified to pH=5 with 1N HCl, and extracted with ether (3X). The combined organic layers are washed with brine, dried ($MgSO_4$), filtered and concentrated to a yellow oil. The crude product is purified by chromatography on $SiO_2$. Elution with hexanes-EtOAc (80:20) gives 130 mg (66%) of methyl 4-hydroxy-3-vinylbenzoate as a white solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 8.12, 7.86, 6.93, 6.85, 5.84, 5.50, 5.46, 3.92.

To a stirred solution of methyl 4-hydroxy-3-vinylbenzoate (410 mg, 2.3 mmol), triphenylphosphine (787 mg, 3.0 mmol), 3-buten-2-ol (260 μL, 3.0 mmol) in THF (15 mL) at 0°C is added a solution of diethyl azadicarboxylate (472 μL, 3.0 mmol) in THF (5 mL). The mixture is allowed to warm to rt and stir overnight. The mixture is concentrated *in vacuo* and the residue is purified by chromatography on $SiO_2$. Elution with hexanes-EtOAc (95:5) gives 371 mg (69%) of methyl 3-formyl-4-[(1-methylprop-2-enyl)oxy]benzoate as a clear oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 8.18, 7.89, 7.08, 6.90, 5.94, 5.86, 5.36-5.30, 4.93, 3.91, 1.51.

To a stirred solution of methyl 3-formyl-4-[(1-methylprop-2-enyl)oxy]-benzoate (370 mg, 1.59 mmol) in $CH_2Cl_2$ (8 mL) at rt is added benzylidene-bis(tricyclohexylphosphine)dichlororuthenium (56 mg, 0.068 mmol). The mixture is stirred at rt overnight. The reaction mixture is concentrated *in vacuo* to a dark residue. The crude product is purified by flash chromatography on $SiO_2$. Elution with hexanes-EtOAc (95:5) gives 225 mg (69%) of methyl 2-methyl-2H-chromene-6-carboxylate as a clear oil: $^1$H NMR (400 MHz, $CDCl_3$) δ 7.82, 7.68, 6.79, 6.41, 5.71, 5.11, 3.89, 1.48.

To a stirred solution of methyl 2-methyl-2H-chromene-6-carboxylate (225 mg, 1.10 mmol) in MeOH (5 mL) is added NaOH (5 mL of a 5% aqueous solution). The mixture is heated in a 60°C oil bath for 40 min, followed by cooling to rt. The methanol is removed *in vacuo* and the remaining aqueous layer is acidified to pH=5 with 1N HCl. The solution is extracted with EtOAc (2X), washed with brine, dried ($MgSO_4$) and concentrated *in vacuo* to afford 209 mg (100%) of 2-methyl-2H-chromene-6-carboxylic acid as a yellow oil: $^1$H NMR (400 MHz, DMSO-$d_6$) δ 13-12, 7.68, 7.65, 6.80, 6.53, 5.85, 5.10, 1.37.

**Intermediate E13: 3,4-Dihydro-2*H*-pyrano[2,3-c]pyridine-6-carboxylic acid**

2-Chloro-3-pyridinol (20.0 g, 0.154 mole and $NaHCO_3$ (19.5g, 0.232 mole, 1.5 equ) are dissolved in 150 ml of water. The reaction mixture is placed in an oil bath at 90°C and after 5 min is treated with 37% aqueous formaldehyde (40.5 ml, 0.541 mole, 3.5 equ) which is added in six unequal doses; 12 ml initially, 3 x 8 ml followed by 1 x 2.2 ml all at 90 min intervals with the final 2.3 ml added after maintaining at 90°C overnight (15 h). After stirring in the 90°C bath for an additional 4 h, the flask is placed in ice bath, and the contents are treated with 100 ml of crushed ice, acidified with 39 ml of 6 N HCl to pH 1, and the precipitated material is stirred for 1.5 h in an ice bath. The undesired solid is removed by filtration, and the filtrate is extracted seven times with EtOAc. The combined organic extracts are concentrated at reduced pressure, treated with toluene, reconcentrated on rotary evaporator to azeotrope most of the water, suspended in $CH_2Cl_2$ and reconcentrated again at reduced pressure to obtain 19.9 g (81%) of 2-chloro-6-(hydroxymethyl)-3-pyridinol as a pale yellow solid sufficiently pure for subsequent reaction. MS for $C_6H_6ClNO_2$: *m/z*: 159 $(M)^+$.

2-Chloro-6-(hydroxymethyl)-3-pyridinol (11.6 g, 72.7 mmol) and $NaHCO_3$ (18.3 g, 218 mmol) are dissolved in 200 ml water in a flask. The mixture is stirred until homogeneous, is cooled in an ice bath, is treated with iodine (19.4 g, 76.3 mmol), and is stirred over 60 h at rt as the cooling bath expired. The pH of the mixture is adjusted to 3 with 2N $NaHSO_4$, and the mixture is extracted with 4 x 50 ml EtOAc. The combined organic layer is dried ($MgSO_4$) and is concentrated *in vacuo* to a yellow solid. The crude solid is washed with EtOAc to provide 12.9 g (62%) of 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol as an off-white solid. The filtrate is concentrated to a small volume and is chromatographed over 250 g $SiO_2$ (230-400

mesh) eluting with EtOAc/$CH_2Cl_2$/hexane/acetic acid 2.5:4.5:4:0.1. The appropriate fractions are combined and concentrated to afford an additional 2.4 g (12%) of pure 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol. MS for $C_6H_5ClINO_2$, *m/z*: 285 $(M)^+$.

2-Chloro-6-(hydroxymethyl)-4-iodopyridin-3-ol (5.7 g, 20 mmol) is combined with bis (triphenylphosphine) palladium dichloride (1.12 g, 1.6 mmol) in 50 ml DMF under nitrogen. The mixture is treated with tetravinyl tin, is warmed to 60°C for 6 h followed by 50°C for 18 h, and at rt for 72 h. The mixture is diluted with 250 ml EtOAc and is extracted with 4 x 100 ml 2:1:1 water/saturated NaCl/saturated $NaHCO_3$. The organic layer is dried ($MgSO_4$) and is concentrated *in vacuo* to a yellow oil. The crude material is chromatographed over 200 g $SiO_2$ (230-400 mesh) eluting with 37% EtOAc/hexane. The appropriate fractions are combined and concentrated to afford 1.45 g (39%) of 2-chloro-6-(hydroxymethyl)-4-vinylpyridin-3-ol as a pale yellow solid. MS for $C_8H_8ClNO_2$ (EI) *m/z*: 185 $(M)^+$.

2-Chloro-6-(hydroxymethyl)-4-vinylpyridin-3-ol (1.35 g, 7.8 mmol) is dissolved in 12 ml DMF in a dry flask under nitrogen. The yellow solution is treated with 60% sodium hydride (312 mg, 7.8 mmol), is stirred 30 min, and is treated with allyl bromide (744 µL, 8.6 mmol). The reaction is stirred 6 h at RT, is diluted with 50 ml EtOAc, and is washed with 4 x 25 ml 2:1:1 water/sat'd NaCl/sat'd $NaHCO_3$. The organic layer is dried ($MgSO_4$) and is concentrated *in vacuo* to a yellow oil. The crude material is chromatographed over 50 g $SiO_2$ (230-400 mesh) eluting with 30% EtOAc/hexane. The appropriate fractions are combined and concentrated to give 1.43 g (81%) of [5-(allyloxy)-6-chloro-4-vinylpyridin-2-yl]methanol as a white solid. MS for $C_{11}H_{12}ClNO_2$ (EI) *m/z*: 225 $(M)^+$.

[5-(Allyloxy)-6-chloro-4-vinylpyridin-2-yl]methanol (225 mg, 1.0 mmol) is combined with bis (tricyclohexylphosphine) benzylidene ruthenium (IV) dichloride (16.5 mg, 0.02 mmol) in 5 ml $CH_2Cl_2$ and the reaction is stirred 4 h at RT. The volatiles are removed *in vacuo* and the residue is chromatographed over 15 g $SiO_2$ (230-400 mesh) eluting with 40% EtOAc/hexane. The appropriate fractions are combined and concentrated to give 175 mg (89%) of (8-chloro-2H-pyrano[2,3-c]pyridin-6-yl)methanol as a tan solid. MS for $C_9H_8ClNO_2$ (EI) *m/z*: 197 $(M)^+$.

(8-Chloro-2H-pyrano[2,3-c]pyridin-6-yl)methanol (988 mg, 5.0 mmol) is combined with 100 mg 10% Pd/C in 25 ml EtOH containing 3 ml (6 mmol) of 2N aqueous NaOH in a 250 ml PARR shaker bottle. The reaction is hydrogenated at 50

PSI for 48 h, the catalyst is removed by filtration, and the filtrate is concentrated to dryness. The mixture is partitioned between 1 x 10 ml 1:1 saturated NaCl/ conc. $NH_4OH$ and 4 x 10 ml $CH_2Cl_2$ and the combined organic layer is dried ($K_2CO_3$). The mixture is concentrated *in vacuo* to give 730 mg (89%) of 3,4-dihydro-2H-pyrano[2,3-c]pyridin-6-ylmethanol as an off-white solid. HRMS (FAB) calcd for $C_9H_{11}NO_2$ +H: 166.0868, found 166.0868 $(M+H)^+$.

Oxalyl chloride (452µL, 5.1 mmol) is dissolved in 15 ml $CH_2Cl_2$ under nitrogen at -78°C. The solution is treated drop-wise with DMSO (729µL, 10.3 mmol) in 5 ml $CH_2Cl_2$ and the mixture is stirred 30 min at -78°C. 3,4-Dihydro-2H-pyrano[2,3-c]pyridin-6-ylmethanol (731 mg, 4.4 mmol) is added drop-wise to the reaction mixture in 5 ml $CH_2Cl_2$ and the reaction is stirred 30 min at -78°C. The mixture is treated with TEA (3.08 ml, 22.1 mmol), is stirred 30 min at -78°C and 2 h at 0°C. The mixture is washed with 1 x 10 ml saturated $NaHCO_3$, is dried ($K_2CO_3$), and is concentrated *in vacuo*. The crude intermediate is chromatographed over 25 g $SiO_2$ (230-400 mesh) eluting with 35% EtOAc/hexane. The appropriate fractions are combined and concentrated to give 685 mg (95%) of the aldehyde as an off-white solid.

The aldehyde (685 mg, 4.2 mmol) is combined with $NaClO_2$ (80%, 1.42 g, 12.6 mmol) and $KH_2PO_4$ in 15 ml THF/7 ml t-BuOH/ 7 ml water and the reaction is stirred overnight under a stream of nitrogen. The reaction is concentrated to dryness *in vacuo* and the residue is dissolved in 10 ml water. The pH of the mixture is adjusted to 5 with 12 N HCl, the white solid is collected, washed with water, and is dried *in vacuo* at 50°C to afford 565 mg (82%) of 3,4-dihydro-2*H*-pyrano[2,3-c]pyridine-6-carboxylic acid as a white solid. HRMS (FAB) calcd for $C_9H_9NO_3$ +H: 180.0661, found 180.0652 $(M+H)^+$.

Compounds of Formula I where W is (F) are made using the coupling procedures discussed herein and in cited references, making non-critical changes to obtain the desired compounds. The following intermediates to provide W of formula I are for exemplification only and are not intended to limit the scope of the present invention. Other intermediates within the scope of the present invention can be obtained using known procedures or by making slight modifications to known procedures.

**Intermediate F1: 1,3-Benzoxazole-6-carboxylic acid**

A mixture of 4-amino-3-hydroxybenzoic acid (250 mg, 1.63 mmol) and trimethyl orthoformate (500 μL, 4.57 mmol) is heated in an oil bath at 100°C for 2 h. The mixture is cooled to rt and diluted with MeOH. The resulting solution is filtered through a pad of Celite, and the filtrate is concentrated *in vacuo* to give Intermediate F1 as a brown solid (237 mg, 89%): $^1$H NMR (DMSO-$d_6$) δ 13.2, 8.9, 8.3, 8.0, 7.9.

**Intermediate F2: 2-Methyl-1,3-benzoxazole-6-carboxylic acid**

A mixture of 4-amino-3-hydroxybenzoic acid (500 mg, 3.7 mmol) and trimethyl orthoacetate (1.0 mL, 7.9 mmol) is heated in an oil bath to 100°C for 2 h. The mixture is cooled to rt and diluted with MeOH. The resulting solution is filtered through a pad of Celite, and the filtrate is concentrated *in vacuo* to give Intermediate F2 as an off-white solid (266 mg, 46%): $^1$H NMR (DMSO-$d_6$) δ 13.1, 8.2, 8.0, 7.7, 2.7.

**Intermediate F3: 1,3-Benzoxazole-5-carboxylic acid**

A mixture of 4-amino-3-hydroxybenzoic acid (1.0 g, 6.5 mmol) and trimethyl orthoformate (2.0 mL, 18.3 mmol) is heated in an oil bath at 100°C for 30 h. The mixture is cooled to rt and diluted with MeOH. The resulting solution is filtered through a pad of Celite, and the filtrate is concentrated *in vacuo* to give Intermediate F3 as a brown solid (290 mg, 27%): $^1$H NMR (DMSO-$d_6$) δ 13.0, 8.9, 8.3, 8.1, 7.9.

**Intermediate F4: 2-Methyl-1,3-benzoxazole-5-carboxylic acid**

A mixture of 4-amino-3-hydroxybenzoic acid (480 mg, 3.1 mmol) and trimethyl orthoacetate (1.0 mL, 7.9 mmol) is heated in an oil bath to 107°C for 2 h. The mixture is cooled to rt and diluted with MeOH. The resulting solution is filtered through a pad of silica gel and the filtrate is concentrated *in vacuo* to give Intermediate F4 as an orange solid (490 mg, 88%): $^1$H NMR (DMSO-$d_6$) δ 13.0, 8.2, 8.0, 7.8, 2.7.

**Intermediate F5: 5-Indancarboxylic acid**

To a stirred 6% aqueous sodium hypochlorite solution in an oil bath to 55°C is added 1-indane-5-yl-ethanone (1.0 g, 6.2 mmol). The solution is stirred at 55°C for 2 h, followed by cooling to rt. Solid sodium bisulfite is added until the solution became clear. The mixture is diluted with water, followed by aqueous hydrochloric acid (6.0 M). The solid that forms is filtered and washed several times with water. The solid is dried under high vacuum at 60°C for 5 h to afford Intermediate F5 as a white solid (0.96 g, 95%): $^1$H NMR ($CDCl_3$) δ 8.0, 7.9, 7.3, 3.0, 2.1.

**Intermediate F6: [1,3]Oxazolo[5,4-c]pyridine-6-carboxylic acid**

2-Chloro-3-pyridinol (20.0 g, 0.154 mole), $NaHCO_3$ (19.5g, 0.232 mole, 1.5 equ), and 150 mL of water are placed in a flask. The flask is placed in an oil bath at 90°C, and after 5 minutes, 37% aqueous formaldehyde (40.5 mL, 0.541 mole, 3.5 equ) is added in six unequal doses in the following order: 12 mL, 3 x 8 mL, then 2.2 mL all at 90-minute intervals and then the final 2.3 mL after the reaction had stirred for 15 h at 90°C. The reaction is stirred at 90°C for another 4 h and then is cooled by placing the flask in an ice bath. The pH of the reaction is then adjusted to 1 using 6N HCl. The reaction is stirred for 1.5 h in an ice bath allowing an undesired solid to form. The undesired solid is removed by filtration, and the filtrate is extracted seven times with EtOAc. The combined organic extracts are concentrated *in vacuo*, toluene is added to the flask and removed *in vacuo* to azeotrope water, and then $CH_2Cl_2$ is added and removed *in vacuo* to obtain 2-chloro-6-(hydroxymethyl)-3-pyridinol (I-10-F) as a pale yellow solid (81% yield) sufficiently pure for subsequent reaction. MS (EI) for $C_6H_6ClNO_2$, *m/z*: 159$(M)^+$.

I-10-F (11.6 g, 72.7 mmol) and $NaHCO_3$ (18.3 g, 218 mmol) are added to 200 mL water. The mixture is stirred until homogeneous, the flask is placed in an ice bath, iodine (19.4 g, 76.3 mmol) is added, and the reaction is stirred over the weekend at rt. The pH of the mixture is adjusted to 3 with 2N $NaHSO_4$, and the mixture is extracted with 4 x 50 mL EtOAc. The combined organic layer is dried ($MgSO_4$), is filtered, and the filtrate is concentrated *in vacuo* to a yellow solid. The crude solid is washed with EtOAc to provide 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-12-F) as an off-white solid (62% yield), and the filtrate is concentrated to a small volume and is chromatographed over 250 g silica gel (230-400 mesh) eluting with 2.5:4.5:4:0.1 EtOAc/$CH_2Cl_2$/hexane/acetic acid. The desire fractions are combined

and concentrated to afford an additional pure I-12-F (12% yield). MS (EI) for $C_6H_5ClINO_2$, *m/z*: 285$(M)^+$.

4-(Benzylamino)-2-chloro-6-(hydroxymethyl)-3-pyridinol (I-13-F) may be produced by amination of 2-chloro-6-(hydroxymethyl)-4-iodo-3-pyridinol (I-12-F) with benzylamine under palladium catalysis. Amination of aryl iodides with primary amines such as benzylamine under palladium catalysis is generally described in a review by B.H. Yang and S.L. Buchwald in *J. Organomet. Chem.*, 576, 125-146, 1999 and in greater detail in the references therein.

I-13-F may be oxidized to 4-(benzylamino)-2-chloro-3-hydroxypyridine-6-carboxaldehyde (I-14-F) under a wide variety of conditions (e.g., TPAP and NMO in $CH_2Cl_2$). I-14-F may be oxidized to produce the corresponding carboxylic acid I-15-F using an oxidizing reagent such as $NaClO_2$ and $KH_2PO_4$ in DMSO/$H_2O$ or $Ag_2O$, or hydrogen peroxide or ruthenium tetroxide.

Removal of the benzyl group and the chloro group of Acid I-15-F may be accomplished by utilizing hydrogen or a hydrogen source (e.g., cyclohexene, cyclohexadiene, ammonium formate, hydrazine, etc.) in the presence of Pd/C or other catalyst, under a variety of conditions and in various solvents, to produce 4-amino-5-hydroxypyridine-2-carboxylic acid (Acid I-16-F).

Cyclocondensation of Acid I-16-F with trimethyl orthoformate in the presence of catalytic *para*-toluenesulfonic acid may be conducted to produce [1,3]oxazolo[5,4-c]pyridine-6-carboxylic acid.

**Intermediate F7: 2-Benzoisothiophene-5-carboxylic acid**

Intermediate F7 can be made by the saponification of the methyl ester I-20-E, which can be made pursuant to Wynberg, Hans, et al., *Recl. Trav. Chim. Pays-Bas* (1968), 87(10), 1006-1010.

**Intermediate F8: 1,3-Benzothiazole-5-carboxylic acid**

A solution of sodium sulfide•nanohydrate (1.15 g, 4.9 mmol) in methanol-water (ca. 10 mL, 1:1) is warmed on a hot plate. To this solution is added elemental sulfur (150 mg, 4.6 mmol). Heating is continued for 15 min before the solution is poured into a separate solution of 1.0 g (4.6 mmol) of methyl 4-chloro-3-nitrobenzoate (see: Kuene, *J. Am. Chem. Soc.* **1962**, *48*, 837.) in MeOH (5.0 mL).

The mixture is stirred for 30 min, followed by cooling in a refrigerator overnight. The solid precipitate is filtered, washed with water and methanol, and dried *in vacuo* at 50 °C to afford 650 mg (65%) of dimethyl 4,4'-dithio-bis-(3-nitrobenzoate) as a yellow solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 9.0, 8.2, 7.9, 4.0.

To a stirred solution of dimethyl 4,4'-dithio-bis-(3-nitrobenzoate) (900mg, 2.12 mmol) in ethanol is added tin powder (1.91 g, 17.0 mmol). The mixture is heated in a 70°C oil bath for 30 minutes before 2.8 mL of concentrated hydrochloric acid is added drop-wise. After complete addition, the mixture is stirred for an additional 10 min, followed by cooling to RT. The reaction mixture is filtered and the fitrate is concentrated *in vacuo* to a solid. The solid is washed with 1.0M aqueous hydrochloric acid and dried *in vacuo* to afford a yellow solid. The solid (750 mg, 3.42 mmol) is suspended in formic acid (4 mL) in a 100°C oil bath. Zinc dust (15 mg) is added to the reaction. The mixture is stirred for 10 min, followed by cooling to RT. The mixture is diluted with water and extracted with EtOAc. The organic layer is dried ($MgSO_4$), filtered and concentrated *in vacuo* to afford 640 mg (97%) of methyl 1,3-benzothiazole-5-carboxylate as a yellow solid: $^1$H NMR (400 MHz, $CDCl_3$) δ 9.1, 8.9, 8.2, 8.1, 4.0.

To a stirred solution of methyl 1,3-benzothiazole-5-carboxylate (290 mg, 1.5 mmol) in MeOH (20 mL) is added sodium hydroxide (10 mL of a 5% aqueous solution). The mixture is heated in a 65°C oil bath for 30 min, followed by cooling to RT. The mixture is diluted with water and extracted with hexanes-ether (1:1). The organic layer is discarded and the aqueous layer is acidified with concentrated hydrochloric acid to pH=1. The aqueous layer is extracted with ether. The ethereal layer is dried ($MgSO_4$), filtered and concentrated *in vacuo* to a yellow powder for 1,3-benzothiazole-5-carboxylic acid (260 mg, 98%): $^1$H NMR (400 MHz, DMSO-$d_6$) δ 13-12.5, 9.5, 8.6, 8.3, 8.0.

**Intermediate F9: 3-Methyl-1,2-benzisoxazole-6-carboxylic acid**

3-Hydroxybenzoic acid (13.8 g, 100 mmol) is dissolved in concentrated $NH_4OH$ (200 mL) using an overhead stirrer and is treated slowly dropwise with a solution of iodine (23.4 g, 92 mmol) and KI (18.26 g, 110 mmol) in water (100 mL). The solution is stirred for 1 h at rt and then treated rapidly dropwise with concentrated HCl (180 mL). The white solid is collected via filtration, rinsed with water and dried

overnight [by pulling air through the solid] *in vacuo* to afford 13.05 g (54%) of 3-hydroxy-4-iodobenzoic acid as a tan solid. $^1$H NMR (DMSO-$d_6$): δ 7.13, 7.43, 7.80, 10.71, 12.98 ppm.

3-Hydroxy-4-iodobenzoic acid (12.55 g, 47.5 mmol) is dissolved in MeOH (200 mL), treated slowly dropwise with thionyl chloride (32.3 mL, 442.9 mmol) at rt, then heated to reflux for 20 h. The mixture is concentrated to dryness and partitioned between $CH_2Cl_2$ (100 mL) and saturated $NaHCO_3$ (50 mL). Not all of the residue is solubilized, so the mixture is filtered and the solid is washed with a small amount of $CH_2Cl_2$ and MeOH. The original filtrate and the organic washes are combined, concentrated to dryness, dissolved in 10% MeOH / $CH_2Cl_2$ (200 mL), diluted with water (50 mL) and the layers separated. The organics are washed with saturated $NaHCO_3$ (2 x 50 mL), then water (50 mL), dried ($Na_2SO_4$) and concentrated to a tan solid. This solid is triturated with $CH_2Cl_2$ (50 mL) and filtered. The two solids are combined to afford 9.4 g (70%) of methyl 3-hydroxy-4-iodobenzoate as a beige solid. HRMS (FAB) calcd for $C_8H_7IO_3$ +$H_1$: 278.9520, found 278.9521.

Methyl 3-hydroxy-4-iodobenzoate (5.22 g, 18.8 mmol) is combined with trimethylsilylacetylene (3.71 mL, 26.3 mmol), bis(triphenylphosphine)palladium dichloride (386 mg, 0.55 mmol) and cuprous iodide (54 mg, 0.28 mmol) in THF (20 mL) / $CHCl_3$ (40 mL) in a dry flask, under nitrogen. TEA (8.14 mL< 58.4 mmol) is added and the mixture is heated to 50°C for 4 h. The mixture is diluted with $CHCl_3$ (60 mL), washed with 5% HCl (2 x 40 mL), dried ($MgSO_4$) and concentrated to a brown paste (8.31 g). The crude material is chromatographed over a standard 90 g Biotage column, eluting with 10% EtOAc / hexane (1 L) followed by 15 % EtOAc / hexane (1 L). The appropriate fractions are combined and concentrated to afford 4.22 g (91%) of methyl 3-hydroxy-4-[(trimethylsilyl)ethynyl]benzoate as a yellow solid. HRMS (FAB) calcd for $C_{13}H_{16}O_3SI$ +$H_1$: 249.0947, found 249.0947.

Methyl 3-hydroxy-4-[(trimethylsilyl)ethynyl]benzoate (540 mg, 2.17 mmole) is combined with 4 ml formic acid under nitrogen. The reaction is warmed to 80°C for 12 h, is cooled to rt, and the volatiles are removed *in vacuo*. The black residue is chromatographed over 25 g silica gel (230-400 mesh) eluting with 15% EtOAc/hexane. The appropriate fractions are combined and concentrated to provide 350 mg (83%) of methyl 4-acetyl-3-hydroxybenzoate as a pale yellow solid. $^1$H NMR ($CDCl_3$) δ 2.70, 3.95, 7.54, 7.64, 7.82, 12.10 ppm.

Methyl 4-acetyl-3-hydroxybenzoate (350 mg, 1.8 mmole) is combined with 5 ml absolute EtOH. The solution is treated with hydroxylamine hydrochloride (125 mg, 1.8 mmole) dissolved in 0.9 ml 2N aqueous NaOH, and the reaction is stirred overnight at rt. The volatiles are removed *in vacuo* and the residue is washed with $H_2O$, collected, and dried to give 294 mg (78%) of methyl 3-hydroxy-4-[N-hydroxyethanimidoyl]benzoate as a tan solid. MS (EI) *m/z* : 209 ($M^+$).

Methyl 3-hydroxy-4-[N-hydroxyethanimidoyl]benzoate (250 mg, 1.19 mmole) is combined with triphenylphosphine (446 mg, 1.7 mmole) in 14 ml dry THF in a dry flask under nitrogen. The solution is treated slowly dropwise with N,N'-diethylazidodicarboxylate (268 µL, 1.7 mmole) in 10 ml dry THF. The reaction is stirred 4 h at rt. The volatiles are removed *in vacuo* and the residue is chromatographed over 30 g silica gel (230-400 mesh) eluting with 10% EtOAc/hexane. The appropriate fractions are combined and concentrated to provide 125 mg (55%) of methyl 3-methyl-1,2-benzisoxazole-6-carboxylate slightly contaminated (< 10%) with methyl 4-acetyl-3-hydroxybenzoate. $^1H$ NMR ($CDCl_3$) δ 2.64, 4.00, 7.70, 8.01, 8.25 ppm.

Methyl 3-methyl-1,2-benzisoxazole-6-carboxylate (170 mg, 0.89 mmole) is dissolved in 6 ml MeOH under nitrogen. The solution is treated with 2N aqueous NaOH (1 ml, 2 mmole) and the mixture is stirred 4 h at rt. The volatiles are removed *in vacuo* and the residue is dissolved in 4 ml water. The pH of the solution is adjusted to 3 with 10% aqueous HCl, the white precipitate is collected, is washed with water, and is dried to give 144 mg (92%) of 3-methyl-1,2-benzisoxazole-6-carboxylic acid as a white solid. MS *m/z* (ESI): 176.2 $(M-H)^-$.

**Intermediate F10: 3-Methyl-1,2-benzisoxazole-5-carboxylic acid**

Intermediate F13 is obtained according to the methods discussed for preparing Intermediate F12 starting with 4-hydroxybenzoic acid.

**Intermediate F11:** 1*H*-indazole-6-carboxylic acid

To a stirred solution of 3-amino-4-methylbenzoic acid (5.0 g, 33 mmol) in a mixture of water (50 mL) and concentrated hydrochloric acid (15 mL) in an acetone-crushed ice bath is added a solution of sodium nitrite in water (12 mL) dropwise. The solution is stirred for 10 min, followed by the addition of *tert*-butyl mercaptan (1.8

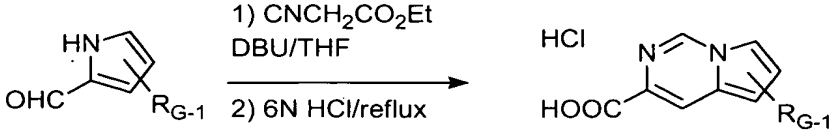
mL, 16 mmol). The mixture is stirred for 1 h. The solid precipitate is filtered, washed with water and dried *in vacuo* to obtain 3.85 g (95%) of 3-[(E)-(*tert*-butylthio)diazenyl]-4-methylbenzoic acid as a tan solid: $^1$H NMR (400 MHz, DMSO-$d_6$) δ 13.2, 7.8, 7.5, 7.3, 2.1, 1.6.

To a stirred solution of potassium *tert*-butoxide (8.1 g, 73 mmol) in DMSO (30 mL) was added a solution of 3-[(E)-(*tert*-butylthio)diazenyl]-4-methylbenzoic acid (1.9 g, 7.3 mmol) at RT. The mixture was stirred overnight, followed by the adition of ice water. The aqueous layer was extracted with ethyl acetate. The organic layer was dicarded. The pH of the aqueous layer was adjusted to 4-5 with aqueous 1N HCl. The aqueous layer was extracted with ethyl acetate. The organic layer was washed with brine, dried ($MgSO_4$), filtered and concentrated *in vacuo* to afford 800 mg (97%) of 1*H*-indazole-6-carboxylic acid as a tan solid: $^1$H NMR (400 MHz, DMSO-$d_6$) δ 13.4, 13.0, 8.2, 8.1, 7.9, 7.7.

Compounds of Formula I where W is (G) are made using the coupling procedures discussed herein and in US 20020049225A1 and US 20020042428A1, making non-critical changes to obtain compounds where Azabicyclo is other than I. The following intermediates to provide W of formula I are for exemplification only and are not intended to limit the scope of the present invention. Other intermediates within the scope of the present invention can be obtained using known procedures or by making slight modifications to known procedures.

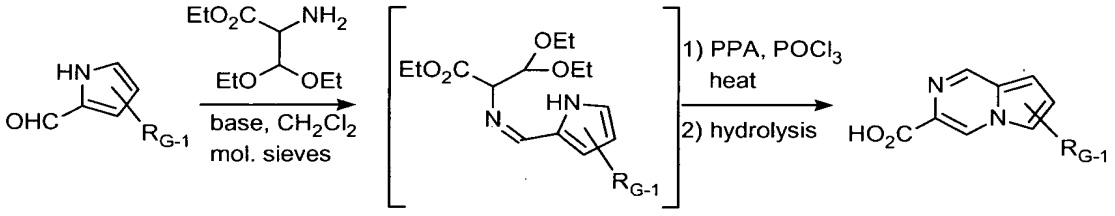
It will be apparent to those skilled in the art that the requisite carboxylic acids can be synthesized by known procedures, or modification thereof, some of which are described herein. For example, 3-(pyrrolo[1,2-c]pyrimidine)carboxylic acid can be synthesized from the corresponding pyrrole-2-carboxaldehyde by reaction with an isocyanoacetate in the presence of base as described in *J. Org. Chem.* **1999**, *64*, 7788 and *J. Org. Chem.* **1976**, *41*, 1482 or by methods described in *Liebigs Ann. Chem.* **1987**, 491. Scheme 1G depicts this transformation.

Scheme 1G



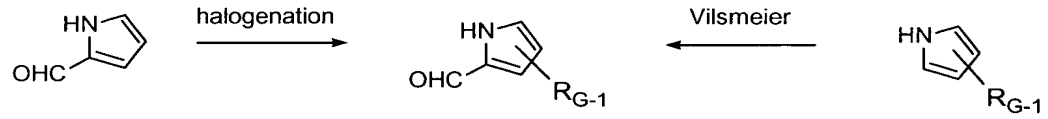
The pyrrolo[1,2-a]pyrazine acid fragment can be prepared using the methods shown in Scheme 2G. The ester intermediate can be prepared using methods described in Dekhane, M.; Potier, P.; Dodd, R. H. *Tetrahedron* **1993**, *49*, 8139-46, whereby the requisite pyrrole-2-carboxaldehyde is reacted with aminoester diethylacetal to form the imine. The imine can then be cyclized under acidic conditions to afford the desired bicyclic core. The resulting ester can be hydrolyzed under typical hydrolysis procedures well known in the art to afford the requisite pyrrolo[1,2-a]pyrazine acids.

Scheme 2G



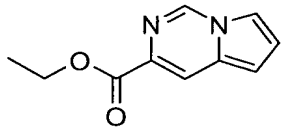

The pyrrole-2-carboxaldehydes can be obtained from commercial sources or can be synthesized by known procedures. For example, pyrrole-2-carboxaldehyde can be converted into 4-halo, 5-halo and 4,5-dihalopyrrole-2-carboxaldehydes as described in *Bull. Soc. Chim. Fr.* **1973**, 351. See Examples 12-22. Alternatively, substituted pyrroles can be converted into pyrrole carboxaldehydes by Vilsmeier formylation using procedures well known in the art (see *J. Het. Chem.* **1991**, *28*, 2053, *Synth. Commun.* **1994**, *24*, 1389 or *Synthesis*, **1995**, 1480. Scheme 3G depicts these transformations.

Scheme 3G




Non-limiting examples of W when W is (G):

Ethyl pyrrolo[1,2-c]pyrimidine-3-carboxylate:




A solution of pyrrole-2-carboxaldehyde (3.6g, 38.1mmol) in 40mL dry THF is added to ethyl isocyanoacetate (4.3g, 38.1mmol) and DBU (5.8g, 38.2mmol) in 60mL

dry THF. After stirring at RT overnight, the reaction is neutralized with 10% AcOH. The solvent is removed *in vacuo*. The residue is taken up in EtOAc/$H_2O$, the aqueous layer is extracted with EtOAc, dried ($MgSO_4$), filtered and concentrated. The residue is purified by flash chromatography on silica gel eluting with 30-70% EtOAc/hexanes. The carboxylate is obtained (4.45g, 61%) as an off-white solid. $^1$H NMR (400MHz, $CDCl_3$) δ 8.86, 8.24, 7.54, 7.01, 6.78, 4.45, 1.44.

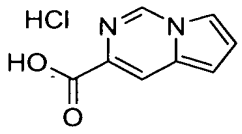
The following compounds are made from the corresponding pyrrole-2-carboxaldehydes, making non-critical variations:

Ethyl 7-chloropyrrolo[1,2-c]pyrimidine-3-carboxylate. Yield 25% starting from 5-chloropyrrole-2-carboxaldehyde. $^1$H NMR (400MHz, $CDCl_3$) δ 8.86, 8.21, 6.91-6.89, 6.80-6.77, 4.50-4.43, 1.47-1.42.

Ethyl 6-chloropyrrolo[1,2-c]pyrimidine-3-carboxylate. Yield 49% starting from 4-chloropyrrole-2-carboxaldehyde. $^1$H NMR (400MHz, $CDCl_3$) δ 8.76, 8.14, 7.51, 6.72, 4.49-4.42, 1.46-1.41.

Ethyl 6-bromopyrrolo[1,2-c]pyrimidine-3-carboxylate. Yield 9% starting from 4-bromopyrrole-2-carboxaldehyde. $^1$H NMR (400MHz, $CDCl_3$) δ 8.77, 8.15, 7.55, 6.79, 4.49-4.42, 1.46-1.41.

**Pyrrolo[1,2-c]pyrimidine-3-carboxylic acid hydrochloride:**




Ethyl pyrrolo[1,2-c]pyrimidine-3-carboxylate (4.1g, 21.2mmol) is dissolved/suspended in 100mL concentrated HCl. The mixture is heated under reflux. After 4h, the reaction is cooled and the solvent is removed *in vacuo*. Absolute EtOH is added and the solvent is removed (twice) to afford a yellow-green solid. The solid is triturated with $Et_2O$ and dried to give 4.28g (100%) of pyrrolo[1,2-c]pyrimidine-3-carboxylic acid as the hydrochloride salt. The solid can be recrystallized from EtOH. $^1$H NMR (400MHz, DMSO) δ 9.24, 8.21, 7.90, 7.06, 6.85.

The following compounds are made from the corresponding ethyl pyrrolo[1,2-c]pyrimidine-3-carboxylates, making non-critical variations:

7-Chloropyrrolo[1,2-c]pyrimidine-3-carboxylic acid hydrochloride. Yield 77%. $^1$H NMR (400MHz, $d_6$-DMSO) δ 9.3, 9.04, 8.25, 7.16-7.14, 6.96-6.94.

6-Chloropyrrolo[1,2-c]pyrimidine-3-carboxylic acid hydrochloride. Yield 95%. $^1$H NMR (400MHz, $d_6$-DMSO) δ 11.15, 9.14, 8.15, 8.04, 6.91.

6-Bromopyrrolo[1,2-c]pyrimidine-3-carboxylic acid hydrochloride. Yield 97%. $^1$H NMR (400MHz, $d_6$-DMSO) δ 10.2, 9.12, 8.15, 8.04, 6.96.

**Imidazo[1,5-a]pyridine-7-carboxylic acid**:

Methyl nicotinate 1-oxide (Coperet, C.; Adolfsson, H.; Khuong, T-A. V.; Yudin, A. K.; Sharpless, K. B. *J. Org. Chem.* **1998**, *63*, 1740-41.) (5.0 g, 32.2 mmol) and dimethylsulfate (3.2 ml, 33.2 mmol) are placed in a 100 ml flask and heated to 65-70°C for 2 h. Upon cooling a salt precipitates. The resulting precipitate is dissolved in water (12 ml). An oxygen free solution of KCN (2.5 g, 38.7 mmol) in water (9.5 ml) is added dropwise to the mixture with vigorous stirring at 0°C. After stirring for 1 h at 0°C, the mixture is warmed to rt and stirred overnight. The solution is extracted with $CH_2Cl_2$ (3 x 25 ml) and the combined organic layers are dried ($NaSO_4$), filtered, and the solvent removed under vacuum. The resulting solid is purified by silica gel chromatography (EtOAc) to give a yellow solid (4.2 g, 25.9 mmol, 80%) for methyl 2-cyanoisonicotinate. MS (ESI+) for $C_8H_6N_2O_2$ *m/z* 163.0 $(M+H)^+$.

To a solution of methyl 2-cyanoisonicotinate (4.22 g, 25.9 mmol) and 10 % palladium on charcoal (2.8 g, 2.6 mmol) in MeOH (400 ml) was added conc. HCl (7.5 ml). The mixture is hydrogenated at rt and balloon pressure, until no more hydrogen is consumed (about 2 h). The reaction mixture is filtered through a pad of celite and the solvent is removed in vacuum to give a yellow solid (4.5 g, 18.8 mmol, 73%) for methyl 2-(aminomethyl) isonicotinate. This compound is used without further purification. MS (ESI+) for $C_8H_{10}N_2O_2$ *m/z* 167.2 $(M+H)^+$; HRMS (FAB) calcd for $C_8H_{10}N_2O_2$+H 167.0820, found 167.0821.

**Procedure A:**

A mixture of methyl 2-(aminomethyl) isonicotinate (4.3 g, 18.0 mmol) and acetic formic anhydride (which is prepared by heating to 50°C acetic anhydride (75.0 ml) and formic acid (65.0 ml) for 2 h) is stirred at rt for 1 h. The reaction mixture is heated to 35°C with an oil bath for 1 h. The reaction mixture is cooled to 0°C in an ice-bath and neutralized with ammonium hydroxide at such a rate that the temperature did not rise above 5°C. The mixture is extracted with $CH_2Cl_2$ (3 x 200 ml) and the

combined organic layers are dried ($NaSO_4$), filtered, and the solvent removed under vacuum. The resulting solid is purified with DOWEX 50WX2-400 ion-exchange resin to give a yellow solid (3.2 g, 18.0 mmol, 100%) for methyl imidazo [1,2-a]pyridin-6-carboxylate. MS (ESI+) for $C_9H_8N_2O_2$ *m/z* 177.03 $(M+H)^+$.

**Procedure B:**

Methyl imidazo [1,2-a]pyridin-6-carboxylate (3.2 g, 18.0 mmol) is dissolved in 3N HCl (200 ml) and heated under reflux for 3 h. The solvent is removed under vacuum and the resulting brown solid is recrystallized from $H_2O$/EtOH/$Et_2O$ to afford a light brown solid (4.3 g, 21.6 mmol, 119%) for imidazo[1,5-a]pyridine-7-carboxylic acid. HRMS (FAB) calcd for $C_8H_6N_2O_2$+H 163.0508, found 163.0489.

**Pyrrolo[1,2-a]pyrazine-3-carboxylic acid hydrochloride:**

**Procedure E:**

Pyrrole-2-carboxaldehyde (recrystallized from EtOAc/hexanes prior to use) (3.67 g, 38.6 mmol) is added to a solution of ethyl 3-ethoxy-O-ethylserinate (7.95 g, 38.6 mmol) in freshly distilled THF or $CH_2Cl_2$ (100 mL) in an oven dried 250 mL flask. 3Å activated molecular sieves (approximately 1/3 the volume of the reaction vessel) are added, and the resulting mixture is allowed to stir under nitrogen until the starting pyrrole-2-carboxaldehyde is consumed as determined by $^1H$ NMR. The reaction mixture is filtered through a pad of celite, and the solvent removed *in vacuo* to give an orange oil (9.59 g) for ethyl 3-ethoxy-O-ethyl-N-(1*H*-pyrrol-2-ylmethylene)serinate that is used without purification: MS (ESI+) for $C_{14}H_{22}N_2O_4$ *m/z* 282.96 $(M+H)^+$.

**Procedure F:**

To a hot (65°C) solution of TFA (44 mL, 510 mmol) and phosphorus oxychloride (39.0 g, 140 mmol) is added drop-wise a solution of ethyl 3-ethoxy-O-ethyl-N-(1*H*-pyrrol-2-ylmethylene)serinate (Dekhane, M; Potier, P; Dodd, R. H. *Tetrahedron*, *49*, **1993**, 8139-46.) (9.6 g, 28.0 mmol) in anhydrous 1,2-dichloroethane (200 mL). The black mixture is allowed to stir at 65°C for 18 h at which point it is cooled to rt and neutralized with sat. $NaHCO_3$ and solid $NaHCO_3$ to pH ~ 9. The phases are separated and the basic phase extracted with EtOAc (4 x 100 mL). The

organic phases are combined, washed with brine, dried ($NaSO_4$), filtered, and concentrated to give a black oil that is purified with silica gel chromatography (35% EtOAc/heptanes to 50% over several liters) to give a light brown solid for ethyl pyrrolo[1,2-a]pyrazine-3-carboxylate. Yield 24%. HRMS (FAB) calcd for $C_{10}H_{10}N_2O_2$+H 191.0820, found 191.0823.

Pyrrolo[1,2-a]pyrazine-3-carboxylic acid hydrochloride is prepared from ethyl pyrrolo[1,2-a]pyrazine-3-carboxylate, using Procedure B to give a pale brown solid. Yield 90%. HRMS (FAB) calcd for $C_8H_6O_2N_2$+H 163.0508, found 163.0513,

**Pyrazino[1,2-a]indole-3-carboxylic acid hydrocholoride:**

To a suspension of lithium aluminum hydride (10.6g, 264 mmol) in THF (200 mL) is added dropwise a solution of ethyl indole-2-carboxylate (50.0 g, 256 mmol) in THF (250 mL) over 25 minutes. After 3 h, water (10.6 mL) is carefully added, followed by 15% NaOH (10.6 mL), followed by additional portion of water (31.8 mL). The resulting suspension is dried ($Na_2SO_4$) and filtered through celite. After concentration under reduced pressure, the white solid (34.0 g) is crystallized from EtOAc/hexanes to give white needles for 1*H*-indol-2-ylmethanol. Yield 83%. HRMS (FAB) calcd for $C_9H_9NO$+H 148.0762, found 148.0771.

1*H*-Indole-2-carbaldehyde is prepared according to Berccalli, E. M., et al, *J. Org. Chem.* **2000,** *65,* 8924-32, and crystallized from EtOAc/hexanes to give a yellow/brown plates. Yield 81%. MS (ESI+) for $C_9H_7NO$ *m/z* 146.1 $(M+H)^+$.

Ethyl 3-ethoxy-O-ethyl-N-(1H-indol-2-ylmethylene)serinate is prepared using Procedure E to give an orange oil. Yield 94%. MS (ESI+) for $C_{18}H_{24}N_2O_4$ *m/z* 333.8 $(M+H)^+$.

**Procedure G:**

Ethyl 9*H*-beta-carboline-3-carboxylate and ethyl pyrazino[1,2-a]indole-3-carboxylate are prepared according to Dekhane, M., et al, *Tetrahedron*, *49*, **1993**, 8139-46, to give a dark colored solid that is purified with silica gel chromatography (20% to 75% EtOAc/hexanes as the eluent) to give the ethyl 9*H*-beta-carboline-3-carboxylate as a brown solid (yield 16%) and the ethyl pyrazino[1,2-a]indole-3-carboxylate as a brown soild (yield 35%). Ethyl 9*H*-beta-carboline-3-carboxylate; MS

(ESI+) for $C_{14}H_{12}N_2O_2$ *m/z* 241.10 $(M+H)^+$; MS (ESI-) for $C_{14}H_{12}N_2O_2$ *m/z* 239.15 $(M-H)^-$.

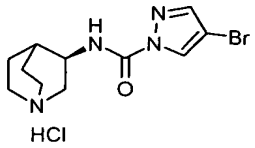
**Procedure H:**

To a solution of ethyl pyrazino[1,2-a]indole-3-carboxylate (0.49 g, 2.0 mmol) in EtOH (30 mL) is added crushed potassium hydroxide (1.1 g, 20.0 mmol) followed by water (30 mL). The resulting dark colored solution is stirred at rt for 40 min and then neutralized with conc. HCl to pH ~2. The acidic mixture is concentrated to dryness to afford pyrazino[1,2-a]indole-3-carboxylic acid hydrochloride. HRMS (FAB) calcd for $C_{12}H_8N_2O_2$+H 213.0664, found 213.0658.

Compounds of Formula I where W is (H) are made using the coupling procedures discussed herein, making non-critical changes. The following intermediates to provide formula I where W is (H) are for exemplification only and are not intended to limit the scope of the present invention. Other intermediates within the scope of the present invention can be obtained using known procedures or by making slight modifications thereof.

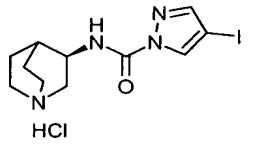
It will be apparent to those skilled in the art that the requisite carboxylic acids or carboxylic acid equivalents for when W is (H) can be obtained through synthesis via literature procedures or through the slight modification thereof. For example, methods to prepare carboxylic acids or carboxylic acid equivalents starting from pyrroles or pyrazoles are known to one of ordinary skill in the art (see *J. Org. Chem.* **1987**, *52*, 2319, *Tetrahedron Lett.* **1999**, *40*, 2733 and Greene, T. W. and Wuts, P. G. M. "Protective Groups in Organic Synthesis", 3rd Edition, p. 549, New York:Wiley, (1999)). Several pyrroles and pyrazoles of the Formula W-H are commercially available or can be obtained by methods described in *Synthesis* **1997**, 563, *J. Heterocyclic Chem.* **1993**, *30*, 865, *Heterocycles* **1982**, *19*, 1223 and *J. Org. Chem.* **1984**, *49*, 3239.

**Example 1(H):** N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-4-bromo-1H-pyrazole-1-carboxamide hydrochloride:



A solution of 4-bromopyrazole (0.52g, 3.5mmol) in 30mL EtOAc is added to excess phosgene (10mL, 20% solution in toluene) in EtOAc. After complete addition, the solution is refluxed for 1 h, cooled and concentrated *in vacuo*. EtOAc is added, and the mixture is concentrated again. The residue is treated with 20mL THF, (*R*)-(+)-3-aminoquinuclidine dihydrochloride (0.71g, 3.5mmol) and excess TEA (5.0mL, 68.1mmol). After 60h, 1N NaOH solution is added. The mixture is extracted with $CHCl_3$, dried ($MgSO_4$), filtered and concentrated. The residue is purified by flash chromatography (Biotage 40S, 90:9:1 $CHCl_3$/MeOH/$NH_4OH$). Example 1(H) is prepared and recrystallized from MeOH/EtOAc to afford 289 mg (25%) of a white solid. HRMS (FAB) calcd for $C_{11}H_{15}BrN_4O$+H 299.0508, found 299.0516.

**Example 2(H):** N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-4-iodo-1H-pyrazole-1-carboxamide hydrochloride:

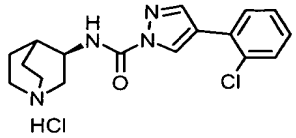


Phenyl chloroformate (0.75mL, 6.0mmol) is added dropwise to a solution of 4-iodopyrazole (1.05g, 5.4mmol) and TEA (0.9mL, 6.5mmol) in 15mL $CH_2Cl_2$. The reaction is stirred at RT. After 60h, water is added. The mixture is extracted with $CH_2Cl_2$, dried ($MgSO_4$), filtered and concentrated. Hexane is added and the solvent is removed *in vacuo*. A white solid forms on standing to provide 1.6g (95%) of phenyl 4-iodo-1H-pyrazole-1-carboxylate. MS (EI) *m/z* 315.1 ($M^+$).

Phenyl 4-iodo-1H-pyrazole-1-carboxylate (1.6g, 5.2mmol) and (R)-(+)-3-aminoquinuclidine dihydrochloride (1.0g, 5.2mmol) are suspended in 10mL DMF. DIEA (2.7mL, 15.5mmol) is added dropwise. After 36 h, the solvent is removed and the residue is taken up in 1N NaOH and $CHCl_3$. The aqueous layer is extracted with $CHCl_3$, dried ($MgSO_4$), filtered and concentrated. The residue is purified by chromatography (Biotage 40S, 90:9:1 $CHCl_3$/MeOH/$NH_4OH$) to provide 1.66g (93%) of the product as a white solid. A portion of the material is converted into the

hydrochloride salt and recrystallized from MeOH/EtOAc. HRMS (FAB) calcd for $C_{11}H_{15}IN_4O$+H 347.0370, found 347.0357.

**Example 3(H)**: N-[(3R)-1-azabicyclo[2.2.2]oct-3-yl]-4-(2-chlorophenyl)-1H-pyrazole-1-carboxamide hydrochloride:

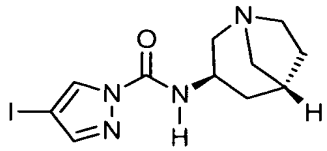



Hydrazine hydrate (0.55mL, 11.3mmol) is added to a suspension of 2-chlorophenylmalondialdehyde dissolved in 20mL EtOH. The mixture is heated under reflux for 3 min, then allowed to stir at RT overnight. The solvent is removed *in vacuo* to provide 4-(2-chlorophenyl)-1H-pyrazole as a yellow solid. MS (EI) m/z 177.0 ($M^-$).

4-Nitrophenyl chloroformate (2.3g, 11.5mmol) and 4-(2-chlorophenyl)-1H-pyrazole (2.0g, 11.0mmol) are dissolved in 30mL $CH_2Cl_2$ and cooled to 0°C. TEA (1.7mL, 12.0mmol) is added, and the reaction is allowed to warm to RT. After 30 min, additional 4-nitrophenyl chloroformate (0.25g) and TEA are added. After 1h, water is added. The mixture is extracted with $CH_2Cl_2$, dried ($MgSO_4$), filtered and concentrated to give a solid. The solid is triturated with hexanes, filtered and dried to provide 1.7g (45%) of the crude 4-nitrophenyl 4-(2-chlorophenyl)-1H-pyrazole-1-carboxylate.

A portion of 4-nitrophenyl 4-(2-chlorophenyl)-1H-pyrazole-1-carboxylate (0.34g, 1.0mmol) and (R)-(+)-3-aminoquinuclidine dihydrochloride (0.22g, 1.1mmol) are suspended in 5mL DMF. TEA (0.4mL, 3.0mmol) is added dropwise. After 18 h, 1N NaOH is added, and the solvent is removed under reduced pressure. The residue is taken up in 1N NaOH and $CHCl_3$. The aqueous layer is extracted with $CHCl_3$, dried ($MgSO_4$), filtered and concentrated. The residue is purified by chromatography (Biotage 40S, 90:9:1 $CHCl_3$/MeOH/$NH_4OH$). The hydrochloride salt is prepared and recrystallized from MeOH/EtOAc to provide 102 mg (28%) of the product. HRMS (FAB) calcd for $C_{17}H_{19}ClN_4O$+H 331.1325, found 331.1312.

**Example 4(H):** *N*-[(3*R*,5*R*)-1-azabicyclo[3.2.1]oct-3-yl]-4-iodo-1H-pyrazole-1-carboxamide:



A solution of 4-iodopyrazole (1.05 g, 5.4 mmol) in 15 mL $CH_2Cl_2$ is treated with TEA (0.90 mL, 6.5 mmol) and phenylchloroformate (0.75 ml, 6.0 mmol). The mixture is stirred for 5h and treated with H2O (1 mL). The aqueous layer is discarded and the organic dried ($MgSO_4$). The mixture is filtered, and evaporated to a yellow oil which solidifies upon evaporation from hexane. A portion of this solid (0.628 g, 2.0 mmol) is added to DMF (10 ml) containing (3*R*,5*R*)-1-azabicyclo[3.2.1]octan-3-amine dihydrochloride (0.398 g, 2.0 mmol). Diisopropylethyl amine (1.1 mL, 6.0 mmol) is added and the mixture becomes nearly homogeneous. The mixture is extracted between EtOAc and $H_2O$. The organic layer is washed with $H_2O$ (3X), brine, dried ($MgSO_4$), and the mixture is evaporated. The resulting material is taken up in hot EtOAc, filtered through celite, and allowed to stand at RT. The resulting solid is collected and dried to afford Example 4(H) (0.142 g, 20 %) as a white solid: HRMS (ESI) calcd for $C_{11}H_{15}N_4OI$ (MH+) 347.0370, found 347.0370. Anal. Calcd for $C_{11}H_{15}IN_4O$: C, 38.17; H, 4.37; N, 16.18. Found: C, 38.43; H, 4.42; N, 16.11.

Materials and Methods for identifying binding constants:

Membrane Preparation. Male Sprague-Dawley rats (300-350g) are sacrificed by decapitation and the brains (whole brain minus cerebellum) are dissected quickly, weighed and homogenized in 9 volumes/g wet weight of ice-cold 0.32 M sucrose using a rotating pestle on setting 50 (10 up and down strokes). The homogenate is centrifuged at 1,000 x g for 10 minutes at 4 °C. The supernatant is collected and centrifuged at 20,000 x g for 20 minutes at 4 °C. The resulting pellet is resuspended to a protein concentration of 1-8 mg/mL. Aliquots of 5 mL homogenate are frozen at -80 °C until needed for the assay. On the day of the assay, aliquots are thawed at room temperature and diluted with Kreb's - 20 mM Hepes buffer pH 7.0 (at room temperature) containing 4.16 mM $NaHCO_3$, 0.44 mM $KH_2PO_4$, 127 mM NaCl, 5.36 mM KCl, 1.26 mM $CaCl_2$, and 0.98 mM $MgCl_2$, so that 25 - 150 μg protein are added per test tube. Proteins are determined by the Bradford method (Bradford, M.M., *Anal. Biochem.*, 72, 248-254, 1976) using bovine serum albumin as the standard.

Binding Assay. For saturation studies, 0.4 mL homogenate are added to test tubes containing buffer and various concentrations of radioligand, and are incubated

in a final volume of 0.5 mL for 1 hour at 25 °C. Nonspecific binding was determined in tissues incubated in parallel in the presence of 0.05 ml MLA for a final concentration of 1 μM MLA, added before the radioligand. In competition studies, drugs are added in increasing concentrations to the test tubes before addition of 0.05 ml [$^3$H]-MLA for a final concentration of 3.0 to 4.0 nM [$^3$H]-MLA. The incubations are terminated by rapid vacuum filtration through Whatman GF/B glass filter paper mounted on a 48 well Brandel cell harvester. Filters are pre-soaked in 50 mM Tris HCl pH 7.0 - 0.05 % polyethylenimine. The filters are rapidly washed two times with 5 mL aliquots of cold 0.9% saline and then counted for radioactivity by liquid scintillation spectrometry.

Data Analysis. In competition binding studies, the inhibition constant (Ki) was calculated from the concentration dependent inhibition of [$^3$H]-MLA binding obtained from non-linear regression fitting program according to the Cheng-Prusoff equation (Cheng, Y.C. and Prussoff, W.H., *Biochem. Pharmacol.*, 22, p. 3099-3108, 1973). Hill coefficients were obtained using non-linear regression (GraphPad Prism sigmoidal dose-response with variable slope).

It will be apparent to those skilled in the art that the requisite carboxylic acids or carboxylic acid equivalents for when W is (H) can be obtained through synthesis via literature procedures or through the slight modification thereof. For example, methods to prepare carboxylic acids or carboxylic acid equivalents starting from pyrroles or pyrazoles are known to one of ordinary skill in the art (see *J. Org. Chem.* **1987**, *52*, 2319, *Tetrahedron Lett.* **1999**, *40*, 2733 and Greene, T. W. and Wuts, P. G. M. "Protective Groups in Organic Synthesis", 3rd Edition, p. 549, New York:Wiley, (1999)). Several pyrroles and pyrazoles of the Formula W-H are commercially available or can be obtained by methods described in *Synthesis* **1997**, 563, *J. Heterocyclic Chem.* **1993**, *30*, 865, *Heterocycles* **1982**, *19*, 1223 and *J. Org. Chem.* **1984**, *49*, 3239.

Blood-Brain Barrier Penetration

Pharmacokinetics of the compounds of formula I can be evaluated in mice to determine the ability of each compound to penetrate the blood-brain barrier. Each mouse receives a single intravenous administration at 5 mg/kg. Blood samples are collected by serial sacrifice at 5 min (IV only), 0.5, 1, 2, 4, and 8 h after dosing with two mice per collection time. Blood was placed into tubes containing heparin and

centrifuged for plasma. Brain samples were also collected at 0.5 and 1 h increments from the same mouse used for blood collection. Plasma and brain samples were analyzed for drug concentrations using a LC-MS/MMS method. Pharmacokinetics (clearance, volume of distribution, and half-life) were evaluated from the plasma concentration-time data (See Gibaldi and Perrier in Pharmacokinetics, Vol I, 2nd ed, New York: Marcel Dekker, 1982). Compounds having a large volume of distribution will have good distribution into the body tissues. Comparison of the drug concentration in brain and plasma (brain/plasma ratio) provides the direct information of brain penetration. Higher numbers refer to higher brain penetration.

Conclusion

Both combinations of two, three and all four types of drugs are described and claimed herein; however a combination of two drugs where one is a full agonist to an α7 Nicotinic Acetylcholine Receptor (nAChR) otherwise known as an α7 nAChR full agonist, examples provided above, are preferred. Where three drugs are used in combination it is preferred that one be an α7 nAChR full agonist. The combinations of drugs may be administered either at the same or different times, either in the same or different form. In one embodiment they may be given a month apart or they may be given in a co-administration where the two or three drugs are given on or about the same time in the same manner. Here the combination refers to administration such that the patients blood contains the two, three or four drugs at the same time at some point during treatment.

Also disclosed are specific administrations where the two or three drugs must be provided to the patient at about the same time, that is within a week and more preferably on the same day.